

Electronic Structure of Correlated Materials: LDA+U, DMFT, and Others

Purpose:

Understanding the limitation of standard local approximations to describe the correlated electron systems

Understanding the basic idea of LDA+U and related methods

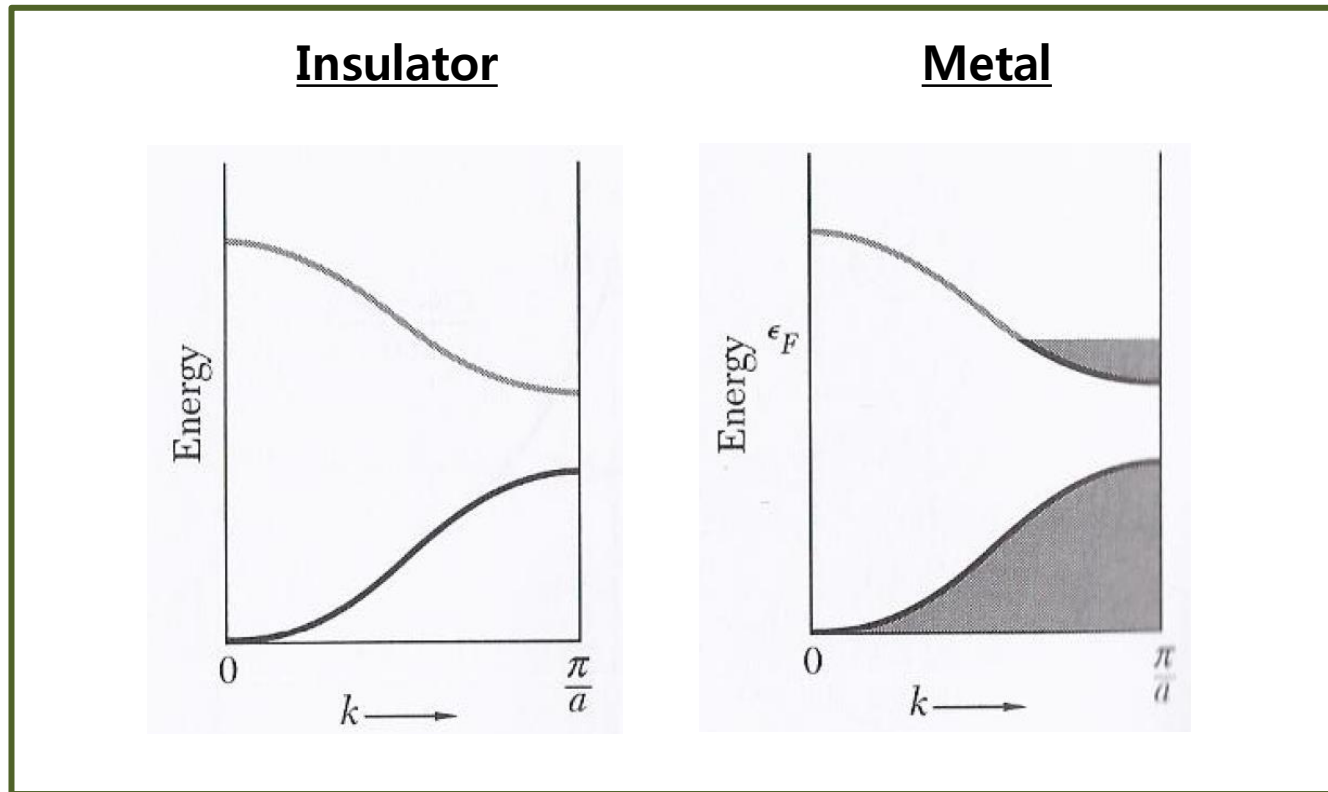
Suggested Reading:

R. G. Parr and W. Yang, “Density functional theory of atoms and molecules (OUP 1989)”

R. M. Martin, “Electronic structure: Basic theory and practical methods (CUP 2004)”

V. I. Anisimov et al., “Strong Coulomb correlations in electronic structure calculations: Beyond the local density approximation (Gordon & Breach 2000)”

Very Basic of Band Theory

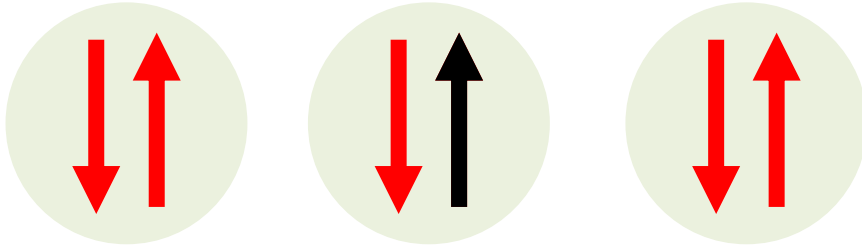


Kittel, Introduction to Solid State Physics

✓ A material with partially-filled band(s) should be a metal

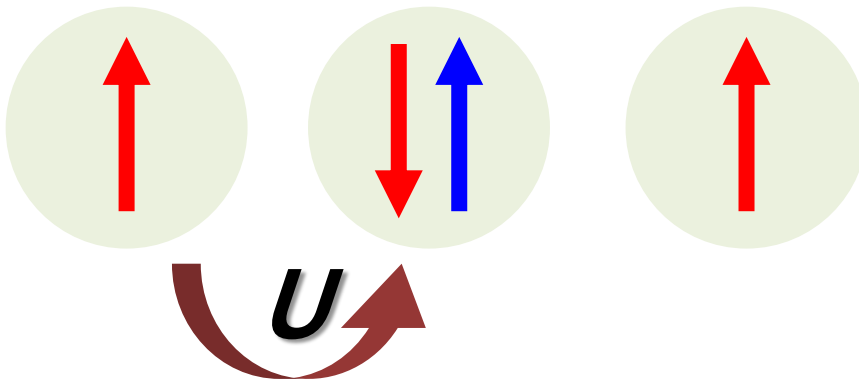
Insulator: Pauli and Mott

Pauli exclusion: band insulator



W. E. Pauli

Coulomb repulsion: Mott (-Hubbard) insulator



N. F. Mott

Proc. Phys. Soc. London (1949)

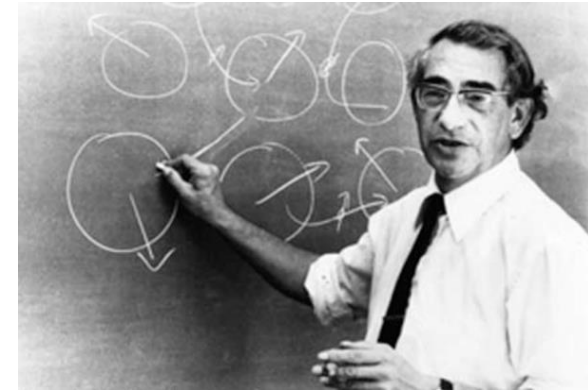
Localized Orbital and Hubbard Model

Hubbard Model (1964)

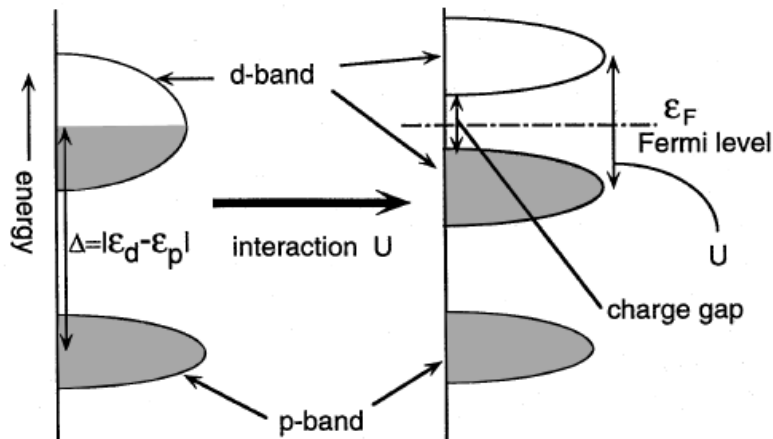
$$H = - \sum_{i,j,\sigma} t_{ij} (c_{i\sigma}^\dagger c_{j\sigma} + \text{H.c.}) + U \sum_i n_{i\uparrow} n_{i\downarrow}$$

'Hopping' term
between the sites

On-site Coulomb repulsion
in the correlated orbitals



<http://theor.jinr.ru/~kuzemsky/jhbio.html>



Additional electron occupation requires
the energy cost :

$$U = E(d^{n+1}) + E(d^{n-1}) - 2E(d^n)$$

Imada, Fujimori, Tokura, Rev. Mod. Phys. (1998)

Actually Happening Quite Often

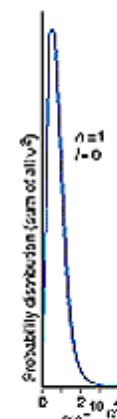
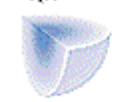
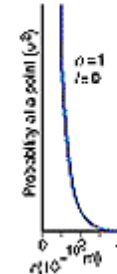
hydrogen 1 H 1.0079																	helium 2 He 4.0026																																																												
lithium 3 Li 6.941	beryllium 4 Be 9.0122																	neon 10 Ne 20.180																																																											
sodium 11 Na 22.990	magnesium 12 Mg 24.305																	argon 18 Ar 39.948																																																											
potassium 19 K 39.098	calcium 20 Ca 40.078																	krypton 36 Kr 83.80																																																											
rubidium 37 Rb 85.468	strontium 38 Sr 87.62																	iodine 53 I 126.904																																																											
caesium 55 Cs 132.91	barium 56 Ba 137.33																	astatine 85 At [210]																																																											
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<table><tr><td>scandium 21 Sc 44.956</td><td>titanium 22 Ti 47.867</td><td>vanadium 23 V 50.942</td><td>chromium 24 Cr 51.996</td><td>manganese 25 Mn 54.938</td><td>iron 26 Fe 55.845</td><td>cobalt 27 Co 58.933</td><td>nickel 28 Ni 58.693</td><td>copper 29 Cu 63.546</td><td>zinc 30 Zn 65.39</td></tr><tr><td>yttrium 39 Y 88.906</td><td>zirconium 40 Zr 91.224</td><td>niobium 41 Nb 92.906</td><td>molybdenum 42 Mo 95.94</td><td>technetium 43 Tc [98]</td><td>ruthenium 44 Ru 101.07</td><td>rhodium 45 Rh 102.91</td><td>palladium 46 Pd 106.42</td><td>silver 47 Ag 107.87</td><td>cadmium 48 Cd 112.41</td></tr><tr><td>lanthanum 57 La 138.905</td><td>cerium 58 Ce 140.12</td><td>praseodymium 59 Pr 140.908</td><td>neodymium 60 Nd 144.24</td><td>promethium 61 Pm [145]</td><td>samarium 62 Sm 150.36</td><td>europium 63 Eu 151.964</td><td>gadolinium 64 Gd 157.25</td><td>terbium 65 Tb 158.925</td><td>dysprosium 66 Dy 162.50</td></tr><tr><td>holmium 67 Ho 164.930</td><td>erbium 68 Er 167.259</td><td>thulium 69 Tm 168.930</td><td>ytterbium 70 Yb 173.054</td><td>lutetium 71 Lu 174.967</td><td>hafnium 72 Hf 178.49</td><td>tantalum 73 Ta 180.948</td><td>tungsten 74 W 183.84</td><td>rhenium 75 Re 186.21</td><td>osmium 76 Os 190.23</td></tr><tr><td>iridium 77 Ir 192.222</td><td>platinum 78 Pt 195.084</td><td>gold 79 Au 196.967</td><td>mercury 80 Hg 200.59</td><td>thallium 81 Tl 204.38</td><td>lead 82 Pb 207.2</td><td>bismuth 83 Bi 208.98</td><td>polonium 84 Po [209]</td><td>astatine 85 At [210]</td><td>radon 86 Rn [222]</td></tr><tr><td>bohrium 107 Bh [264]</td><td>hassium 108 Hs [265]</td><td>meitnerium 109 Mt [266]</td><td>darmstadtium 110 Ds [271]</td><td>roentgenium 111 Rg [272]</td><td>copernicium 112 Cn [277]</td><td>nihonium 113 Nh [286]</td><td>flerovium 114 Fl [289]</td><td>tennessium 115 Ts [294]</td><td>oganesson 116 Og [294]</td></tr></table>																		scandium 21 Sc 44.956	titanium 22 Ti 47.867	vanadium 23 V 50.942	chromium 24 Cr 51.996	manganese 25 Mn 54.938	iron 26 Fe 55.845	cobalt 27 Co 58.933	nickel 28 Ni 58.693	copper 29 Cu 63.546	zinc 30 Zn 65.39	yttrium 39 Y 88.906	zirconium 40 Zr 91.224	niobium 41 Nb 92.906	molybdenum 42 Mo 95.94	technetium 43 Tc [98]	ruthenium 44 Ru 101.07	rhodium 45 Rh 102.91	palladium 46 Pd 106.42	silver 47 Ag 107.87	cadmium 48 Cd 112.41	lanthanum 57 La 138.905	cerium 58 Ce 140.12	praseodymium 59 Pr 140.908	neodymium 60 Nd 144.24	promethium 61 Pm [145]	samarium 62 Sm 150.36	europium 63 Eu 151.964	gadolinium 64 Gd 157.25	terbium 65 Tb 158.925	dysprosium 66 Dy 162.50	holmium 67 Ho 164.930	erbium 68 Er 167.259	thulium 69 Tm 168.930	ytterbium 70 Yb 173.054	lutetium 71 Lu 174.967	hafnium 72 Hf 178.49	tantalum 73 Ta 180.948	tungsten 74 W 183.84	rhenium 75 Re 186.21	osmium 76 Os 190.23	iridium 77 Ir 192.222	platinum 78 Pt 195.084	gold 79 Au 196.967	mercury 80 Hg 200.59	thallium 81 Tl 204.38	lead 82 Pb 207.2	bismuth 83 Bi 208.98	polonium 84 Po [209]	astatine 85 At [210]	radon 86 Rn [222]	bohrium 107 Bh [264]	hassium 108 Hs [265]	meitnerium 109 Mt [266]	darmstadtium 110 Ds [271]	roentgenium 111 Rg [272]	copernicium 112 Cn [277]	nihonium 113 Nh [286]	flerovium 114 Fl [289]	tennessium 115 Ts [294]	oganesson 116 Og [294]
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actinium 89 Ac [227]	thorium 90 Th 232.04	protactinium 91 Pa 231.04	uranium 92 U 238.03	neptunium 93 Np [237]	plutonium 94 Pu [244]	americium 95 Am [243]	curium 96 Cm [247]	berkelium 97 Bk [247]	californium 98 Cf [251]	einsteinium 99 Es [252]	fermium 100 Fm [257]	mendelevium 101 Md [258]	nobelium 102 No [259]

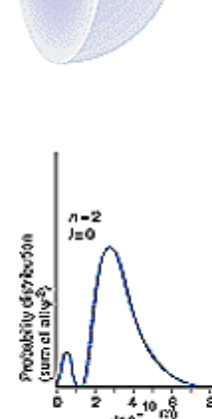
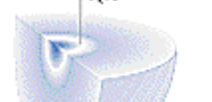
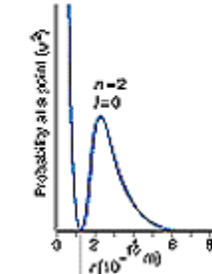
Localized valence wavefunctions

Partially filled 3d, 4f, 5f orbitals

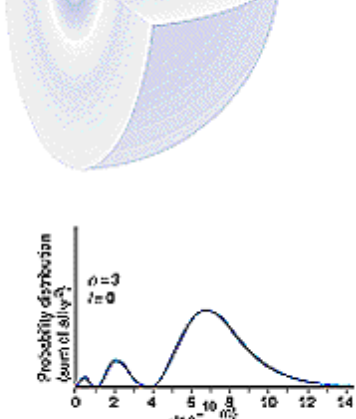
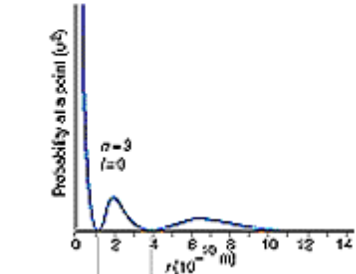
→ magnetism and others



A 1s orbital



B 2s orbital

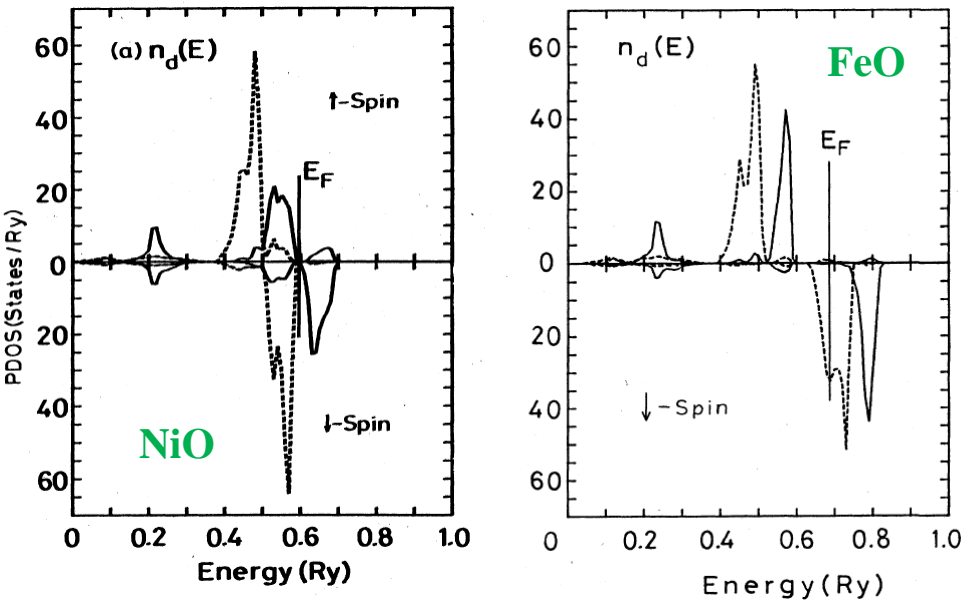


C 3d orbital

Applying LDA to Mott Insulators

Oguchi et al., PRB (1983)

Terakura et al., PRB (1984)



- ✓ Too small or zero band gap
- ✓ Magnetic moment underestimated
- ✓ Too large exchange coupling (T_c)

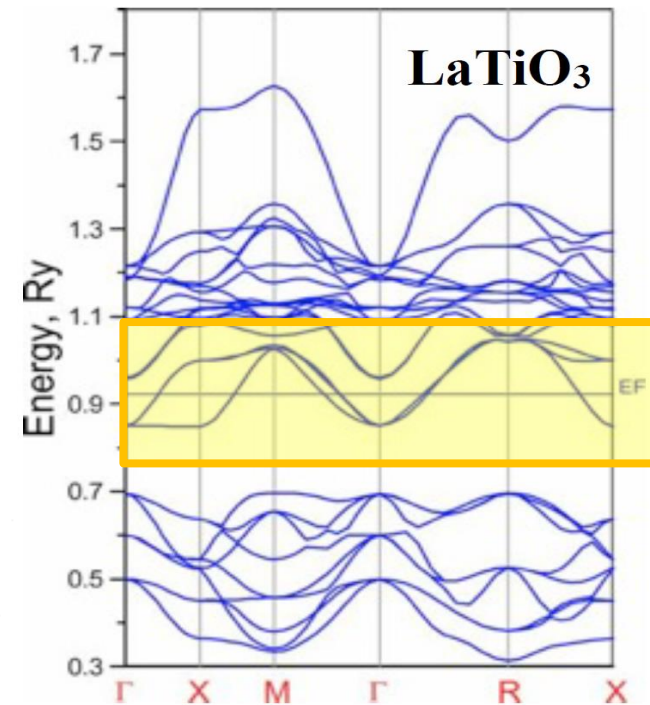


TABLE III. Experimental and theoretical exchange parameters (in K) of the first and the second neighbors.

MnO		MnS		NiO		Reference
J_1	J_2	J_1	J_2	J_1	J_2	
-14.4	-7.0	-8.0	-9.0	-50	-85	22
-9.0	-10.4					25
				16	-222	26
-10.0	-11.0					23
-28	-28					24
-30.3	-29.8	-10	-27	61	-1230	present paper

Combining LDA with Hubbard Model

PHYSICAL REVIEW B

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Band theory and Mott insulators: Hubbard U instead of Stoner I

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(Received 4 September 1990; revised manuscript received 5 March 1991)

We propose a form for the exchange-correlation potential in local-density band theory, appropriate for Mott insulators. The idea is to use the “constrained-local-density-approximation” Hubbard parameter U as the quantity relating the single-particle potentials to the magnetic- (and orbital-) order parameters. Our energy functional is that of the local-density approximation plus the mean-field approximation to the remaining part of the U term. We argue that such a method should make sense, if one accepts the Hubbard model and the success of constrained-local-density-approximation parameter calculations. Using this *ab initio* scheme, we find that all late-3d-transition-metal monoxides, as well as the parent compounds of the high- T_c compounds, are large-gap magnetic insulators of the charge-transfer type. Further, the method predicts that LiNiO_2 is a low-spin ferromagnet and NiS a local-moment p -type metal. The present version of the scheme fails for the early-3d-transition-metal monoxides and for the late 3d transition metals.

Basic idea: Introduce Hubbard-like term into the energy functional (and subtract the equivalent LDA term to avoid the double counting)

$$E_{LDA+U} = E_{LDA} + E_U - E_{dc} \quad \text{where} \quad E_U = \frac{1}{2} \sum_{ilm\sigma} U n_{ilm}^{\sigma} n_{ilm'}^{\sigma'}$$

LDA+U Functional

The original functional form (Anisimov et al. 1991)

$$E = E^{\text{LDA}} + \frac{1}{2} \sum_{m, m', \sigma} U (n_{im\sigma} - n^0) (n_{im'-\sigma} - n^0) \\ + \frac{1}{2} \sum_{\substack{m, m', \sigma \\ (m \neq m')}} (U - J) (n_{im\sigma} - n^0) (n_{im'\sigma} - n^0) .$$

i : site index (orbitals)

n⁰ : average d-orbital
occupation (no double
counting correction)

J: Hund coupling constant

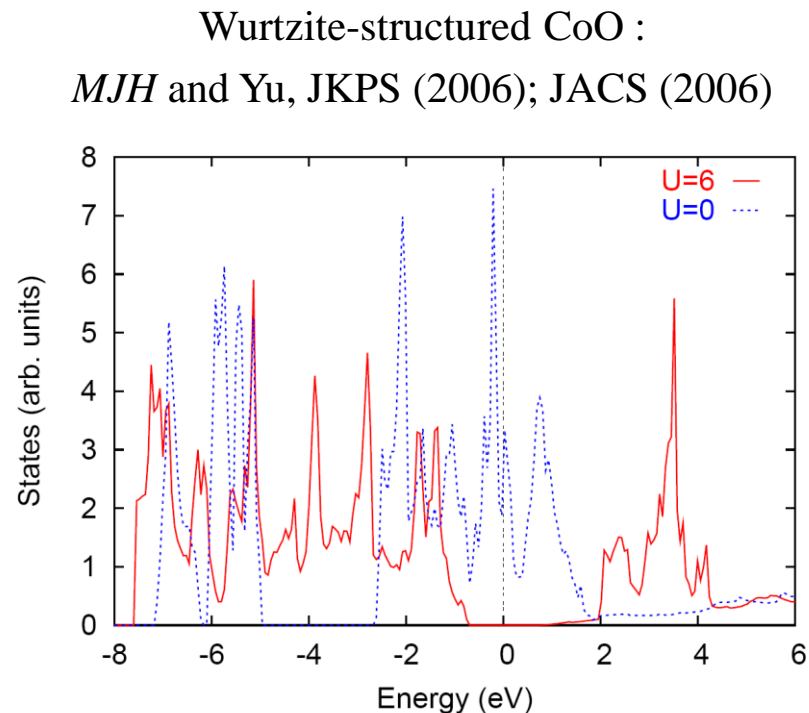
Orbital-dependent potential

$$V_{m\sigma} = U \sum_{m'} (n_{m'-\sigma} - n^0) \\ + (U - J) \sum_{m'(\neq m)} (n_{m'\sigma} - n^0) + V^{\text{LDA}}$$

LDA+U Result

TABLE I. Calculated band gaps of MnO, FeO, CoO, and NiO as a function of \bar{U} values. Here we use the dual representation for the description of the on-site density matrix.

\bar{U} (eV)	MnO	FeO	CoO	NiO
0	0.04	0.00	0.00	0.36
2	1.68	0.76	1.20	1.56
4	2.12	1.96	2.20	2.53
6	4.21	2.77	3.01	3.89
Calc. (LSDA)	0.8 ^a	0.0 ^a	0.0 ^a	0.2 ^a
Calc. (LDA+U)	3.5 ^b	3.2 ^c	3.2 ^d	3.1 ^e
Expt.	3.6-3.8 ^f	2.4 ^g	2.4 ^h	4.0 ⁱ , 4.3 ^j



\bar{U} (eV)	Mulliken			Voronoi			Other group results
	Full	Dual	On-site	Full	Dual	On-site	
0	1.30	1.30	1.30	1.28	1.28	1.28	1.0 ($\bar{U}=0.0$) ^a
2	1.48	1.54	1.59	1.46	1.51	1.55	
4	1.59	1.66	1.71	1.55	1.61	1.66	
6	1.66	1.74	1.79	1.62	1.69	1.72	1.59 ($\bar{U}=6.9$) ^b
Expt.							1.77 ^c , 1.64 ^d , 1.90 ^e

MJH, Ozaki and Yu
 PRB (2006)

Further Issues

✓ Rotational invariance and several different functional forms

$$E^U[\{n\}] = \frac{1}{2} \sum_{\{m\}, \sigma} \{ \langle m, m'' | V_{ee} | m', m''' \rangle n_{mm'}^\sigma n_{m''m'''}^{-\sigma} \\ + (\langle m, m'' | V_{ee} | m', m''' \rangle \\ - \langle m, m'' | V_{ee} | m''', m' \rangle) n_{mm'}^\sigma n_{m''m'''}^\sigma \},$$

(So-called) fully localized limit:
Liechtenstein et al. PRB (1995)

$$E_{\text{dc}}[\{n^\sigma\}] = \frac{1}{2} U n(n-1) - \frac{1}{2} J [n^\uparrow (n^\uparrow - 1) + n^\downarrow (n^\downarrow - 1)].$$

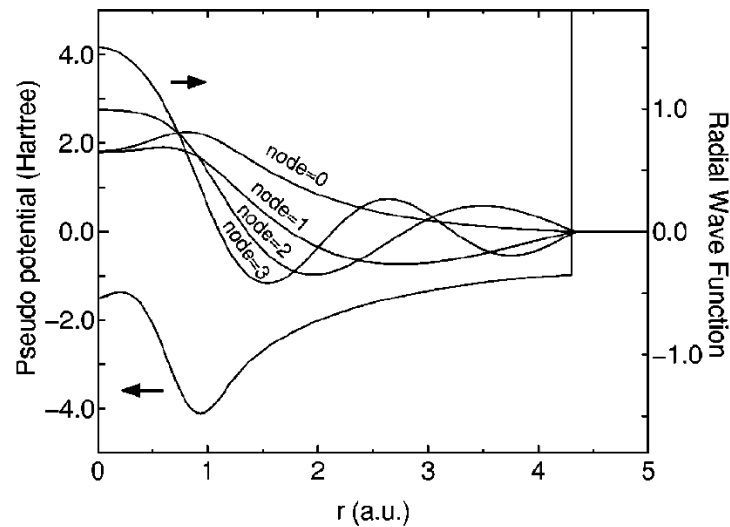
(So-called) around the mean field limit:
Czyzyk et al. PRB (1994)

$$E^{\text{LSDA}+\text{AMF}} = E^{\text{LSDA}} + \frac{1}{2} \sum_{m, m', \sigma} U_{mm'} (n_{m\sigma} - n_\sigma^0) \\ \times (n_{m'-\sigma} - n_{-\sigma}^0) \\ + \frac{1}{2} \sum_{m, m', m \neq m', \sigma} (U_{mm'} - J_{mm'}) (n_{m\sigma} - n_\sigma^0) \\ \times (n_{m'\sigma} - n_\sigma^0). \quad (3)$$

LDA+U based on LCPAO (1)

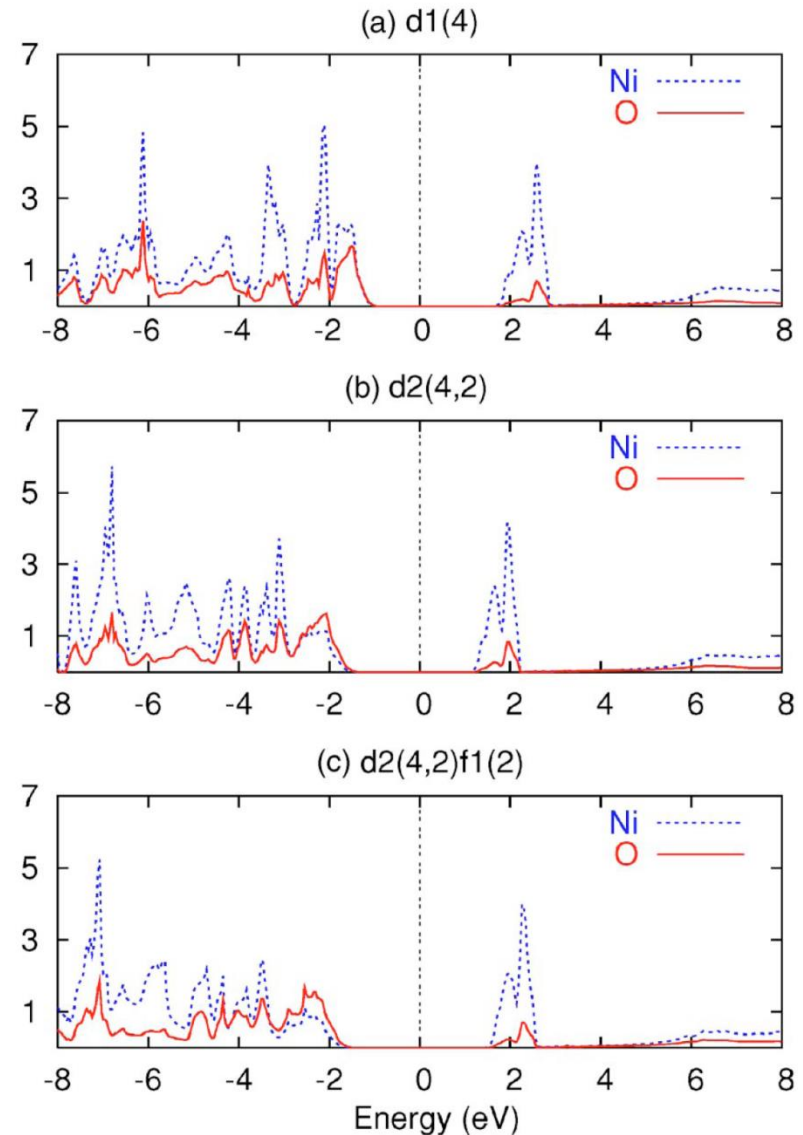
- ✓ Numerically generated (pseudo-) atomic orbital basis set:

Non-orthogonal multiple d-/f-orbitals with arbitrarily-chosen cutoff radii



T. Ozaki, Phys. Rev. B (2003)

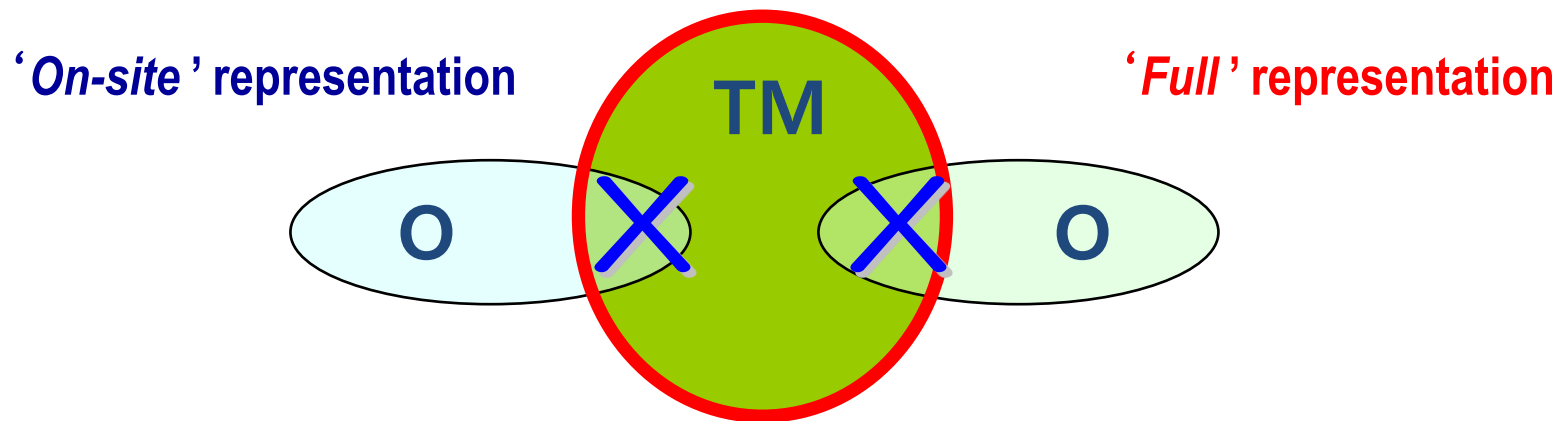
MJH, Ozaki, Yu, Phys. Rev. B (2006)



LDA+U based on LCPAO (2)

- ✓ **Non-orthogonality and no guarantee for the sum rule**

See, for example, Pickett et al., (1998)



Proposed 'dual' representation:

$$\hat{\rho}_{smm'}^{\sigma} = \frac{1}{2}(|s\tilde{m}\sigma\rangle\langle sm'\sigma| + |sm\sigma\rangle\langle s\tilde{m}'\sigma|)$$

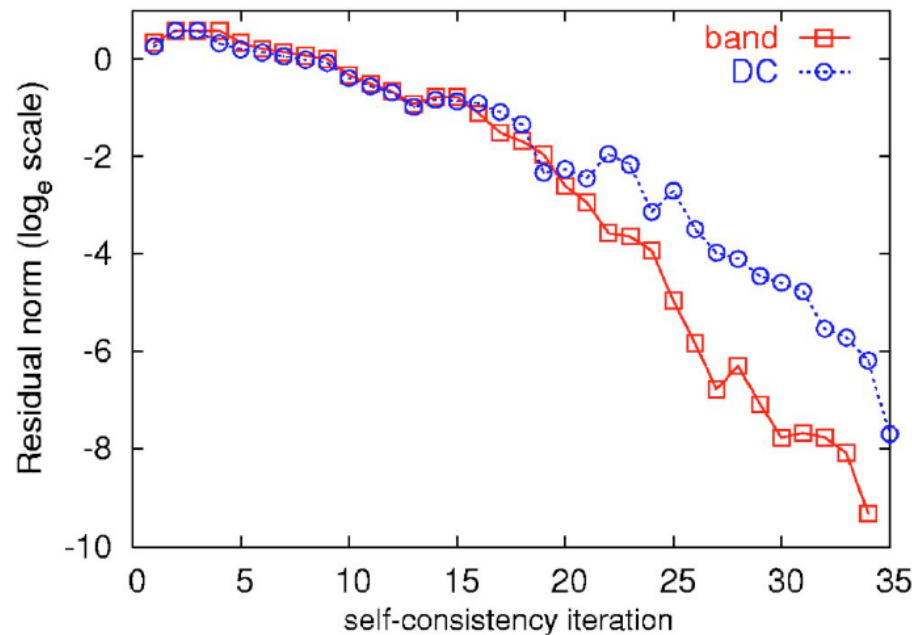
Sum rule satisfied:

$$\sum_{\sigma} Tr(n^{\sigma}) = \sum_{\sigma} \frac{1}{2}[Tr(S\rho^{\sigma}) + Tr(\rho^{\sigma}S)] = N_{ele}$$

where n^{σ} is the density matrix.

LDA+U based on LCPAO (3)

- ✓ **LMTO:** Anisimov et al. Phys. Rev. B (1991)
- ✓ **FLAPW:** Shick et al. Phys. Rev. B (1999)
- ✓ **PAW:** Bengone et al. Phys. Rev. B (2000)
- ✓ **PP-PW:** Sawada et al., (1997); Cococcioni et al., (2005)
- ✓ **LCPAO and O(N) LDA+U:** Large-scale correlated electron systems



MJH, Ozaki, Yu,
Phys. Rev. B (2006)

Limitations

- ✓ **How to determine the U and J values?**

No fully satisfactory way to determine the key parameters

- ✓ **How to define the double-counting energy functional?**

Rotationally invariant versions, fully-local form, around the mean-field form, etc

See, Anisimov et al. (1991); Czyzyk and Sawatzky (1994); Dudarev et al. (1998)

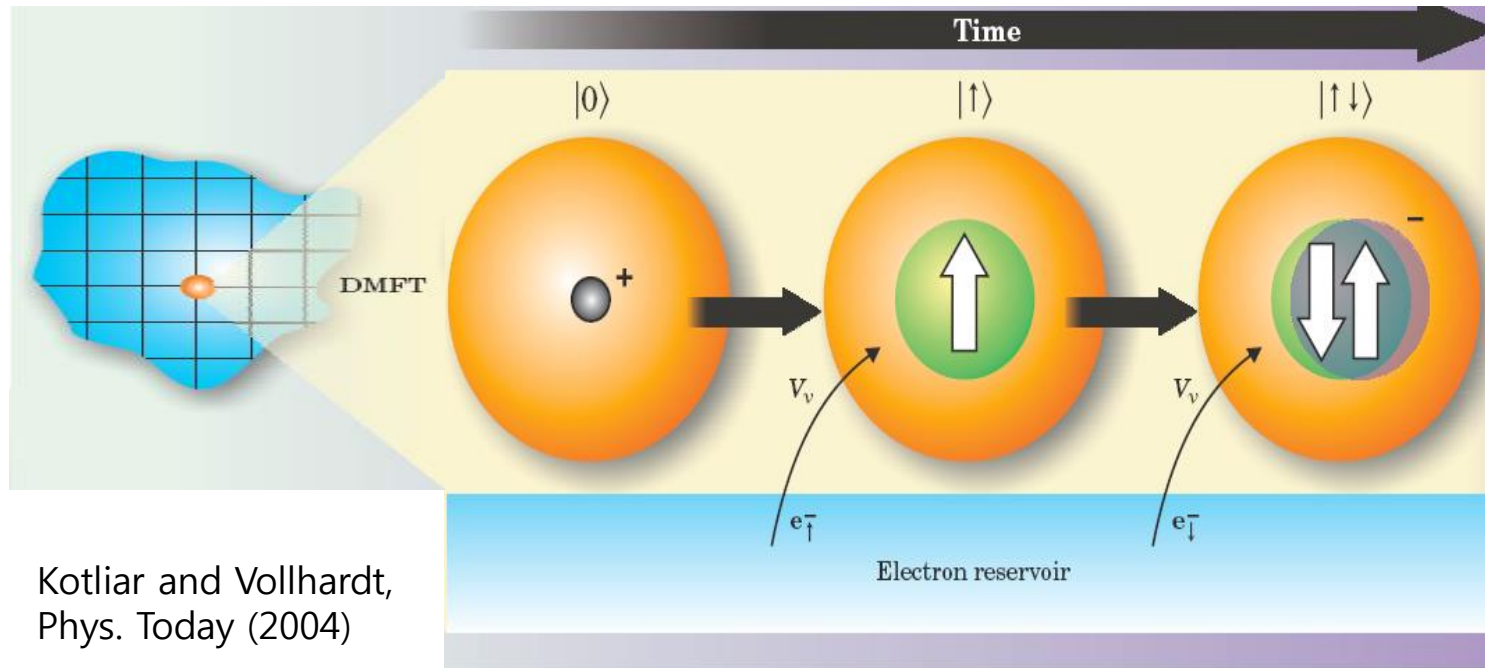
- ✓ **It is a static Hartree-Fock method**

The correlation effect beyond this static limit cannot be captured

→ **Dynamical mean-field theory**

Dynamical Correlation and DMFT

✓ Dynamical mean-field theory



Mapping 'Hubbard Hamiltonian' into 'Anderson Impurity Hamiltonian' plus 'self-consistent equation'

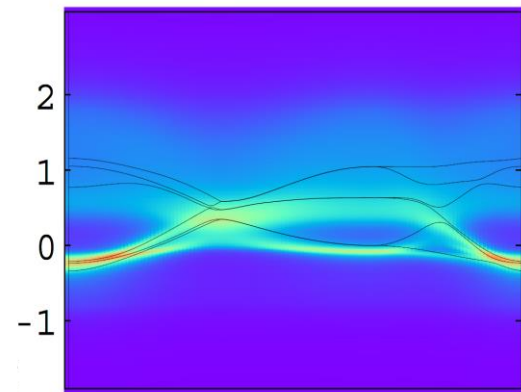
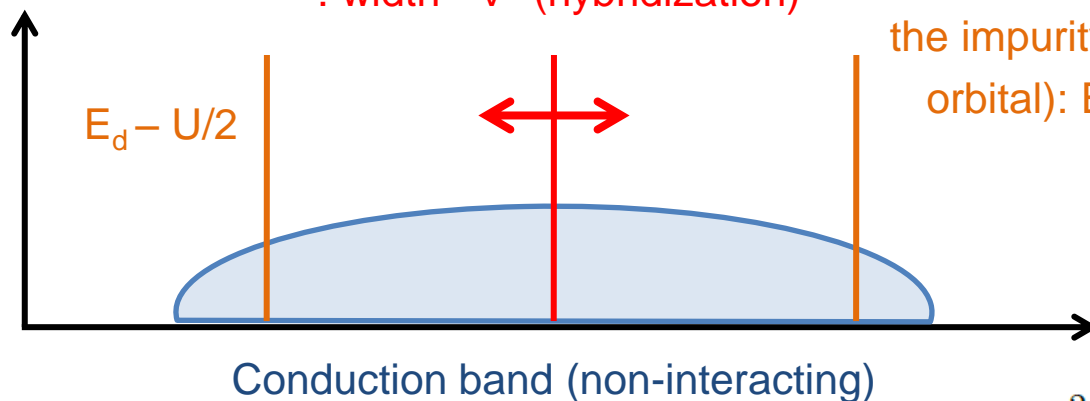
Georges and Kotliar Phys. Rev. B (1992)

Georges et al., Rev. Mod. Phys.(1996); Kotliar et al., Rev. Mod. Phys.(2006);

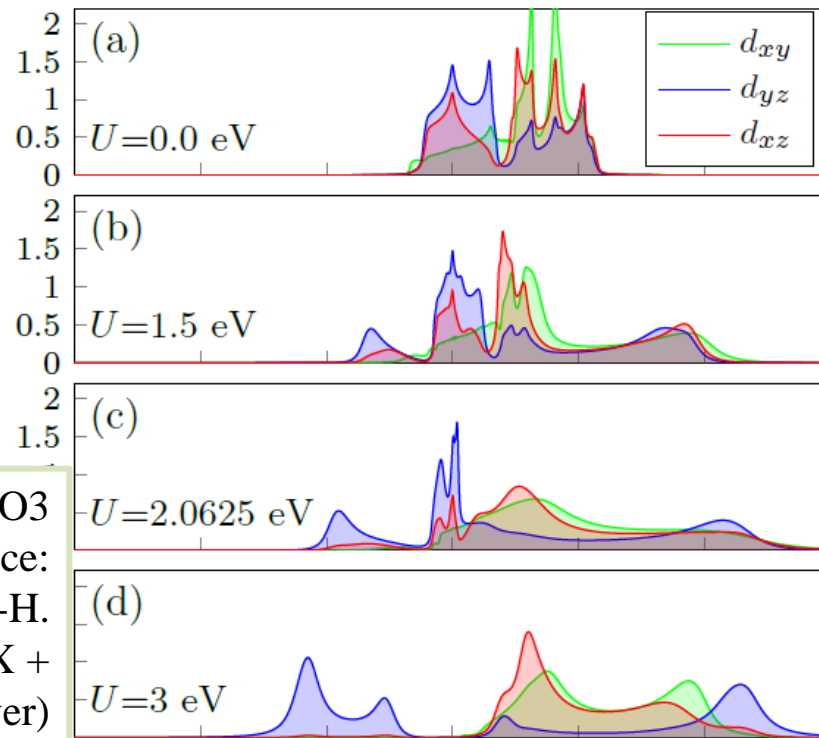
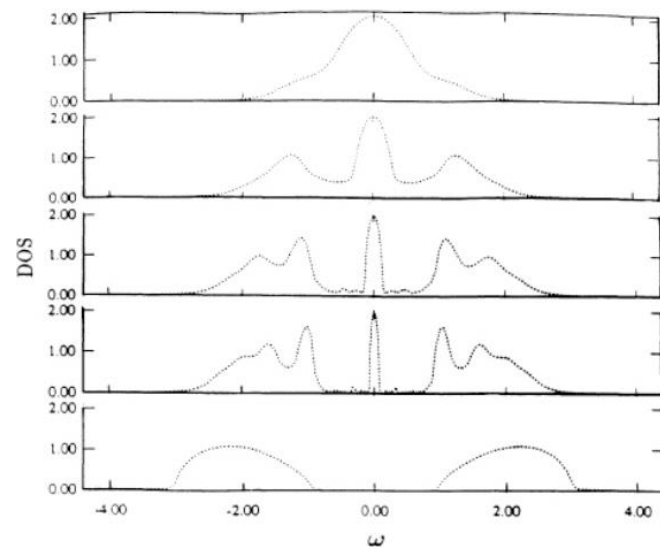
DMFT Result

Impurity level (atomic-like)
: width $\sim V^2$ (hybridization)

On-site correlation at
the impurity site (or
orbital): $E_d + U/2$



Zhang et al., PRL (1993)



LaTiO₃/LaAlO₃
superlattice:
calculated by J.-H.
Sim (OpenMX +
ALPS-DMFT solver)

Comparison

LDA

No on-site correlation
(homogeneous electron gas)

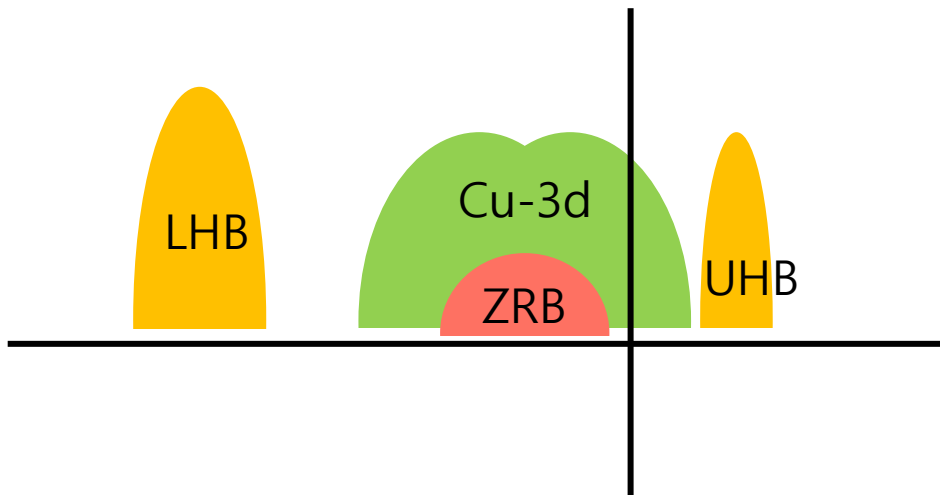
LDA+U

Hubbard-U correlation
Static approximation

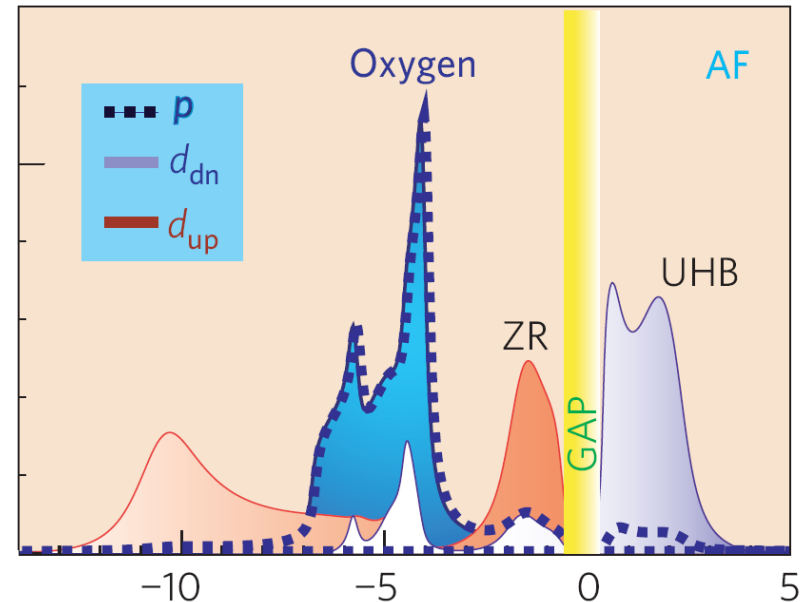
LDA+DMFT

Hubbard-U correlation
Dynamic correlation

Application to high- T_c cuprate



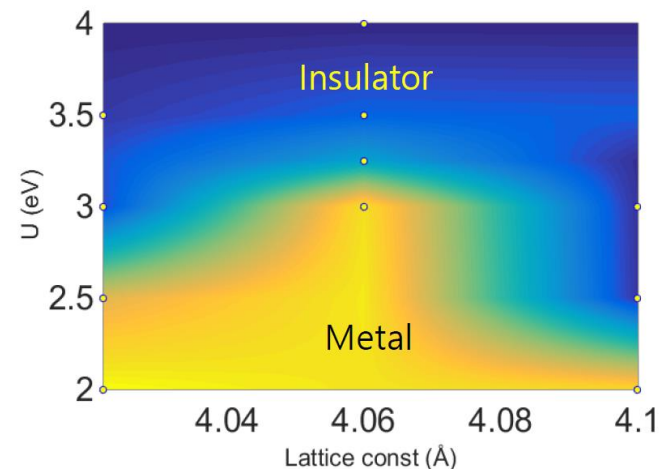
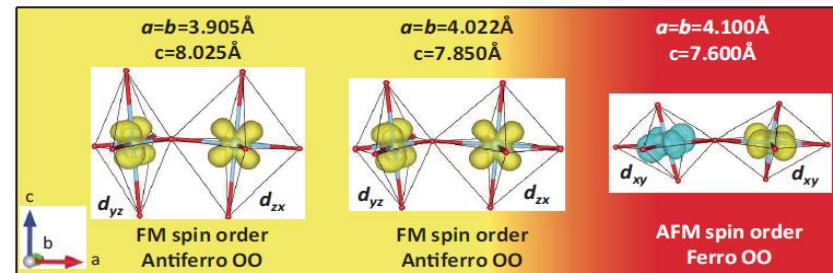
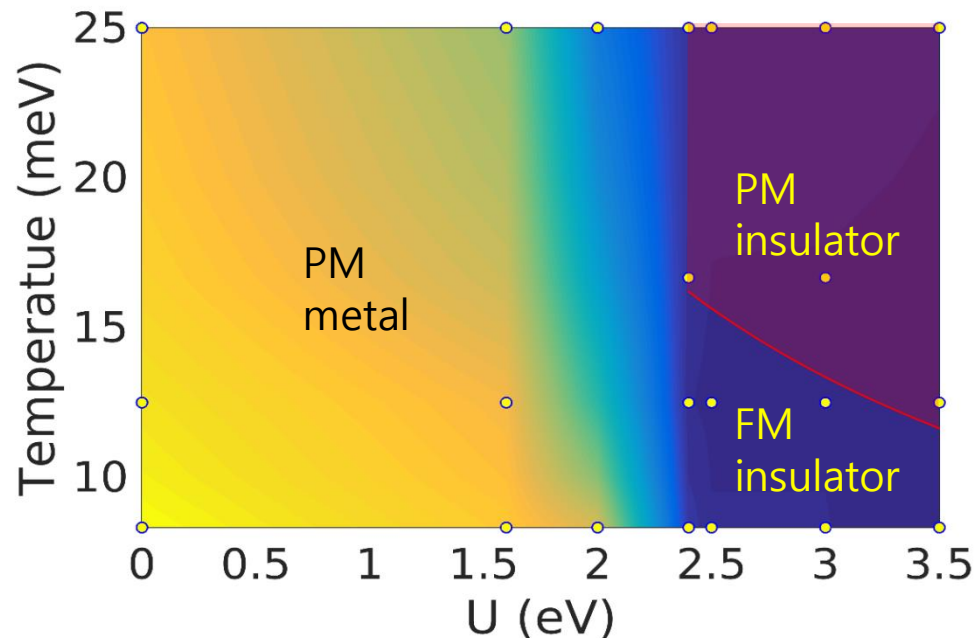
Weber et al., Nature Phys. (2010)



LDA+U and DMFT

- ✓ LDA+U is static (Hartree) approximation of DMFT →
- ✓ Temperature dependency
- ✓ Electronic property near the phase boundary
- ✓ Paramagnetic insulating and correlated metallic phase

LaTiO₃/LaAlO₃ superlattice: calculated by J.-H. Sim (OpenMX + ALPS-DMFT solver)



Other Methods

✓ Hybrid functionals, self-interaction correction, etc

- Inclusion of atomic nature can always be helpful
- ‘Controllability’ versus ‘parameter-free’-ness
- Hidden parameters (or factors)
- Computation cost (→ relaxation etc)

✓ (Self-consistent) GW

- Parameter-free way to include the well-defined self energy
- No way to calculate total energy, force,...etc
- Fermi liquid limit

Tran and Blaha,
PRL (2009)

