

Tutorial of OpenMX

- About OpenMX
- Implementation of OpenMX
- Tutorial of OpenMX

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OpenMX **Open** source package for **M**aterial **eX**plorer

- Software package for density functional calculations of molecules and bulks
- Norm-conserving pseudopotentials (PPs)
- Variationally optimized numerical atomic basis functions

Basic functionalities

- SCF calc. by LDA, GGA, DFT+U
- Total energy and forces on atoms
- Band dispersion and density of states
- Geometry optimization by BFGS, RF, EF
- Charge analysis by Mulliken, Voronoi, ESP
- Molecular dynamics with NEV and NVT ensembles
- Charge doping
- Fermi surface
- Analysis of charge, spin, potentials by cube files
- Database of optimized PPs and basis functions

Extensions

- $O(N)$ and low-order scaling diagonalization
- Non-collinear DFT for non-collinear magnetism
- Spin-orbit coupling included self-consistently
- Electronic transport by non-equilibrium Green function
- Electronic polarization by the Berry phase formalism
- Maximally localized Wannier functions
- Effective screening medium method for biased system
- Reaction path search by the NEB method
- Band unfolding method
- STM image by the Tersoff-Hamann method
- etc.

History of OpenMX

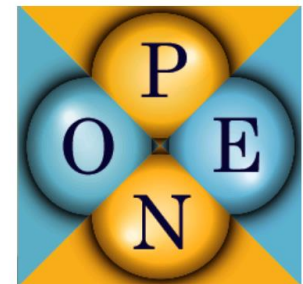
- 2000 Start of development
- 2003 Public release (GNU-GPL)
- 2003 Collaboration:
AIST, NIMS, SNU
KAIST, JAIST,
Kanazawa Univ.
CAS, UAM
NISSAN, Fujitsu Labs.
etc.
- 2019 19 public releases
Latest version: 3.9

Welcome to OpenMX
open source package for Material eXplorer

Google C

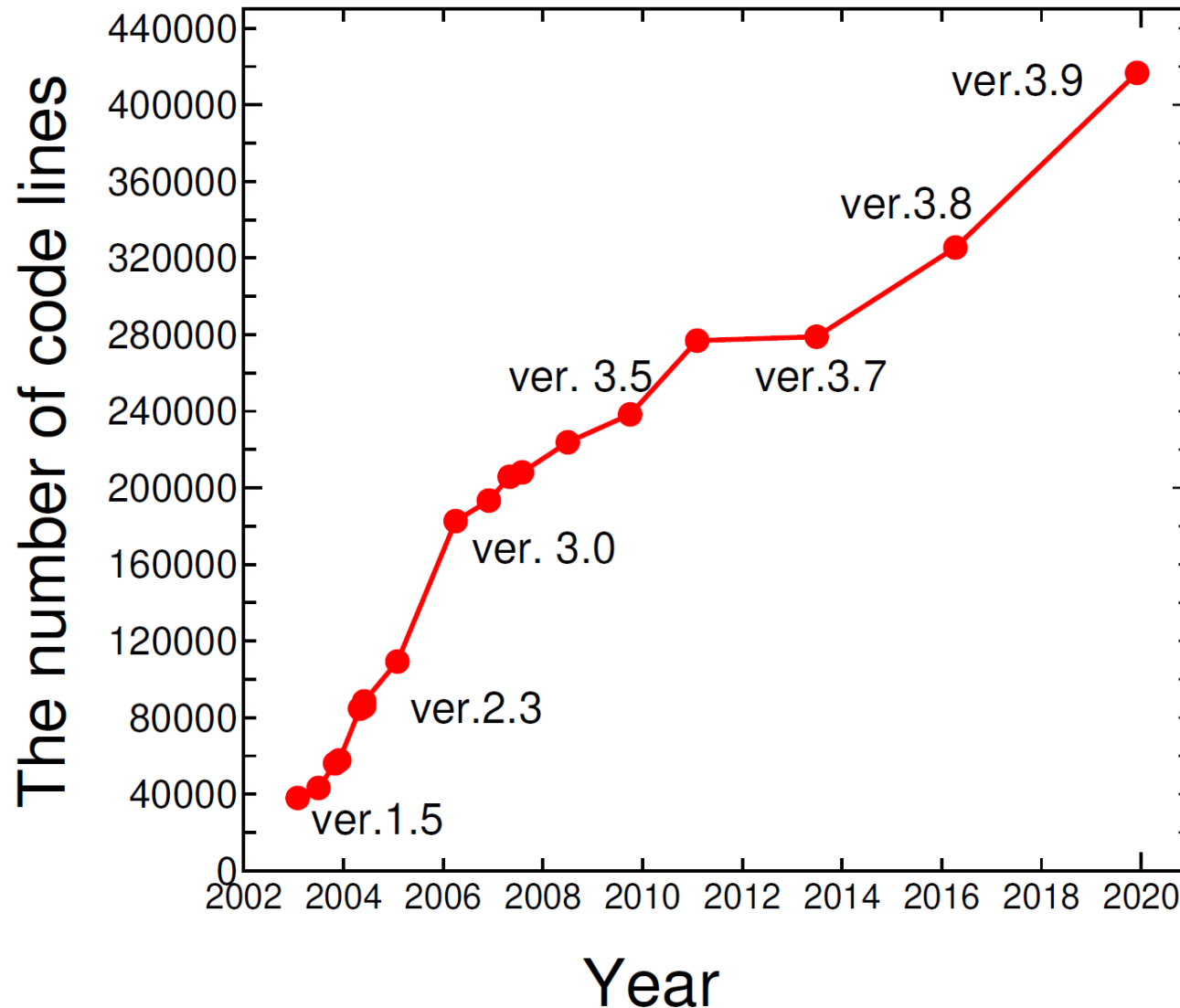
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<http://www.openmx-square.org>

Development of OpenMX code



Contributors to OpenMX development

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Materials studied by OpenMX

First characterization of silicene on ZrB_2 in collaboration with experimental groups

A. Fleurence et al., Phys. Rev. Lett. 108, 245501 (2012).

First identification of $\text{Jeff}=1/2$ Mott state of Ir oxides

B.J. Kim et al., Phys. Rev. Lett. 101, 076402 (2008).

Theoretical proposal of topological insulators

C.-H. Kim et al., Phys. Rev. Lett. 108, 106401 (2012).

H. Weng et al., Phys. Rev. X 4, 011002 (2014).

First-principles molecular dynamics simulations for Li ion battery

T. Ohwaki et al., J. Chem. Phys. 136, 134101 (2012).

T. Ohwaki et al., J. Chem. Phys. 140, 244105 (2014).

Magnetic anisotropy energy of magnets

Z. Torbatian et al., Appl. Phys. Lett. 104, 242403 (2014).

I. Kitagawa et al., Phys. Rev. B 81, 214408 (2010).

Electronic transport of graphene nanoribbon on surface oxidized Si

H. Jippo et al., Appl. Phys. Express 7, 025101 (2014).

M. Ohfuchi et al., Appl. Phys. Express 4, 095101 (2011).

Interface structures of carbide precipitate in bcc-Fe

H. Sawada et al., Modelling Simul. Mater. Sci. Eng. 21, 045012 (2013).

Universality of medium range ordered structure in amorphous metal oxides

K. Nishio et al., Phys. Rev. Lett. 340, 155502 (2013).

Materials treated so far

Silicene, graphene

Carbon nanotubes

Transition metal oxides

Topological insulators

Intermetallic compounds

Molecular magnets

Rare earth magnets

Lithium ion related materials

Structural materials

etc.

About 1200 published papers

Information of OpenMX

<http://www.openmx-square.org/index.html>

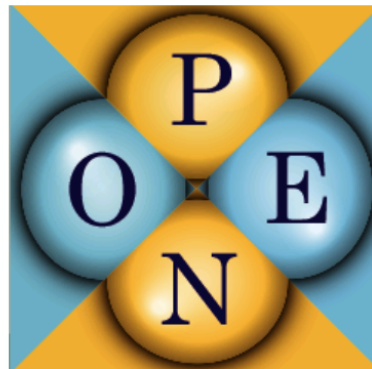
Welcome to OpenMX
Open source package for Material explorer

- **What's new**

- Patch3.9.2 to OpenMX Ver 3.9 (Feb. 11, 2020)

- Patch3.9.1 to OpenMX Ver 3.9 (Jan. 02, 2020)

- **What is OpenMX?**
- **Download**
- **Manual of Ver. 3.9**
- **Manual of Ver. 3.8**
- **Technical Notes**
- **Video Lectures**
- **Publications**
- **OpenMX Forum**
- **OpenMX Viewer**
- **Workshop**
- **Database of Results**
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 - **Ver. 2019 for core excitations**
- **ADPACK**
- **Miscellaneous informations**
- **Contributors**
- **Acknowledgment**
- **Opening positions**
- **Links**



- Source codes
- Manual (Eng./日本語)
- OpenMX Forum
- Technical notes
- Publications
- Database of basis functions and pseudopotentials

Getting started with OpenMX...

- **Install by yourself**
- **Use of preinstalled version in Computer centers**
- **Use of MateriApps Live!**

Let's start a simple calculation

Perform a SCF calculation of methane molecule.

```
ozaki@mx17:~/openmx3.9/work
```

```
[ozaki@mx17 work]$ mpirun -np 4 ./openmx Methane.dat | tee met.std
```

After finishing the calculation, several files are output as follows:

```
ozaki@mx17:~/openmx3.9/work
```

```
[ozaki@mx17 work]$ ls met.*  
met.cif      met.ene  met.md2  met.std      met.v0.cube  met.xyz  
met.dden.cube met.md   met.out  met.tden.cube met.vhart.cube  
[ozaki@mx17 work]$
```

Input file of OpenMX

```
System.CurrentDirectory    ./    # default=./
System.Name               met
level.of.stdout           1    # default=1 (1-3)
level.of.fileout          1    # default=1 (0-2)
```

- (1) Specify a value after keyword
- (2) Arbitrary order of keywords
- (3) # by comment

```
#
# Definition of Atomic Species
#
```

```
Species.Number           2
<Definition.of.Atomic.Species>
  H   H5.0-s1            H_PBE13
  C   C5.0-s1p1          C_PBE13
Definition.of.Atomic.Species>
```

Several keywords:

System name: files will be saved with the name

of species in the system

Definition of species

```
#
# Atoms
#
```

of atoms

```
Atoms.Number             5
Atoms.SpeciesAndCoordinates.Unit  Ang # Ang|AU
```

Unit of atomic coordinate

```
<Atoms.SpeciesAndCoordinates>
  1  C      0.000000    0.000000    0.000000    2.0  2.0
  2  H     -0.889981   -0.629312    0.000000    0.5  0.5
  3  H      0.000000    0.629312   -0.889981    0.5  0.5
  4  H      0.000000    0.629312    0.889981    0.5  0.5
  5  H      0.889981   -0.629312    0.000000    0.5  0.5
```

Atomic coordinates

Initial occupation for up and down states of each atom

```
Atoms.SpeciesAndCoordinates>
```

Output files

After finishing the calculation of 'Methane.dat', 11 files and one directory are generated.

<code>met.std</code>	standard output of the SCF calculation
<code>met.out</code>	input file and standard output
<code>met.xyz</code>	final geometrical structure
<code>met.ene</code>	values computed at every MD step
<code>met.md</code>	geometrical structures at every MD step
<code>met.md2</code>	geometrical structure of the final MD step
<code>met.cif</code>	cif file of the initial structure for Material Studio
<code>met.tden.cube</code>	total electron density in the Gaussian cube format
<code>met.v0.cube</code>	Kohn-Sham potential in the Gaussian cube format
<code>met.vhart.cube</code>	Hartree potential in the Gaussian cube format
<code>met.dden.cube</code>	difference electron density measured from atomic density
<code>met_rst/</code>	directory storing restart files

Contents of met.out

History of SCF

```
*****
*****
SCF history at MD= 1
*****
*****
```

SCF=	1	NormRD=	1.000000000000	Uele=	-3.523143659974
SCF=	2	NormRD=	0.567253699744	Uele=	-4.405605131921
SCF=	3	NormRD=	0.103433490729	Uele=	-3.982266241934
SCF=	4	NormRD=	0.024234990593	Uele=	-3.906896836134
SCF=	5	NormRD=	0.011006215697	Uele=	-3.893084558820
SCF=	6	NormRD=	0.006494145332	Uele=	-3.890357113476
SCF=	7	NormRD=	0.002722267527	Uele=	-3.891668816209
SCF=	8	NormRD=	0.000000672350	Uele=	-3.889285164733
SCF=	9	NormRD=	0.000000402419	Uele=	-3.889285102456
SCF=	10	NormRD=	0.000000346348	Uele=	-3.889285101128
SCF=	11	NormRD=	0.000000515395	Uele=	-3.889285101063

KS eigenvalues

```
*****
*****
Eigenvalues (Hartree) for SCF KS-eq.
*****
*****
```

Chemical Potential (Hartree) = 0.000000000000
Number of States = 8.000000000000
HOMO = 4
Eigenvalues

	Up-spin	Down-spin
1	-0.69897190537228	-0.69897190537228
2	-0.41522646150979	-0.41522646150979
3	-0.41522645534084	-0.41522645534084
4	-0.41521772830844	-0.41521772830844
5	0.21218282298348	0.21218282298348
6	0.21218282358344	0.21218282358344
7	0.21227055734372	0.21227055734372
8	0.24742493684297	0.24742493684297

Total energy

```
*****
Total energy (Hartree) at MD = 1
*****
```

Uele.	-3.889285101063
Ukin.	5.533754016241
UHO.	-14.855520072374
UH1.	0.041395625260
Una.	-5.040583803800
Unl.	-0.134640939010
Uxc0.	-1.564720823137
Uxc1.	-1.564720823137
Ucore.	9.551521413583
Uhub.	0.000000000000
Ucs.	0.000000000000
Uzs.	0.000000000000
Uzo.	0.000000000000
Uef.	0.000000000000
UvdW	0.000000000000
Utot.	-8.033515406373

Mulliken population

```
*****
*****
Mulliken populations
*****
*****
```

Total spin S = 0.000000000000

	Up spin	Down spin	Sum	Diff
1 C	2.509755704	2.509755704	5.019511408	0.000000000
2 H	0.372561098	0.372561098	0.745122197	0.000000000
3 H	0.372561019	0.372561019	0.745122038	0.000000000
4 H	0.372561127	0.372561127	0.745122254	0.000000000
5 H	0.372561051	0.372561051	0.745122102	0.000000000
Sum of MuLP:	up = 4.00000	down = 4.00000		
	total= 8.00000	ideal(neutral)= 8.00000		

Standard output

```

2   H  MulP   0.2867  0.2867 sum   0.5733
3   H  MulP   0.2867  0.2867 sum   0.5733
4   H  MulP   0.2867  0.2867 sum   0.5733
5   H  MulP   0.2867  0.2867 sum   0.5733
Sum of MulP: up   =    4.00000 down   =    4.00000
              total=    8.00000 ideal(neutral)=    8.00000
<DFT> Total Spin Moment (muB) = 0.000000000000
<DFT> Mixing_weight= 0.028087938921
<DFT> Uele   = -4.381387140168 dUele   = 0.188324995269
<DFT> NormRD = 0.541215648239 Criterion = 0.000000000100

```

By looking 'met.std' storing the standard output, one can confirm how the SCF proceeds.

```

***** MD= 1 SCF= 5 *****
<Poisson> Poisson's equation using FFT...
<Set_Hamiltonian> Hamiltonian matrix for VNA+dvH+Vxc...
<Cluster> Solving the eigenvalue problem...
1   C  MulP   2.8340  2.8340 sum   5.6681
2   H  MulP   0.2915  0.2915 sum   0.5830
3   H  MulP   0.2915  0.2915 sum   0.5830
4   H  MulP   0.2915  0.2915 sum   0.5830
5   H  MulP   0.2915  0.2915 sum   0.5830
Sum of MulP: up   =    4.00000 down   =    4.00000
              total=    8.00000 ideal(neutral)=    8.00000
<DFT> Total Spin Moment (muB) = 0.000000000000
<DFT> Mixing_weight= 0.028087938921
<DFT> Uele   = -4.352426233354 dUele   = 0.028960906814
<DFT> NormRD = 0.509921689438 Criterion = 0.000000000100

```

Steps of MD and SCF

Mulliken population of each atom

Difference between the current and previous steps in the KS-band energy

```

***** MD= 1 SCF= 6 *****
<Poisson> Poisson's equation using FFT...
<Set_Hamiltonian> Hamiltonian matrix for VNA+dvH+Vxc...
<Cluster> Solving the eigenvalue problem...
1   C  MulP   2.5075  2.5075 sum   5.0150
2   H  MulP   0.3731  0.3731 sum   0.7463
3   H  MulP   0.3731  0.3731 sum   0.7463
4   H  MulP   0.3731  0.3731 sum   0.7463
5   H  MulP   0.3731  0.3731 sum   0.7463
Sum of MulP: up   =    4.00000 down   =    4.00000
              total=    8.00000 ideal(neutral)=    8.00000
<DFT> Total Spin Moment (muB) = 0.000000000000
<DFT> Mixing_weight= 0.028087938921
<DFT> Uele   = -3.886371199687 dUele   = 0.466055033667
<DFT> NormRD = 0.004026023251 Criterion = 0.000000000100

```

Norm of the difference between the input and output electron densities

Convergence criterion for SCF

```

***** MD= 1 SCF= 7 *****
<Poisson> Poisson's equation using FFT...
<Set_Hamiltonian> Hamiltonian matrix for VNA+dvH+Vxc...
<Cluster> Solving the eigenvalue problem...

```

Calculation of binding energy of a hydrogen molecule

Calculation of a hydrogen atom

 ozaki@mx17:~/openmx3.9/work

```
[ozaki@mx17 work]$ mpirun -np 1 ./openmx H.dat | tee h.std
```

You can download the input files from <https://www.openmx-square.org/examples/H.dat>

Calculation of a hydrogen molecule

 ozaki@mx17:~/openmx3.9/work

```
[ozaki@mx17 work]$ mpirun -np 2 ./openmx H2.dat | tee h2.std
```

You can download the input files from <https://www.openmx-square.org/examples/H2.dat>

After finishing the two calculations, one has the following files.

 ozaki@mx17:~/openmx3.9/work

```
[ozaki@mx17 work]$ ls h.*
h.cif      h.den0.cube  h.ene  h.md2  h.sden.cube  h.tden.cube  h.v1.cube  h.xyz
h.dden.cube h.den1.cube h.md   h.out  h.std        h.v0.cube  h.vhart.cube
[ozaki@mx17 work]$ ls h2.*
h2.Dos.val  h2.dden.cube      h2.lumo0_0_r.cube  h2.out      h2.v0.cube
h2.Dos.vec  h2.ene            h2.md              h2.std      h2.vhart.cube
h2.cif      h2.homo0_0_r.cube h2.md2             h2.tden.cube h2.xyz
[ozaki@mx17 work]$
```

Total energies from 'h.out' and 'h2.out'

The total energies of the hydrogen atom and hydrogen molecule can be obtained from 'h.out' and 'h2.out', respectively.

Hydrogen atom	Utot.	-0.499351255995
---------------	-------	-----------------

Hydrogen molecule	Utot.	-1.165981886187
-------------------	-------	-----------------

Thus, the binding energy can be calculated as

$$\begin{aligned}\text{Binding energy} &= 2 \text{ H} - \text{H}_2 \\ &= 2 \times (-0.49935) - (-1.16598) \\ &= 0.1673 \text{ (Hartree)} \\ &= 4.55 \text{ (eV)}\end{aligned}$$

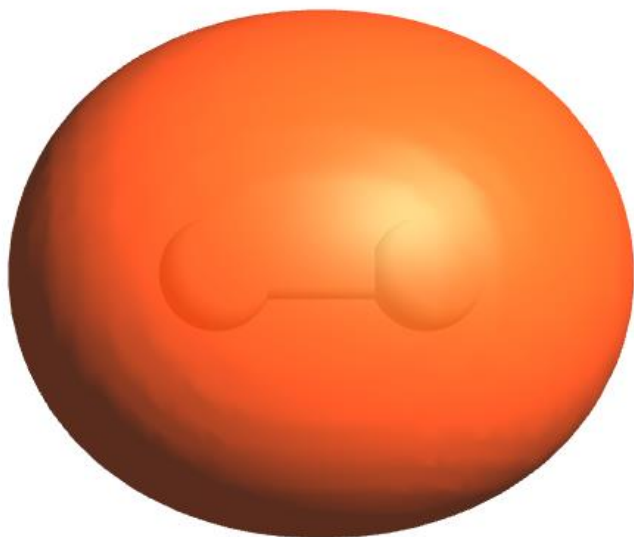
Experimental value:

4.75 (eV)

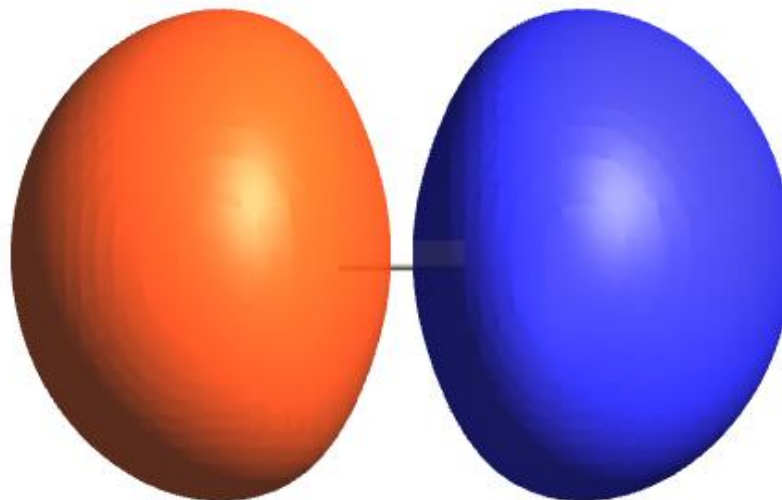
Visualization of cube files by OpenMX Viewer

After the calculation of the hydrogen molecule, the cube files for HOMO and LUMO are generated, which can be visualized by OpenMX Viewer or VESTA.

h2.homo0_0_r.cube



h2.lumo0_0_r.cube



Website of OpenMX Viewer: <https://www.openmx-square.org/viewer/>

Website of VESTA: <https://jp-minerals.org/vesta/en>

LCPAO method

(Linear-Combination of Pseudo Atomic Orbital Method)

One-particle KS orbital

$$\psi_{\sigma\mu}^{(\mathbf{k})}(\mathbf{r}) = \frac{1}{\sqrt{N}} \sum_{\mathbf{n}} e^{i\mathbf{R}_{\mathbf{n}} \cdot \mathbf{k}} \sum_{i\alpha} c_{\sigma\mu,i\alpha}^{(\mathbf{k})} \phi_{i\alpha}(\mathbf{r} - \tau_i - \mathbf{R}_{\mathbf{n}}),$$

is expressed by a linear combination of atomic like orbitals in the method.

$$\phi(\mathbf{r}) = Y_l^m(\hat{\mathbf{r}}) R(r)$$

Features:

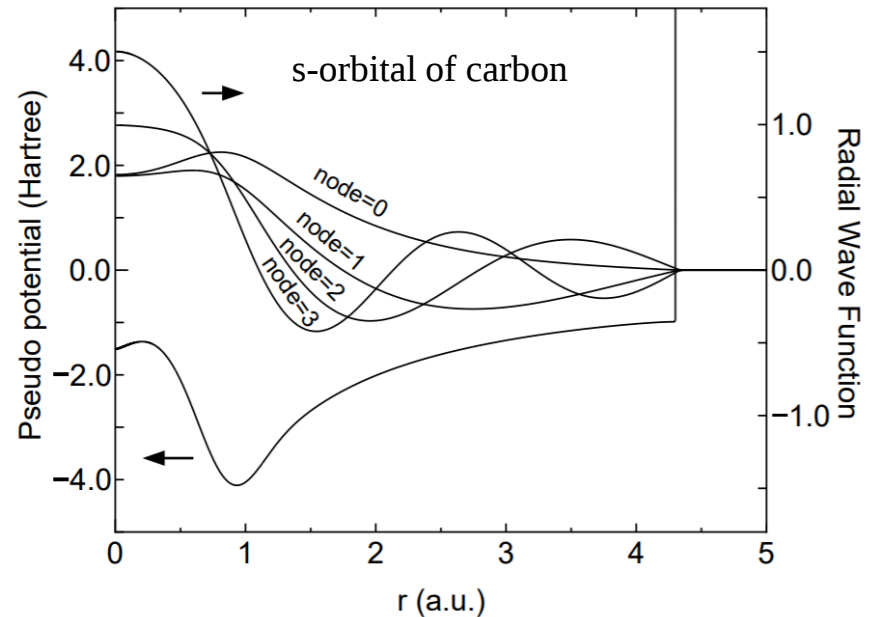
- It is easy to interpret physical and chemical meanings, since the KS orbitals are expressed by the atomic like basis functions.
- It gives rapid convergent results with respect to basis functions due to physical origin. (however, it is not a complete basis set, leading to difficulty in getting full convergence.)
- The memory and computational effort for calculation of matrix elements are $O(N)$.
- It well matches the idea of linear scaling methods.

Primitive pseudo-atomic orbitals

1. Solve an atomic Kohn-Sham eq. under a confinement potential:

$$V_{\text{core}}(r) = \begin{cases} -\frac{Z}{r} & \text{for } r \leq r_1 \\ \sum_{n=0}^3 b_n r^n & \text{for } r_1 < r \leq r_c \\ h & \text{for } r_c < r, \end{cases}$$

2. Construct the norm-conserving pseudopotentials.
3. Solve ground and excited states for the the pseudopotential for each L-channel.



Ozaki, PRB 67, 155108 (2003).

Ozaki and Kino, PRB 69, 195113 (2004).

In most cases, the accuracy and efficiency can be controlled by

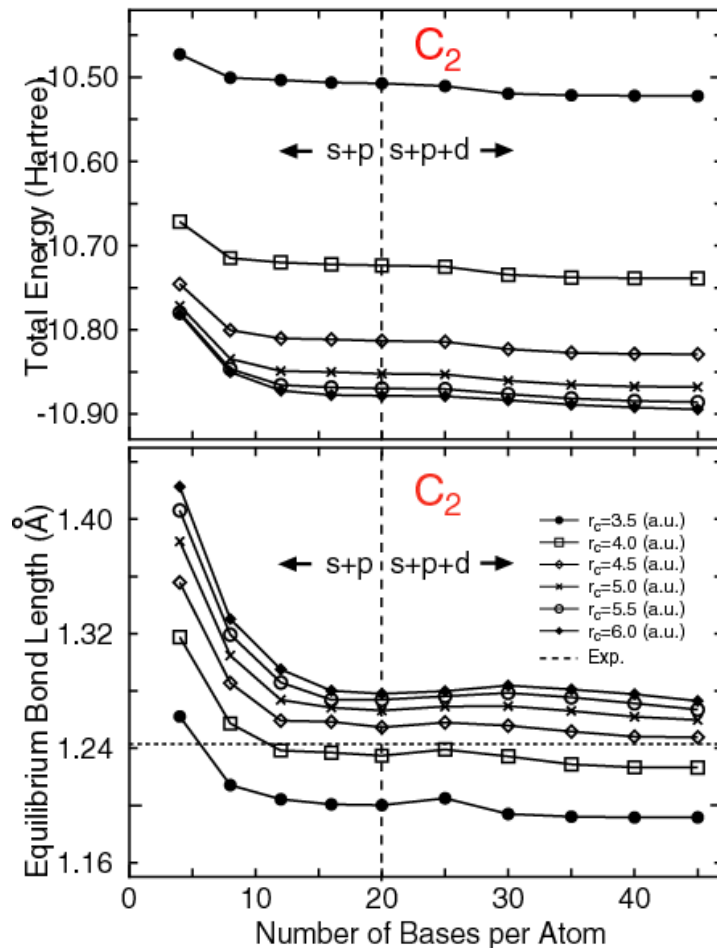
Cutoff radius

Number of orbitals

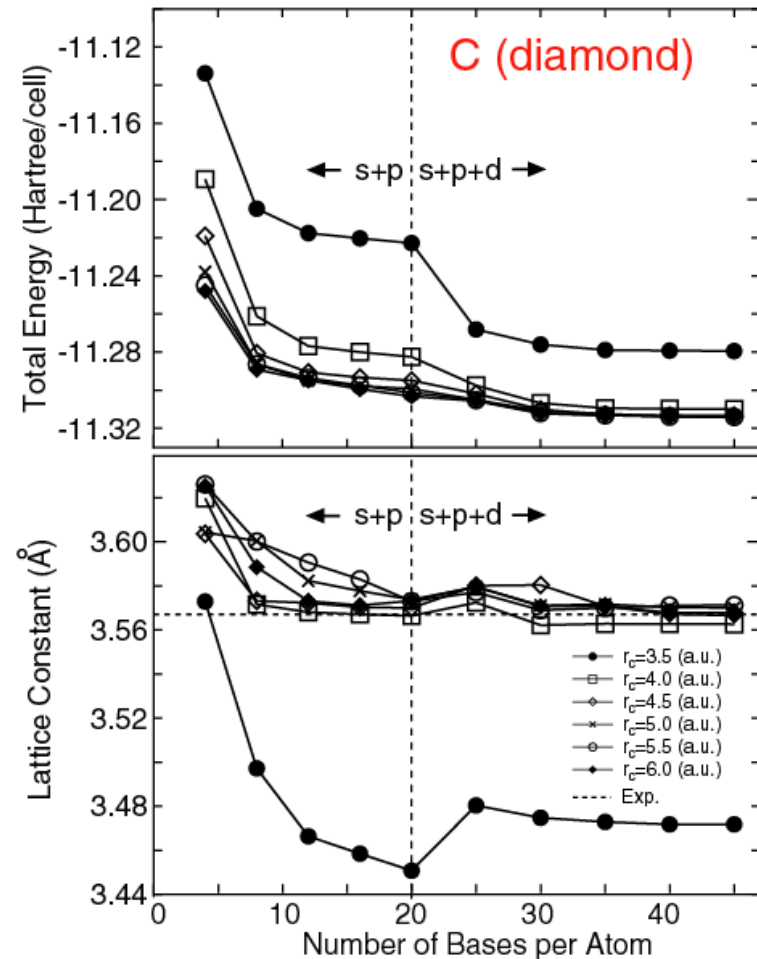
Convergence with respect to basis functions

The two parameters can be regarded as variational parameters.

Molecule



Bulk



Benchmark of primitive basis functions

Ground state calculations of dimer using primitive basis functions

Dimer	Expt.	Calc.	Dimer	Expt.	Calc.
H ₂ (H4.5- <i>s</i> 2)	$1\Sigma_g^+ a$	$1\Sigma_g^+ (1s\sigma_g^2)$	K ₂ (K10.0- <i>s</i> 2 <i>p</i> 2)	$1\Sigma_g^+ f$	$1\Sigma_g^+ (3p\pi_g^4 3p\sigma_u^2 4s\sigma_g^2)$
He ₂ (He7.0- <i>s</i> 2)	$1\Sigma_g^+ b$	$1\Sigma_g^+ (1s\sigma_g^2 1s\sigma_u^2)$	CaO (Ca7.0- <i>s</i> 2 <i>p</i> 2 <i>d</i> 2)	$1\Sigma^+ k$	$1\Sigma^+ (s\sigma^2 s\sigma^2 p\pi^4)$
Li ₂ (Li8.0- <i>s</i> 2)	$1\Sigma_g^+ c$	$1\Sigma_g^+ (2s\sigma_g^2)$	ScO (Sc7.0- <i>s</i> 2 <i>p</i> 2 <i>d</i> 2)	$2\Sigma^+ l$	$2\Sigma^+ (d\pi^4 s\sigma^2 s\sigma^1)$
BeO (Be6.0- <i>s</i> 2 <i>p</i> 2)	$1\Sigma^+ d$	$1\Sigma^+ (s\sigma^2 s\sigma^2 p\pi^4)$	Ti ₂ (Ti7.0- <i>s</i> 2 <i>p</i> 2 <i>d</i> 2)	$3\Delta_g m$	$3\Delta_g (4s\sigma_g^2 3d\sigma_g^1 3d\pi_u^4 3d\delta_g^1)$
B ₂ (B5.5- <i>s</i> 2 <i>p</i> 2)	$3\Sigma_g^- e$	$3\Sigma_g^- (2s\sigma_g^2 2s\sigma_u^2 2\pi_u^2)$	V ₂ (V7.5- <i>s</i> 2 <i>p</i> 2 <i>d</i> 2)	$3\Sigma_g^- n$	$1\Sigma_g^+ (4s\sigma_g^2 3d\sigma_g^2 3d\pi_u^4 3d\delta_g^2)$
C ₂ (C5.0- <i>s</i> 2 <i>p</i> 2)	$1\Sigma_g^+ f$	$1\Sigma_g^+ (2s\sigma_g^2 2s\sigma_u^2 2p\pi_u^4)$	V ₂ (V7.5- <i>s</i> 4 <i>p</i> 4 <i>d</i> 4 <i>f</i> 2)	$3\Sigma_g^- n$	$3\Sigma_g^- (4s\sigma_g^2 3d\sigma_g^2 3d\pi_u^4 3d\delta_g^2)$
N ₂ (N5.0- <i>s</i> 2 <i>p</i> 2)	$1\Sigma_g^+ f$	$1\Sigma_g^+ (2s\sigma_u^2 2p\pi_u^4 2p\sigma_g^2)$	Cr ₂ (Cr7.0- <i>s</i> 2 <i>p</i> 2 <i>d</i> 2)	$1\Sigma_g^+ o$	$1\Sigma_g^+ (4s\sigma_g^2 3d\sigma_g^2 3d\pi_u^4 3d\delta_g^4)$
O ₂ (O5.0- <i>s</i> 2 <i>p</i> 2)	$3\Sigma_g^- f$	$3\Sigma_g^- (2p\sigma_g^2 2p\pi_u^4 2p\pi_g^2)$	MnO (Mn7.0- <i>s</i> 2 <i>p</i> 2 <i>d</i> 2)	$6\Sigma^+ p$	$6\Sigma^+ (d\sigma^1 d\pi^4 d\delta^2 d\pi^{*2})$
F ₂ (F5.0- <i>s</i> 2 <i>p</i> 2)	$1\Sigma_g^+ f$	$1\Sigma_g^+ (2p\sigma_g^2 2p\pi_u^4 2p\pi_g^4)$	Fe ₂ (Fe7.0- <i>s</i> 2 <i>p</i> 2 <i>d</i> 2)	$7\Delta_u q$	$7\Delta_u (4s\sigma_g^2 3d\sigma_g^2 3d\sigma_u^1 3d\pi_u^4 3d\pi_g^2 3d\delta_g^3 3d\delta_u^2)$
Ne ₂ (Ne7.0- <i>s</i> 2 <i>p</i> 2)	$1\Sigma_g^+ g$	$1\Sigma_g^+ (2p\pi_u^4 2p\pi_g^4 2p\sigma_u^2)$	Co ₂ (Co7.0- <i>s</i> 2 <i>p</i> 2 <i>d</i> 2)		$5\Delta_g (4s\sigma_g^2 3d\sigma_g^2 3d\sigma_u^1 3d\pi_u^4 3d\pi_g^2 3d\delta_g^4 3d\delta_u^3)$
Na ₂ (Na9.0- <i>s</i> 2 <i>p</i> 2)	$1\Sigma_g^+ f$	$1\Sigma_g^+ (2p\pi_g^4 2p\sigma_u^2 3s\sigma_g^2)$	Ni ₂ (Ni7.0- <i>s</i> 2 <i>p</i> 2 <i>d</i> 2)	Ωr	$3\Sigma_g^- (4s\sigma_g^2 3d\sigma_g^2 3d\sigma_u^2 3d\pi_u^4 3d\pi_g^2 3d\delta_g^4 3d\delta_u^4)$
MgO (Mg7.0- <i>s</i> 2 <i>p</i> 2)	$1\Sigma^+ h$	$1\Sigma^+ (s\sigma^2 s\sigma^2 p\pi^4)$	Cu ₂ (Cu7.0- <i>s</i> 2 <i>p</i> 2 <i>d</i> 2)	$1\Sigma_g^+ s$	$1\Sigma_g^+ (4s\sigma_g^2 3d\sigma_g^2 3d\sigma_u^2 3d\pi_u^4 3d\pi_g^2 3d\delta_g^4 3d\delta_u^4)$
Al ₂ (Al6.5- <i>s</i> 2 <i>p</i> 2)	$3\Pi_u i$	$3\Sigma_g^- (3s\sigma_g^2 3s\sigma_u^2 3p\pi_u^2)$	ZnH (Zn7.0- <i>s</i> 2 <i>p</i> 2 <i>d</i> 2)	$2\Sigma_g^+ t$	$2\Sigma_g^+ (s\sigma^2 s\sigma^{*1} d\sigma^2 d\pi^4 d\delta^4)$
Al ₂ (Al6.5- <i>s</i> 4 <i>p</i> 4 <i>d</i> 2)	$3\Pi_u i$	$3\Sigma_g^- (3s\sigma_g^2 3s\sigma_u^2 3p\pi_u^2)$	GaH (Ga7.0- <i>s</i> 2 <i>p</i> 2)	$1\Sigma^+ u$	$1\Sigma^+ (s\sigma^2 s\sigma^{*2})$
Si ₂ (Si6.5- <i>s</i> 2 <i>p</i> 2)	$3\Sigma_g^- f$	$3\Pi_u (3s\sigma_u^2 3s\sigma_g^1 3p\pi_u^3)$	GeO (Ge7.0- <i>s</i> 2 <i>p</i> 2)	$1\Sigma^+ f$	$1\Sigma^+ (ss\sigma^2 sp\sigma^2 pp\pi^4 pp\sigma^2)$
Si ₂ (Si6.5- <i>s</i> 2 <i>p</i> 2 <i>d</i> 1)	$3\Sigma_g^- f$	$3\Sigma_g^- (3s\sigma_u^2 3p\pi_u^2 3s\sigma_g^2)$	As ₂ (As7.0- <i>s</i> 2 <i>p</i> 2 <i>d</i> 1)	$1\Sigma_g^+ f$	$1\Sigma_g^+ (4s\sigma_g^2 4s\sigma_u^2 4p\sigma_g^2 4p\pi_u^4)$
P ₂ (P6.0- <i>s</i> 2 <i>p</i> 2 <i>d</i> 1)	$1\Sigma_g^+ f$	$1\Sigma_g^+ (3s\sigma_u^2 3p\sigma_g^2 3p\pi_u^4)$	Se ₂ (Se7.0- <i>s</i> 2 <i>p</i> 2 <i>d</i> 1)	$3\Sigma_g^- f$	$3\Sigma_g^- (4s\sigma_g^2 4s\sigma_u^2 4p\sigma_g^2 4p\pi_u^4 4p\pi_g^2)$
S ₂ (S6.0- <i>s</i> 2 <i>p</i> 2)	$3\Sigma_g^- f$	$3\Sigma_g^- (3p\sigma_g^2 3p\pi_u^4 3p\pi_g^2)$	Br ₂ (Br7.0- <i>s</i> 2 <i>p</i> 2 <i>d</i> 1)	$1\Sigma_g^+ f$	$1\Sigma_g^+ (4s\sigma_g^2 4s\sigma_u^2 4p\sigma_g^2 4p\pi_u^4 4p\pi_g^4)$
Cl ₂ (Cl6.0- <i>s</i> 2 <i>p</i> 2 <i>d</i> 2)	$1\Sigma_g^+ f$	$1\Sigma_g^+ (3p\sigma_g^2 3p\pi_u^4 3p\pi_g^4)$	Kr ₂ (Kr7.0- <i>s</i> 2 <i>p</i> 2)	$1\Sigma_g^+ v$	$1\Sigma_g^+ (4s\sigma_g^2 4s\sigma_u^2 4p\sigma_g^2 4p\sigma_u^2 4p\pi_u^4 4p\pi_g^4)$
Ar ₂ (Ar7.0- <i>s</i> 2 <i>p</i> 2)	$1\Sigma_g^+ j$	$1\Sigma_g^+ (3p\pi_u^4 3p\pi_g^4 3p\sigma_u^2)$			

Ozaki and Kino, PRB 69, 195113 (2004).

All the successes and failures by the LDA are reproduced by the modest size of basis functions (DNP in most cases).

Variational optimization of basis functions

One-particle wave functions

$$\psi_{\mu}(\mathbf{r}) = \sum_{i\alpha} c_{\mu,i\alpha} \phi_{i\alpha}(\mathbf{r} - \mathbf{r}_i)$$

Contracted orbitals

$$\phi_{i\alpha}(\mathbf{r}) = \sum_q a_{i\alpha q} \chi_{iq}(\mathbf{r})$$

The variation of E with respect to c with fixed a gives

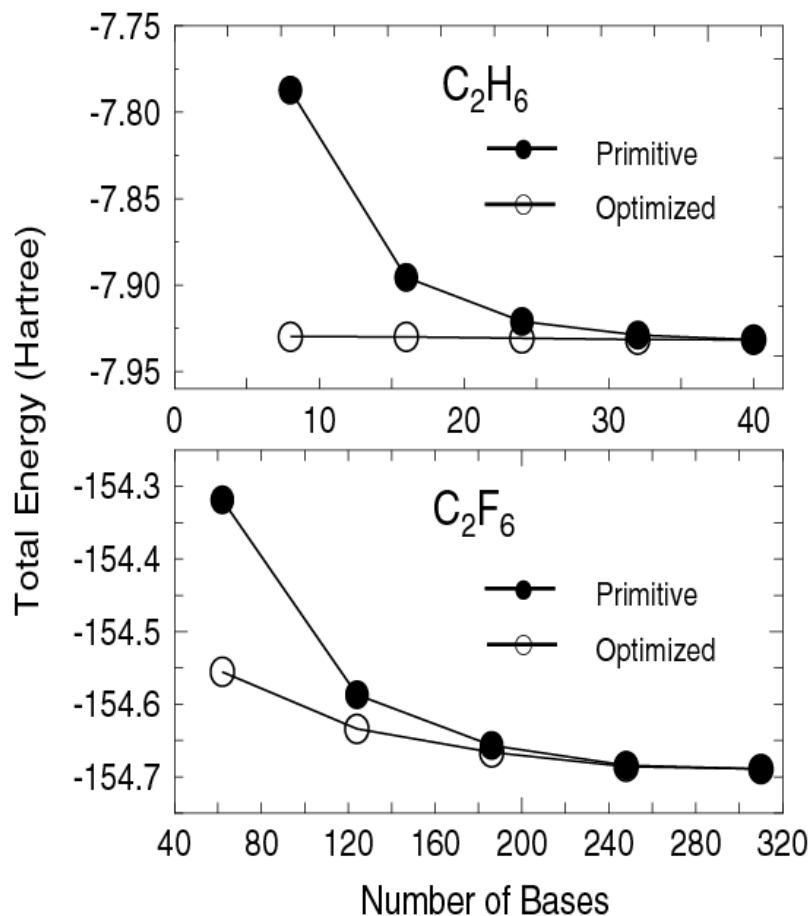
$$\partial E_{\text{tot}} / \partial c_{\mu,i\alpha} = 0 \quad \rightarrow \quad \sum_{j\beta} \langle \phi_{i\alpha} | \hat{H} | \phi_{j\beta} \rangle c_{\mu,j\beta} = \varepsilon_{\mu} \sum_{j\beta} \langle \phi_{i\alpha} | \phi_{j\beta} \rangle c_{\mu,j\beta}$$

Regarding c as dependent variables on a and assuming KS eq. is solved self-consistently with respect to c, we have

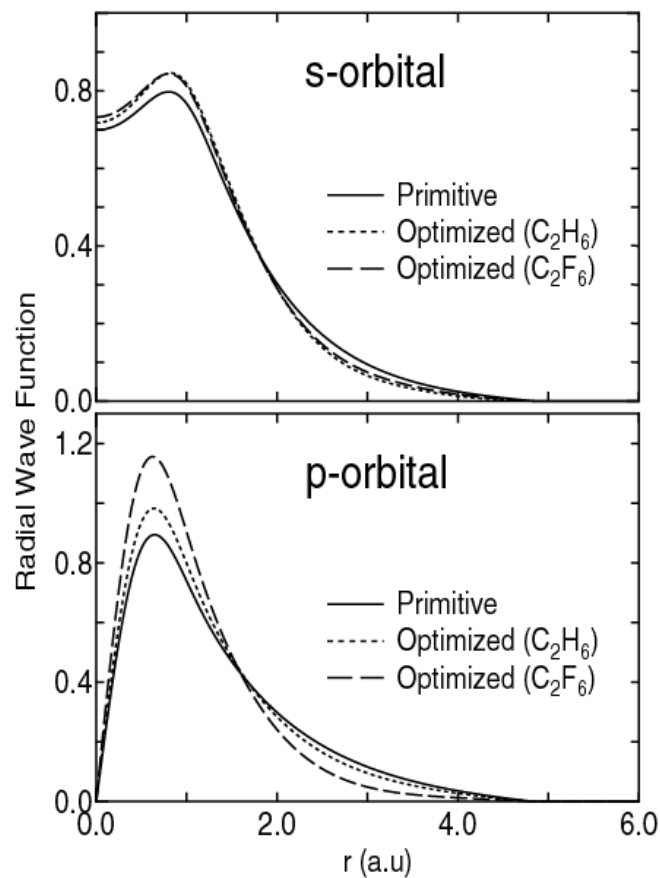
$$\begin{aligned} \frac{\partial E_{\text{tot}}}{\partial a_{i\alpha q}} &= \int d\mathbf{r}^3 \frac{\delta E_{\text{tot}}}{\delta \rho(\mathbf{r})} \frac{\delta \rho(\mathbf{r})}{\delta a_{i\alpha q}} \\ &= 2 \left(\sum_{j\beta} \Theta_{i\alpha,j\beta} \langle \chi_{iq} | \hat{H} | \phi_{j\beta} \rangle - E_{i\alpha,j\beta} \langle \chi_{iq} | \phi_{j\beta} \rangle \right) \end{aligned}$$

Primitive vs. Optimized

Energy convergence



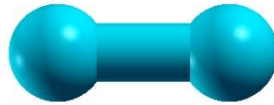
Radial shape in carbon atom



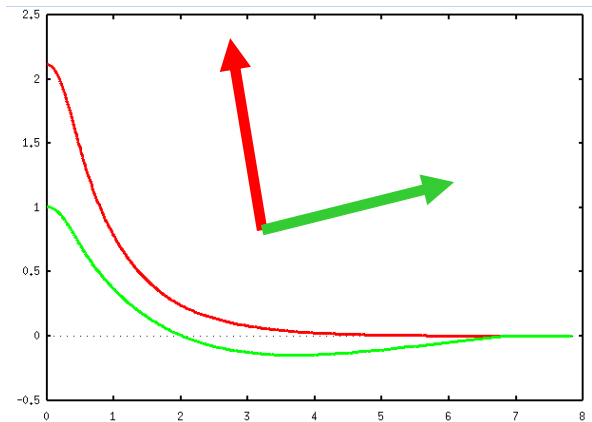
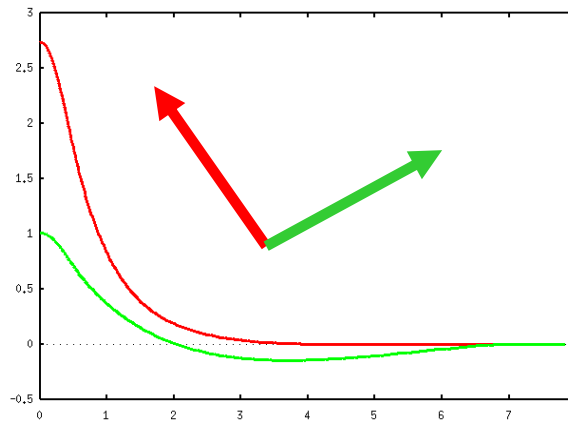
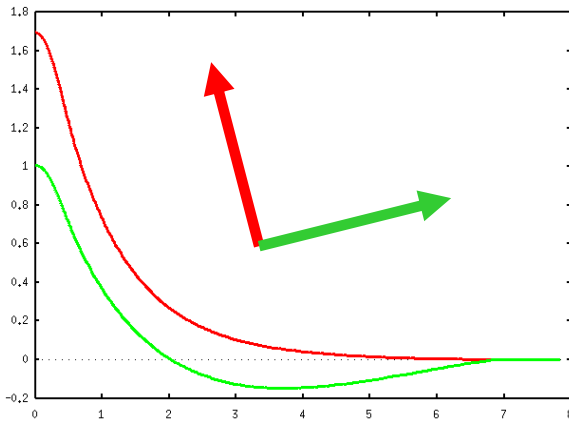
Fluorine atoms pull electrons out of the p orbitals of carbon atoms. So, the p orbital of the carbon atom is largely localized.

Optimization of basis functions

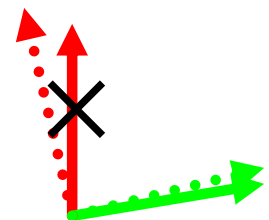
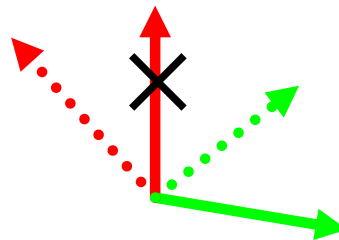
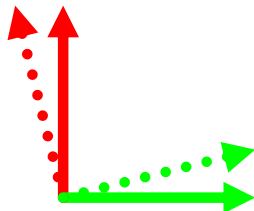
1. Choose typical chemical environments



2. Optimize variationally the radial functions



3. Rotate a set of optimized orbitals within the subspace, and discard the redundant functions



Database of optimized VPS and PAO

Database (2019) of optimized VPS and PAO

The database (2019) of fully relativistic pseudopotentials (VPS) and pseudo-atomic orbitals (PAO), generated by ADPACK, which could be an input data of program package, OpenMX. The data of elements with the underline are currently available. When you use these data, VPS and PAO, in the program package, OpenMX, then copy them to the directory, openmx*/DFT_DATA19/VPS/ and openmx*/DFT_DATA19/PAO/, respectively. The delta gauge of OpenMX with the database (2019) is found at [here](#).

<u>E</u>	Guideline for choice of PAOs																
<u>H</u>	See also the pages 51-60 in the tutorial material																<u>He</u>
<u>Li</u>	<u>Be</u>											<u>B</u>	<u>C</u>	<u>N</u>	<u>O</u>	<u>F</u>	<u>Ne</u>
<u>Na</u>	<u>Mg</u>											<u>Al</u>	<u>Si</u>	<u>P</u>	<u>S</u>	<u>Cl</u>	<u>Ar</u>
<u>K</u>	<u>Ca</u>	<u>Sc</u>	<u>Ti</u>	<u>V</u>	<u>Cr</u>	<u>Mn</u>	<u>Fe</u>	<u>Co</u>	<u>Ni</u>	<u>Cu</u>	<u>Zn</u>	<u>Ga</u>	<u>Ge</u>	<u>As</u>	<u>Se</u>	<u>Br</u>	<u>Kr</u>
<u>Rb</u>	<u>Sr</u>	<u>Y</u>	<u>Zr</u>	<u>Nb</u>	<u>Mo</u>	<u>Tc</u>	<u>Ru</u>	<u>Rh</u>	<u>Pd</u>	<u>Ag</u>	<u>Cd</u>	<u>In</u>	<u>Sn</u>	<u>Sb</u>	<u>Te</u>	<u>I</u>	<u>Xe</u>
<u>Cs</u>	<u>Ba</u>	<u>L</u>	<u>Hf</u>	<u>Ta</u>	<u>W</u>	<u>Re</u>	<u>Os</u>	<u>Ir</u>	<u>Pt</u>	<u>Au</u>	<u>Hg</u>	<u>Tl</u>	<u>Pb</u>	<u>Bi</u>	<u>Po</u>	At	<u>Rn</u>
Fr	Ra	<u>A</u>															
	<u>L</u>	<u>La</u>	<u>Ce</u>	<u>Pr</u>	<u>Nd</u>	<u>Pm</u>	<u>Sm</u>	Eu	Gd	Tb	<u>Dy</u>	<u>Ho</u>	Er	Tm	Yb	<u>Lu</u>	
	<u>A</u>	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	

Implementation: Total energy (1)

The total energy is given by the sum of six terms, and a proper integration scheme for each term is applied to accurately evaluate the total energy.

$$E_{\text{tot}} = E_{\text{kin}} + E_{\text{ec}} + E_{\text{ee}} + E_{\text{xc}} + E_{\text{cc}} = E_{\text{kin}} + E_{\text{na}} + E_{\text{ec}}^{(\text{NL})} + E_{\delta\text{ee}} + E_{\text{xc}} + E_{\text{scc}}.$$

$$E_{\text{kin}} = \sum_{\sigma} \sum_{\mathbf{n}} \sum_{i\alpha, j\beta} \rho_{\sigma, i\alpha j\beta}^{(\mathbf{R}_{\mathbf{n}})} h_{i\alpha j\beta, \text{kin}}^{(\mathbf{R}_{\mathbf{n}})}. \quad \text{Kinetic energy}$$

$$\begin{aligned} E_{\text{ec}} &= E_{\text{ec}}^{(\text{L})} + E_{\text{ec}}^{(\text{NL})}, \quad \text{Coulomb energy with external potential} \\ &= \sum_{\sigma} \sum_{\mathbf{n}} \sum_{i\alpha, j\beta} \rho_{\sigma, i\alpha j\beta}^{(\mathbf{R}_{\mathbf{n}})} \langle \phi_{i\alpha}(\mathbf{r} - \tau_i) | \sum_I V_{\text{core}, I}(\mathbf{r} - \tau_I) | \phi_{j\beta}(\mathbf{r} - \tau_j - \mathbf{R}_{\mathbf{n}}) \rangle \\ &\quad + \sum_{\sigma} \sum_{\mathbf{n}} \sum_{i\alpha, j\beta} \rho_{\sigma, i\alpha j\beta}^{(\mathbf{R}_{\mathbf{n}})} \langle \phi_{i\alpha}(\mathbf{r} - \tau_i) | \sum_I V_{\text{NL}, I}(\mathbf{r} - \tau_I) | \phi_{j\beta}(\mathbf{r} - \tau_j - \mathbf{R}_{\mathbf{n}}) \rangle, \end{aligned}$$

$$\begin{aligned} E_{\text{ee}} &= \frac{1}{2} \int d\mathbf{r}^3 n(\mathbf{r}) V_{\text{H}}(\mathbf{r}), \quad \text{Hartree energy} \\ &= \frac{1}{2} \int d\mathbf{r}^3 n(\mathbf{r}) \{ V_{\text{H}}^{(\text{a})}(\mathbf{r}) + \delta V_{\text{H}}(\mathbf{r}) \}, \end{aligned}$$

$$E_{\text{xc}} = \int d\mathbf{r}^3 \{ n_{\uparrow}(\mathbf{r}) + n_{\downarrow}(\mathbf{r}) + n_{\text{pcc}}(\mathbf{r}) \} \epsilon_{\text{xc}}(n_{\uparrow} + \frac{1}{2}n_{\text{pcc}}, n_{\downarrow} + \frac{1}{2}n_{\text{pcc}}), \quad \text{Exchange-correlation energy}$$

$$E_{\text{cc}} = \frac{1}{2} \sum_{I, J} \frac{Z_I Z_J}{|\tau_I - \tau_J|}. \quad \text{Core-core Coulomb energy}$$

Implementation: Total energy (2)

The reorganization of Coulomb energies gives three new energy terms.

$$E_{\text{ec}}^{(L)} + E_{\text{ee}} + E_{\text{cc}} = E_{\text{na}} + E_{\delta\text{ee}} + E_{\text{scc}},$$

The neutral atom energy

$$E_{\text{na}} = \int d\mathbf{r}^3 n(\mathbf{r}) \sum_I V_{\text{na},I}(\mathbf{r} - \tau_I),$$

Short range and separable to two-center integrals

$$= \sum_{\sigma} \sum_{\mathbf{n}} \sum_{i\alpha, j\beta} \rho_{\sigma, i\alpha j\beta}^{(\mathbf{R}_{\mathbf{n}})} \sum_I \langle \phi_{i\alpha}(\mathbf{r} - \tau_i) | V_{\text{na},I}(\mathbf{r} - \tau_I) | \phi_{j\beta}(\mathbf{r} - \tau_j - \mathbf{R}_{\mathbf{n}}) \rangle,$$

Difference charge Hartree energy

$$E_{\delta\text{ee}} = \frac{1}{2} \int d\mathbf{r}^3 \delta n(\mathbf{r}) \delta V_{\text{H}}(\mathbf{r}),$$

Long range but minor contribution

Screened core-core repulsion energy

$$E_{\text{scc}} = \frac{1}{2} \sum_{I,J} \left[\frac{Z_I Z_J}{|\tau_I - \tau_J|} - \int d\mathbf{r}^3 n_I^{(\text{a})}(\mathbf{r}) V_{\text{H},J}^{(\text{a})}(\mathbf{r}) \right].$$

Short range and two-center integrals

Difference charge

$$\begin{aligned} \delta n(\mathbf{r}) &= n(\mathbf{r}) - n^{(\text{a})}(\mathbf{r}), \\ &= n(\mathbf{r}) - \sum_i n_i^{(\text{a})}(\mathbf{r}), \end{aligned}$$

Neutral atom potential

$$V_{\text{na},I}(\mathbf{r} - \tau_I) = V_{\text{core},I}(\mathbf{r} - \tau_I) + V_{\text{H},I}^{(\text{a})}(\mathbf{r} - \tau_I).$$

Implementation: Total energy (3)

So, the total energy is given by

$$E_{\text{tot}} = E_{\text{kin}} + E_{\text{na}} + E_{\text{ec}}^{(\text{NL})} + E_{\delta\text{ee}} + E_{\text{xc}} + E_{\text{scc}}.$$

Each term is evaluated by using a different numerical grid with consideration on accuracy and efficiency.

$$\left. \begin{array}{l} E_{\text{kin}} \\ E_{\text{na}} \\ E_{\text{ec}}^{(\text{NL})} \end{array} \right\} \text{ Spherical coordinate in momentum space}$$

$$\left. \begin{array}{l} E_{\delta\text{ee}} \\ E_{\text{xc}} \end{array} \right\} \text{ Real space regular mesh}$$

$$E_{\text{scc}} \quad \text{Real space fine mesh}$$

Two center integrals

Fourier-transformation of basis functions

$$\begin{aligned}
 \tilde{\phi}_{i\alpha}(\mathbf{k}) &= \left(\frac{1}{\sqrt{2\pi}}\right)^3 \int d\mathbf{r}^3 \phi_{i\alpha}(\mathbf{r}) e^{-i\mathbf{k}\cdot\mathbf{r}} \\
 &= \left(\frac{1}{\sqrt{2\pi}}\right)^3 \int d\mathbf{r}^3 Y_{lm}(\hat{\mathbf{r}}) R_{pl}(r) \left\{ 4\pi \sum_{L=0}^{\infty} \sum_{M=-L}^L (-i)^L j_L(kr) Y_{LM}(\hat{\mathbf{k}}) Y_{LM}^*(\hat{\mathbf{r}}) \right\}, \\
 &= \left(\frac{1}{\sqrt{2\pi}}\right)^3 4\pi \sum_{L=0}^{\infty} \sum_{M=-L}^L (-i)^L Y_{LM}(\hat{\mathbf{k}}) \int dr r^2 R_{pl}(r) j_L(kr) \int d\theta d\phi \sin(\theta) Y_{lm}(\hat{\mathbf{r}}) Y_{LM}^*(\hat{\mathbf{r}}), \\
 &= \left[\left(\frac{1}{\sqrt{2\pi}}\right)^3 4\pi (-i)^l \int dr r^2 R_{pl}(r) j_L(kr) \right] Y_{lm}(\hat{\mathbf{k}}), \\
 &= \tilde{R}_{pl}(k) Y_{lm}(\hat{\mathbf{k}}),
 \end{aligned}$$

Integrals for angular parts are analytically performed. Thus, we only have to perform one-dimensional integrals along the radial direction.

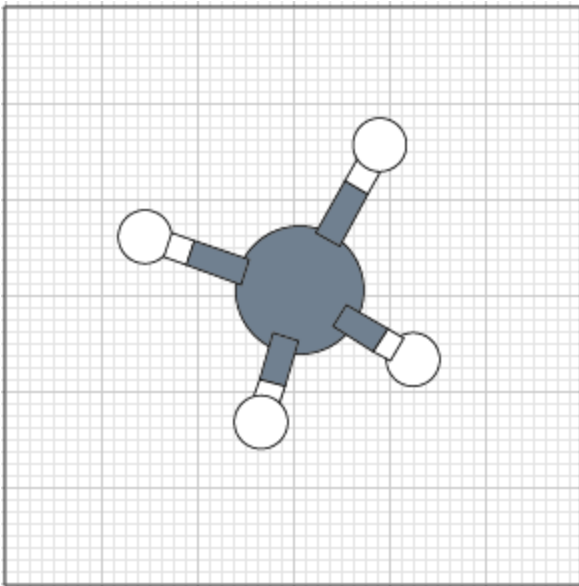
e.g., overlap integral

$$\begin{aligned}
 \langle \phi_{i\alpha}(\mathbf{r}) | \phi_{j\beta}(\mathbf{r} - \tau) \rangle &= \int d\mathbf{r}^3 \phi_{i\alpha}^*(\mathbf{r}) \phi_{j\beta}(\mathbf{r} - \tau), \\
 &= \int d\mathbf{r}^3 \left(\frac{1}{\sqrt{2\pi}}\right)^3 \int dk^3 \tilde{R}_{pl}^*(k) Y_{lm}^*(\hat{\mathbf{k}}) e^{-i\mathbf{k}\cdot\mathbf{r}} \left(\frac{1}{\sqrt{2\pi}}\right)^3 \int dk'^3 \tilde{R}_{p'l'}(k') Y_{l'm'}(\hat{\mathbf{k}}') e^{i\mathbf{k}'\cdot(\mathbf{r}-\tau)}, \\
 &= \left(\frac{1}{2\pi}\right)^3 \int dk^3 \int dk'^3 e^{-i\mathbf{k}'\cdot\tau} \tilde{R}_{pl}^*(k) Y_{lm}^*(\hat{\mathbf{k}}) \tilde{R}_{p'l'}(k') Y_{l'm'}(\hat{\mathbf{k}}') \int d\mathbf{r}^3 e^{i(\mathbf{k}'-\mathbf{k})\cdot\mathbf{r}}, \\
 &= \int dk^3 e^{-i\mathbf{k}\cdot\tau} \tilde{R}_{pl}^*(k) Y_{lm}^*(\hat{\mathbf{k}}) \tilde{R}_{p'l'}(k) Y_{l'm'}(\hat{\mathbf{k}}),
 \end{aligned}$$

Cutoff energy for regular mesh

The two energy components $E_{\delta ee} + E_{xc}$ are calculated on real space regular mesh. The mesh fineness is determined by plane-wave cutoff energies.

```
scf.energycutoff      150.0      # default=150 (Ry)
```



The cutoff energy can be related to the mesh fineness by the following eqs.

$$E_{\text{cut}}^{(1)} = \frac{1}{2} \mathbf{g}\mathbf{b}_1 \cdot \mathbf{g}\mathbf{b}_1, \quad E_{\text{cut}}^{(2)} = \frac{1}{2} \mathbf{g}\mathbf{b}_2 \cdot \mathbf{g}\mathbf{b}_2, \quad E_{\text{cut}}^{(3)} = \frac{1}{2} \mathbf{g}\mathbf{b}_3 \cdot \mathbf{g}\mathbf{b}_3,$$

$$\mathbf{g}\mathbf{a}_1 = \frac{\mathbf{a}_1}{N_1}, \quad \mathbf{g}\mathbf{a}_2 = \frac{\mathbf{a}_2}{N_2}, \quad \mathbf{g}\mathbf{a}_3 = \frac{\mathbf{a}_3}{N_3},$$

$$\mathbf{g}\mathbf{b}_1 = 2\pi \frac{\mathbf{g}\mathbf{a}_2 \times \mathbf{g}\mathbf{a}_3}{\Delta V}, \quad \mathbf{g}\mathbf{b}_2 = 2\pi \frac{\mathbf{g}\mathbf{a}_3 \times \mathbf{g}\mathbf{a}_1}{\Delta V}, \quad \mathbf{g}\mathbf{b}_3 = 2\pi \frac{\mathbf{g}\mathbf{a}_1 \times \mathbf{g}\mathbf{a}_2}{\Delta V},$$

$$\Delta V = \mathbf{g}\mathbf{a}_1 \cdot (\mathbf{g}\mathbf{a}_2 \times \mathbf{g}\mathbf{a}_3),$$

Forces

$$\begin{aligned} \mathbf{F}_i &= -\frac{\partial E_{\text{tot}}}{\partial \mathbf{R}_i} \\ &= -\boxed{\frac{\partial E_{\text{kin}}}{\partial \mathbf{R}_i}} - \boxed{\frac{\partial E_{\text{na}}}{\partial \mathbf{R}_i}} - \boxed{\frac{\partial E_{\text{scc}}}{\partial \mathbf{R}_i}} - \boxed{\frac{\partial E_{\delta\text{ee}}}{\partial \mathbf{R}_i}} - \boxed{\frac{\partial E_{\text{xc}}}{\partial \mathbf{R}_i}} - \boxed{\frac{\partial E_{\text{cc}}}{\partial \mathbf{R}_i}} \end{aligned}$$

$$\frac{\partial E_{\delta\text{ee}}}{\partial \mathbf{R}_k} = \sum_{\mathbf{p}} \frac{\partial n(\mathbf{r}_{\mathbf{p}})}{\partial \mathbf{R}_k} \frac{\partial E_{\delta\text{ee}}}{\partial n(\mathbf{r}_{\mathbf{p}})} + \sum_{\mathbf{p}} \frac{\partial n^{\text{a}}(\mathbf{r}_{\mathbf{p}})}{\partial \mathbf{R}_k} \frac{\partial E_{\delta\text{ee}}}{\partial n^{\text{a}}(\mathbf{r}_{\mathbf{p}})}.$$

$$\begin{aligned} \frac{\partial E_{\delta\text{ee}}}{\partial n(\mathbf{r}_{\mathbf{p}})} &= \frac{1}{2} \Delta V \{ \delta V_{\text{H}}(\mathbf{r}_{\mathbf{p}}) + \sum_{\mathbf{q}} \delta n(\mathbf{r}_{\mathbf{q}}) \frac{\partial \delta V_{\text{H}}(\mathbf{r}_{\mathbf{q}})}{\partial n(\mathbf{r}_{\mathbf{p}})} \}, \\ &= \frac{1}{2} \Delta V \{ \delta V_{\text{H}}(\mathbf{r}_{\mathbf{p}}) + \frac{4\pi}{N_{\text{rsg}}} \sum_{\mathbf{G}} \frac{1}{|\mathbf{G}|^2} \sum_{\mathbf{q}} \delta n(\mathbf{r}_{\mathbf{q}}) e^{i\mathbf{G} \cdot (\mathbf{r}_{\mathbf{q}} - \mathbf{r}_{\mathbf{p}})} \}, \\ &= \Delta V \delta V_{\text{H}}(\mathbf{r}_{\mathbf{p}}). \end{aligned}$$



Easy calc.



See the left

$$\begin{aligned} \frac{\partial E_{\delta\text{ee}}}{\partial n^{\text{a}}(\mathbf{r}_{\mathbf{p}})} &= -\frac{1}{2} \Delta V \{ \delta V_{\text{H}}(\mathbf{r}_{\mathbf{p}}) - \sum_{\mathbf{q}} \delta n(\mathbf{r}_{\mathbf{q}}) \frac{\partial \delta V_{\text{H}}(\mathbf{r}_{\mathbf{q}})}{\partial n^{\text{a}}(\mathbf{r}_{\mathbf{p}})} \}, \\ &= -\frac{1}{2} \Delta V \{ \delta V_{\text{H}}(\mathbf{r}_{\mathbf{p}}) + \frac{4\pi}{N_{\text{rsg}}} \sum_{\mathbf{G}} \frac{1}{|\mathbf{G}|^2} \sum_{\mathbf{q}} \delta n(\mathbf{r}_{\mathbf{q}}) e^{i\mathbf{G} \cdot (\mathbf{r}_{\mathbf{q}} - \mathbf{r}_{\mathbf{p}})} \}, \\ &= -\Delta V \delta V_{\text{H}}(\mathbf{r}_{\mathbf{p}}). \end{aligned}$$

$$\begin{aligned} \frac{\partial n(\mathbf{r}_{\mathbf{p}})}{\partial \mathbf{R}_k} &= \sum_{i\alpha,j\beta} \sum_{\nu} \{ \frac{\partial c_{i\alpha,\nu}^*}{\partial \mathbf{R}_k} c_{j\beta,\nu} \chi_{i\alpha}(\mathbf{r}) \chi_{j\beta}(\mathbf{r}) + c_{i\alpha,\nu}^* \frac{\partial c_{j\beta,\nu}}{\partial \mathbf{R}_k} \chi_{i\alpha}(\mathbf{r}_{\mathbf{p}}) \chi_{j\beta}(\mathbf{r}_{\mathbf{p}}) \} \\ &\quad + 2 \sum_{\alpha,j\beta} \rho_{k\alpha,j\beta} \frac{\partial \chi_{k\alpha}(\mathbf{r}_{\mathbf{p}})}{\partial \mathbf{R}_k} \chi_{j\beta}(\mathbf{r}_{\mathbf{p}}). \end{aligned}$$

$$\begin{aligned} \frac{\partial E_{\text{xc}}}{\partial \mathbf{R}_k} &= \sum_{\mathbf{p}} \frac{\partial n(\mathbf{r}_{\mathbf{p}})}{\partial \mathbf{R}_k} \frac{\partial E_{\text{xc}}}{\partial n(\mathbf{r}_{\mathbf{p}})}, \\ &= \Delta V \sum_{\mathbf{p}} \frac{\partial n(\mathbf{r}_{\mathbf{p}})}{\partial \mathbf{R}_k} v_{\text{xc}}(n(\mathbf{r}_{\mathbf{p}})). \end{aligned}$$

Forces are always analytic at any grid fineness and at zero temperature, even if numerical basis functions and numerical grids.

Norm-conserving pseudopotential by MBK

I. Morrion, D.M. Bylander, and L. Kleinman, PRB 47, 6728 (1993).

If $Q_{ij} = 0$, the non-local terms can be transformed to a diagonal form.

$$\begin{aligned} V_{\text{NL}} &= \sum_{i,j} B_{ij} |\beta_i\rangle \langle \beta_j|, \\ &= \sum_i \lambda_i |\alpha_i\rangle \langle \alpha_i| \end{aligned}$$

The form is equivalent to that obtained from the Blochl expansion for TM norm-conserving pseudopotentials. Thus, common routines can be utilized for the MBK and TM pseudopotentials, resulting in easiness of the code development.

To satisfy $Q_{ij}=0$, pseudofunctions are now given by

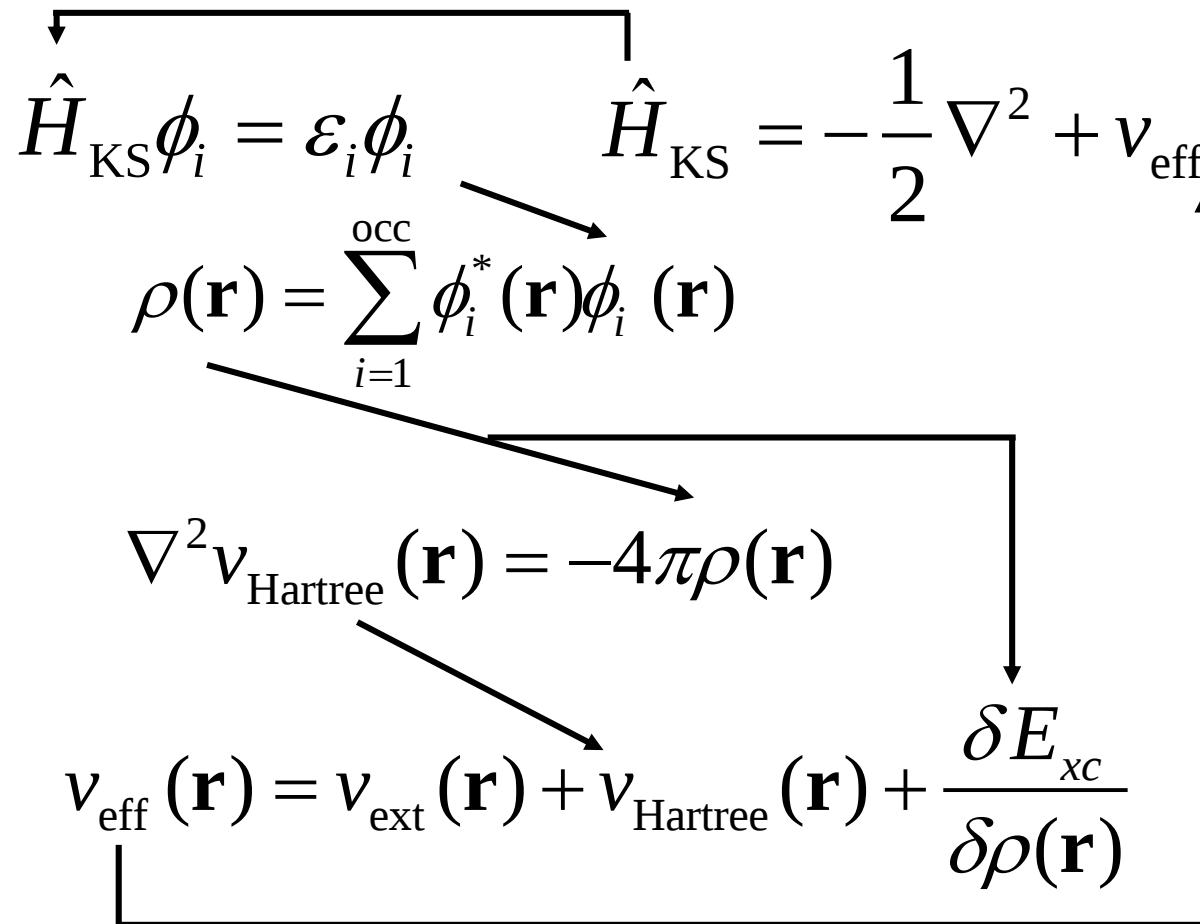
$$\phi_i = \phi_{\text{TM},i} + f_i \quad f_i = \sum_{l=0} c_l \left[r j_l \left(\frac{r}{r_c} u_{li} \right) \right]$$

The coefficients $\{c\}$ are determined by agreement of derivatives and $Q_{ij}=0$. Once a set of $\{c\}$ is determined, χ is given by

$$\chi_i = V_{\text{TM}}^{(i)} \phi_{\text{TM},i} + \varepsilon_i f_i - V_{\text{loc}} \phi_i - \frac{1}{2} \sum_i c_l \left(\frac{u_{li}}{r_c} \right)^2 \left[r j_l \left(\frac{r}{r_c} u_{li} \right) \right]$$

Algorithmic structure of KS eq.

3D coupled non-linear differential equations have to be solved self-consistently.



Input charge = Output charge $\rightarrow$ Self-consistent condition

Self-consistency: Simple charge mixing

The KS effective potential is constructed from ρ .

However, ρ is evaluated from eigenfunctions of KS eq.

$$\hat{H}_{\text{KS}}\phi_i = \varepsilon_i\phi_i \quad \hat{H}_{\text{KS}} = -\frac{1}{2}\nabla^2 + v_{\text{eff}}$$

$$v_{\text{eff}} = v_{\text{ext}}(\mathbf{r}) + v_{\text{Hartree}}(\mathbf{r}) + \frac{\delta E_{\text{xc}}}{\delta \rho(\mathbf{r})}$$

$$\rho(\mathbf{r}) = \sum_i \phi_i^*(\mathbf{r})\phi_i(\mathbf{r})$$

Simple charge mixing method

The next input density is constructed by a simple mixing of input and output densities.

$$\rho_{n+1}^{(\text{in})} = \alpha \rho_n^{(\text{in})} + (1 - \alpha) \rho_n^{(\text{out})},$$

It works well for large gap systems and small sized systems.

Self-consistency: RMM-DIIS

Idea:

Minimize the norm of a linear combination of previous residual vectors.

$$\bar{R}_{n+1} = \sum_{m=n-(p-1)}^n \alpha_m R_m,$$

$$R_n(\mathbf{q}) \equiv \tilde{n}_n^{(\text{out})}(\mathbf{q}) - \tilde{n}_n^{(\text{in})}(\mathbf{q}),$$

$$F = \langle \bar{R}_{n+1} | \bar{R}_{n+1} \rangle - \lambda \left(1 - \sum_m^n a_m \right),$$

$$\langle R_m | R_{m'} \rangle \equiv \sum_{\mathbf{q}} \frac{R_m^*(\mathbf{q}) R_{m'}(\mathbf{q})}{w(\mathbf{q})},$$

$$= \sum_{m,m'} \alpha_m \alpha_{m'} \langle R_m | R_{m'} \rangle - \lambda \left(1 - \sum_m^n a_m \right).$$

Kerker factor

$$w(\mathbf{q}) = \frac{|\mathbf{q}|^2}{|\mathbf{q}|^2 + q_0^2},$$

Minimization of F leads to

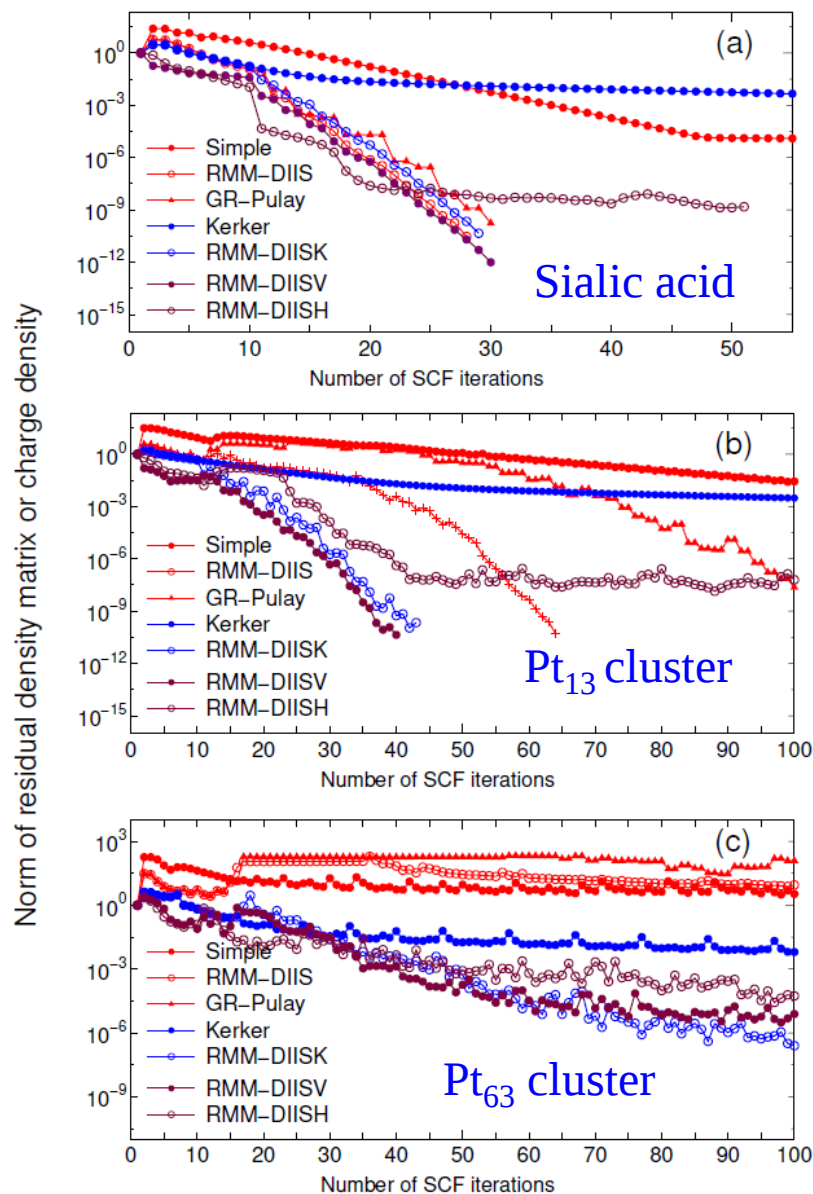
Long wave length components corresponding to small $|\mathbf{q}|$ are taken into account.

$$\frac{\partial F}{\partial \alpha_k} = 0 \quad \left(\begin{array}{ccccc} \langle R_{n-(p-1)} | R_{n-(p-1)} \rangle & \cdots & \cdots & 1 \\ \cdots & \cdots & \cdots & 1 \\ \cdots & \cdots & \langle R_n | R_n \rangle & \cdots \\ 1 & 1 & \cdots & 0 \end{array} \right) \left(\begin{array}{c} \alpha_{n-(p-1)} \\ \alpha_{n-(p-1)+1} \\ \vdots \\ \frac{1}{2}\lambda \end{array} \right) = \left(\begin{array}{c} 0 \\ 0 \\ \vdots \\ 1 \end{array} \right).$$

Optimum input density might be given by

$$\rho_{n+1}^{(\text{in})} = \sum_{m=n-(p-1)}^n \alpha_m \rho_m^{(\text{in})} + \beta \sum_{m=n-(p-1)}^n \alpha_m R_m$$

Mixing methods



Available mixing methods:

Simple mixing (Simple)

Residual minimization method in the direct inversion iterative subspace (RMM-DIIS)

Guaranteed reduction Pulay method (GR-Pulay)

Kerker mixing (Kerker)

RMM-DIIS with Kerker metric (RMM-DIISK)

RMM-DIIS for Kohn-Sham potential (RMM-DIISV)

RMM-DIIS for Hamiltonian (RMM-DIISH)

Recommendation:

RMM-DIISK or RMM-DIISV

For DFT+U and constrained methods

RMM-DIISH

See also the manual at

https://www.openmx-square.org/openmx_man3.9/node40.html

Specification of PAO and VPS

PAO and VPS are specified by the following keyword:

```
<Definition. of. Atomic. Species  
O    O7.0-s2p2d1      O_PBE19  
H    H7.0-s2p1       H_PBE19  
Definition. of. Atomic. Species>
```

- O7.0 means O7.0.pao.
- -s2p2d1 means 2, 2, and 1 radial functions are allocated to s-, p-, and d-orbitals.
- In this case, for oxygen atom, $2 \times 1 + 2 \times 3 + 1 \times 5 = 13$ basis functions are allocated.
- O_PBE19 means O_PBE19.vps.

The path for O7.0.pao and O_PBE13.vps is specified by

```
DATA.PATH    ../DFT_DATA19
```

Default value is ‘../DFT_DATA19’.

How to choose basis functions: H₂O case

By clicking H7.0.pao and O7.0.pao in the database(2019), you may find the following

https://www.openmx-square.org/vps_pao2019/H/index.html

Eigen values(Hartree) of pseudo atomic orbitals

H7.0.pao

Eigenvalues
Lmax= 3 Mul=15

mu	0 0	-0.23595211038442	↖3
mu	0 1	0.14109389991827	
mu	0 2	0.61751730037441	↖6
mu	0 3	1.31890671598573	
mu	0 4	2.24052765608302	
mu	0 5	3.37954791544661	
mu	0 6	4.73488369825610	
mu	0 7	6.30608874470710	
mu	0 8	8.09282718517299	
mu	0 9	10.09464035732420	
mu	0 10	12.31085267019158	
mu	0 11	14.74057314485273	
mu	0 12	17.38277845742691	
mu	0 13	20.23645090753857	
mu	0 14	23.30073926597344	↖5
mu	1 0	0.10914684890465	
mu	1 1	0.47776040452236	
mu	1 2	1.06988680483686	
mu	1 3	1.88261331124981	
mu	1 4	2.91175885838084	
mu	1 5	4.15601184789448	
mu	1 6	5.61454131060210	
mu	1 7	7.28681796296307	
mu	1 8	9.17254361476158	
mu	1 9	11.27156766390586	
mu	1 10	13.58381334569800	
mu	1 11	16.10921637499960	
mu	1 12	18.84767560575107	
mu	1 13	21.79902110024685	
mu	1 14	24.96300480798286	
mu	2 0	0.27851528500170	
mu	2 1	0.76970400958463	
mu	2 2	1.47576814497054	
mu	2 3	2.39881523735167	
mu	2 4	3.537132246447754	

https://www.openmx-square.org/vps_pao2019/O/index.html

Eigen values(Hartree) of pseudo atomic orbitals

O7.0.pao

Eigenvalues
Lmax= 3 Mul=15

mu	0 0	-0.87913976280231	↖1
mu	0 1	0.06809061901229	
mu	0 2	0.52709275865941	↖4
mu	0 3	1.24995140722317	
mu	0 4	2.21829552402723	
mu	0 5	3.41945468859267	
mu	0 6	4.84509607059749	
mu	0 7	6.48825090142865	
mu	0 8	8.34207134316001	
mu	0 9	10.39973244132192	
mu	0 10	12.65513764955926	
mu	0 11	15.10428688136984	
mu	0 12	17.74677530362947	
mu	0 13	20.58633940582683	
mu	0 14	23.62957122674031	↖2
mu	1 0	-0.33075182895384	
mu	1 1	0.16376499567753	
mu	1 2	0.64129274864838	
mu	1 3	1.35995471521821	↖7
mu	1 4	2.31377697411480	
mu	1 5	3.50052618379026	
mu	1 6	4.91841590421346	
mu	1 7	6.56499793348396	
mu	1 8	8.43634105410091	
mu	1 9	10.52733540479795	
mu	1 10	12.83276989780839	
mu	1 11	15.34880171573570	
mu	1 12	18.07414632417004	
mu	1 13	21.01013110934429	
mu	1 14	24.15933607166566	↖8
mu	2 0	0.26162948257116	
mu	2 1	0.70705247436937	
mu	2 2	1.34714706672243	
mu	2 3	2.19799459356269	
mu	2 4	3.26511989658328	

Choosing states with lower eigenvalues leads to H7.0-s2p1 and O7.0-s2p2d1.

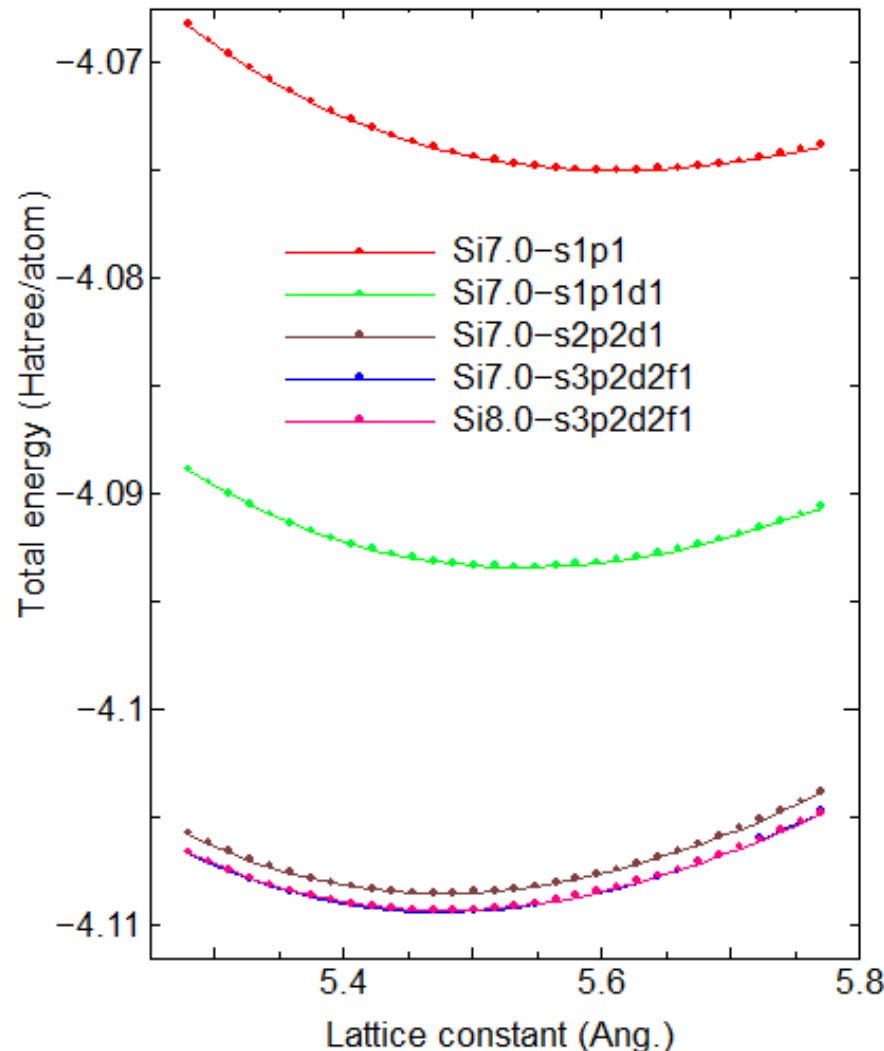
How to choose basis functions: Si case(1)

Eigenvalues
Lmax= 3 Mul=15

Si7.0.pao

mu	0	0	-0.59320968623145	1
mu	0	1	0.01654247265872	4
mu	0	2	0.57915688461708	8
mu	0	3	1.39915528592998	
mu	0	4	2.43106703909053	
mu	0	5	3.63498884739274	
mu	0	6	4.99885012002783	
mu	0	7	6.55485759476897	
mu	0	8	8.33541526908920	
mu	0	9	10.34055461939939	
mu	0	10	12.55818051232040	
mu	0	11	14.98369777887096	
mu	0	12	17.62014479571136	
mu	0	13	20.47011746372617	
mu	0	14	23.53245309073866	
mu	1	0	-0.31460013133101	2
mu	1	1	0.12937640798489	5
mu	1	2	0.71637380083701	
mu	1	3	1.54488697995006	
mu	1	4	2.59211686526084	
mu	1	5	3.84687324239366	
mu	1	6	5.31246180826158	
mu	1	7	6.99509799702661	
mu	1	8	8.89454075291723	
mu	1	9	11.00602102880752	
mu	1	10	13.32692005826443	
mu	1	11	15.85857455673626	
mu	1	12	18.60272353142246	
mu	1	13	21.55894843635971	
mu	1	14	24.72602323954447	
mu	2	0	0.01886411574821	3
mu	2	1	0.35893514996065	7
mu	2	2	0.94629918754692	
mu	2	3	1.76201765644983	
mu	2	4	2.81418723624388	
mu	2	5	4.10656012645961	
mu	2	6	5.63661971875011	
mu	2	7	7.39522693820483	
mu	2	8	9.37222098331867	
mu	2	9	11.56227937764802	
mu	2	10	13.96620568880690	
mu	2	11	16.58651812641581	
mu	2	12	19.42296574188659	
mu	2	13	22.47308081363278	
mu	2	14	25.73538272482773	
mu	3	0	0.28356769151846	6
mu	3	1	0.79082836114569	
mu	3	2	1.52992065308726	
mu	3	3	2.52537261496132	

Orbitals with lower eigenvalues in Si7.0.pao are taken into account step by step as the quality of basis set is improved.

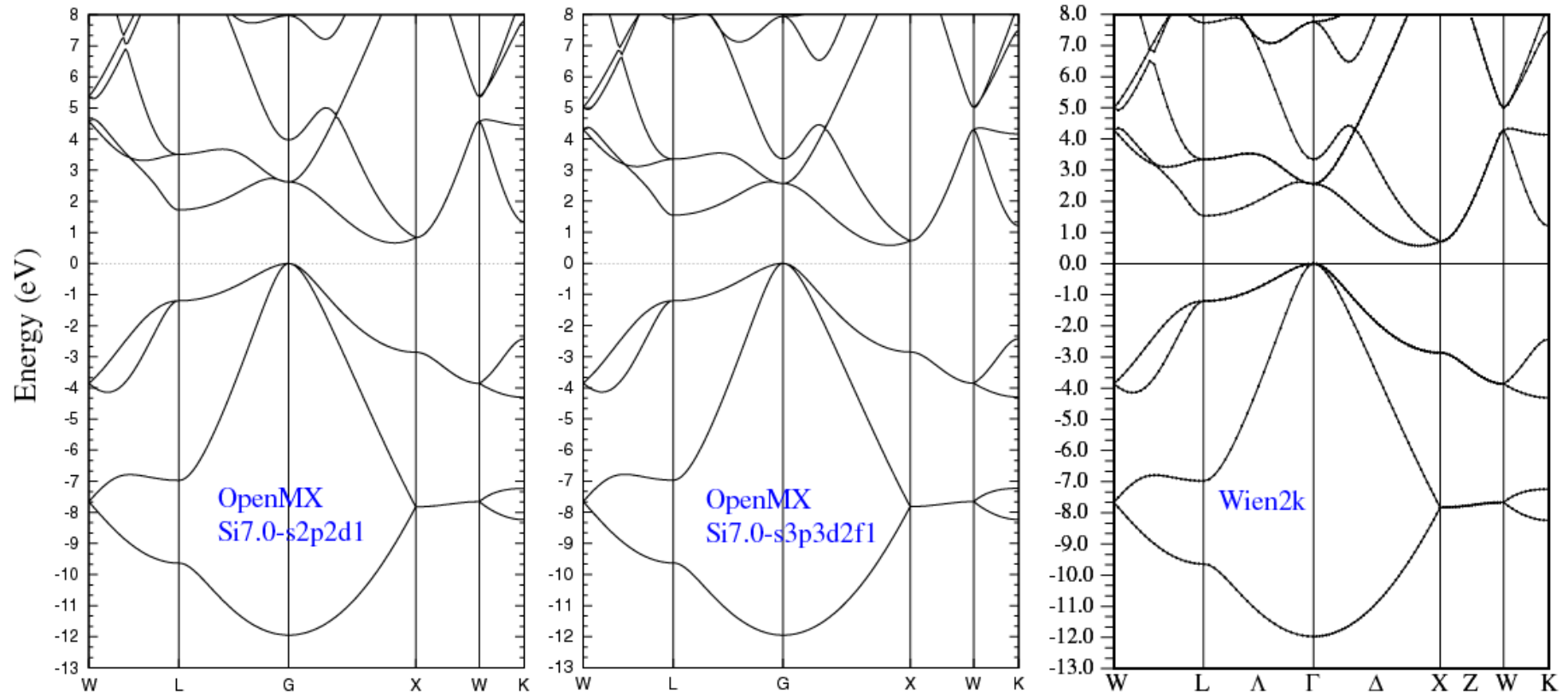


Si7.0-s2p2d1 is enough to discuss structural properties.

By comparing Si7.0-s3p2d2f1 with Si8.0-s3p2d2f1, it turns out that convergence is achieved at the cutoff of 7.0(a.u.).

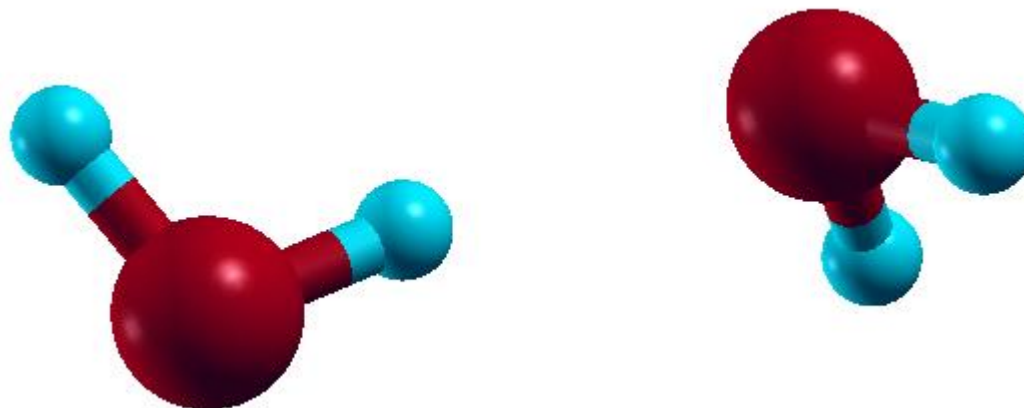
How to choose basis functions: Si case(2)

With respect to band structure, one can confirm that Si7.0-s2p2d1 provides a nearly convent result.



While the convergent result is achieved by use of Si7.0-s3p2d2f1(Si7.0-s3p3d2f1), Si7.0-s2p2d1 is a balanced basis functions compromising accuracy and efficiency to perform a vast range of materials exploration.

Basis set superposition Error (BSSE)



	Equilibrium O-O distance (Ang.)	Dipole moment (Debye)	Binding energy (kcal/mol)	Binding energy (counterpoise corrected) (kcal/mol)
O7.0-s2p2d1, H7.0-s2p1	2.899	2.54	5.57	5.21
O7.0-s3p3d2, H7.0-s3p2	2.897	2.45	5.48	5.48
Other calc.	2.893 ^a		5.15 ^a	
Expt.	2.98 ^b	2.60 ^b	5.44 ^b	

https://www.openmx-square.org/vps_pao2019/O/index.html

A series of benchmark calculations implies that BSSE is
 ~ 0.5 kcal/mol for molecular systems.

Recommendation of choices for PAO

VPS	Valence electrons	Quick	Standard	Precise
E	0.0	Kr10.0-s1p1	Kr10.0-s2p1d1	Kr10.0-s2p2d1f1
H_PBE19	1.0	H5.0-s2	H6.0-s2p1	H7.0-s2p2d1
He_PBE19	2.0	He8.0-s1p1	He8.0-s2p1	He10.0-s2p2d1
Li_PBE19	3.0	Li8.0-s3p1	Li8.0-s3p2	Li8.0-s3p2d1
Be_PBE19	2.0	Be7.0-s2p1	Be7.0-s2p2	Be7.0-s3p2d1
B_PBE19	3.0	B7.0-s2p2	B7.0-s2p2d1	B7.0-s3p2d2
C_PBE19	4.0	C6.0-s2p2	C6.0-s2p2d1	C6.0-s3p2d2
N_PBE19	5.0	N6.0-s2p2	N6.0-s2p2d1	N6.0-s3p2d2
O_PBE19	6.0	O6.0-s2p2	O6.0-s2p2d1	O6.0-s3p2d2
F_PBE19	7.0	F6.0-s2p2	F6.0-s2p2d1	F6.0-s3p3d2f1
Ne_PBE19	8.0	Ne9.0-s2p2	Ne9.0-s2p2d1	Ne9.0-s3p2d2
Na_PBE19	9.0	Na9.0-s3p2	Na9.0-s3p2d1	Na9.0-s3p2d2
Mg_PBE19	8.0	Mg9.0-s2p2	Mg9.0-s3p2d1	Mg9.0-s3p2d2
Al_PBE19	3.0	Al7.0-s2p1d1	Al7.0-s2p2d1	Al7.0-s3p2d2
Si_PBE19	4.0	Si7.0-s2p1d1	Si7.0-s2p2d1	Si7.0-s3p3d2
P_PBE19	5.0	P7.0-s2p2d1	P7.0-s2p2d1f1	P7.0-s3p2d2f1
S_PBE19	6.0	S7.0-s2p2d1	S7.0-s2p2d1f1	S7.0-s3p2d2f1
Cl_PBE19	7.0	Cl7.0-s2p2d1	Cl7.0-s2p2d1f1	Cl7.0-s3p2d2f1

See https://www.openmx-square.org/openmx_man3.9/node27.html

Structure of a molecule

```

Atoms.Number      5
Atoms.SpeciesAndCoordinates.Unit  Ang # Ang|AU|FRAC
<Atoms.SpeciesAndCoordinates
1 C   0.000000  0.000000  0.000000  2.0 2.0
2 H  -0.889981 -0.629312  0.000000  0.5 0.5
3 H   0.000000  0.629312 -0.889981  0.5 0.5
4 H   0.000000  0.629312  0.889981  0.5 0.5
5 H   0.889981 -0.629312  0.000000  0.5 0.5
Atoms.SpeciesAndCoordinates>
  
```

of atoms

Unit of atomic coordinate

Initial occupation for up and down states of each atom.
See also the manual;
https://www.openmx-square.org/openmx_man3.9/node27.html

```

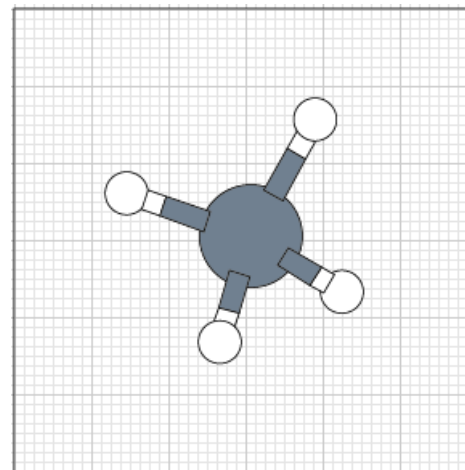
Atoms.UnitVectors.Unit  Ang # Ang|AU
<Atoms.UnitVectors
10.0  0.0  0.0
 0.0 10.0  0.0
 0.0  0.0 10.0
Atoms.UnitVectors>
  
```

Unit of unit vectors

Unit vectors

a1
a2
a3

Atomic coordinates



Structure of a bulk

```
Atoms.Number 2
Atoms.SpeciesAndCoordinates.Unit FRAC
<Atoms.SpeciesAndCoordinates
  1 Si 0.7500000 0.7500000 0.7500000 2.0 2.0
  2 Si 0.5000000 0.5000000 0.5000000 2.0 2.0
Atoms.SpeciesAndCoordinates>
```

of atoms

Unit of atomic coordinate

Initial occupation for up and down states of each atom. See also the manual;

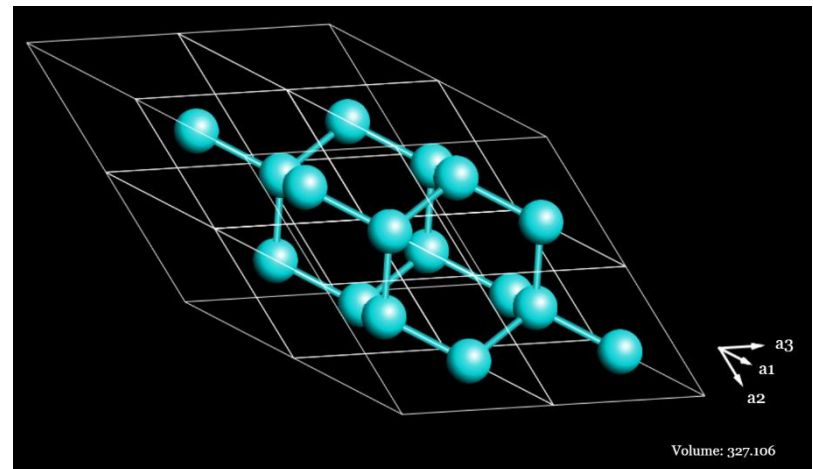
Atomic coordinates

https://www.openmx-square.org/openmx_man3.9/node27.html

```
Atoms.UnitVectors.Unit Ang
<Atoms.UnitVectors
  3.8669746 0.0000000 0.0000000
  1.9334873 3.3488983 0.0000000
  1.9334873 1.1162994 3.1573716
Atoms.UnitVectors>
```

Unit of unit vectors

Unit vectors
a1
a2
a3



Choice of cutoff energy

`scf.energycutoff`

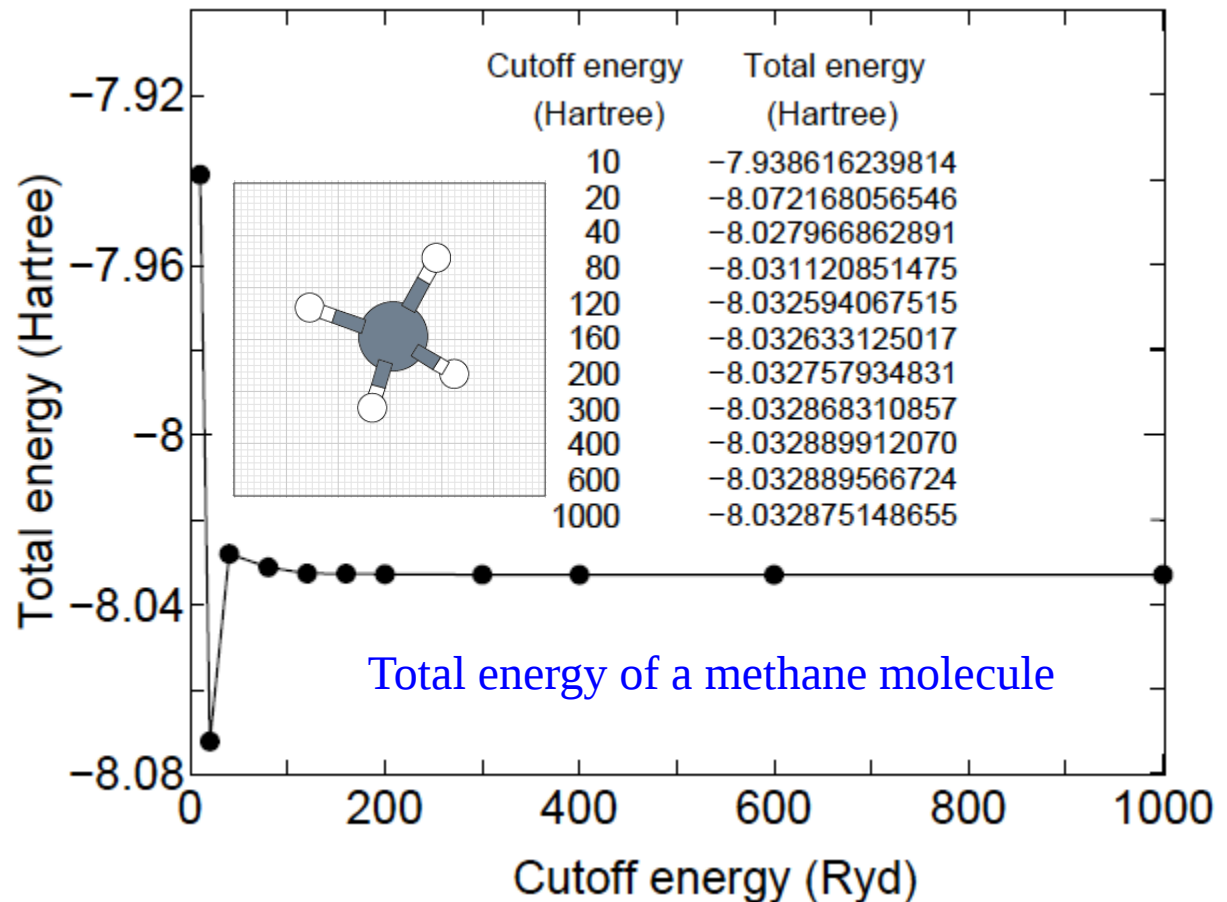
200 # default=150 Ryd

The FFT grid is used to discretize real space and calculate $E_{\delta ee}$, E_{xc} , and can be specified by `scf.energycutoff`.

In most cases, 200 Ryd is enough to get convergence.

However, large cutoff energy (300-400 Ryd) has be used cases such as use of pseudopotentials with deep semi-core states.

Memory requiment
 $O(E^{3/2})$



Choice of cutoff energy

Geometry optimization of H₂O

Dependency of optimized structure of H₂O on scf.energycutoff. It turns out that 180Ryd. is enough to reach the convergence.

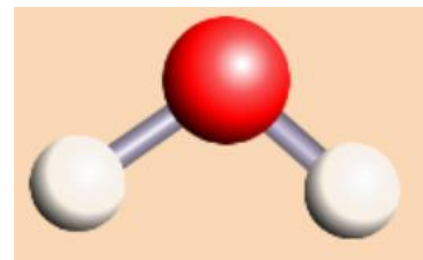


Table 1: Convergence of structural parameters, dipole moment of a water molecule with respect to the cutoff energy. The input file is 'H2O.dat' in the directory 'work'.

Ecut(Ryd)	r(H-O) (Å)	∠ (H-O-H) (deg)	Dipole moment (Debye)
60	0.970	103.4	1.838
90	0.971	103.7	1.829
120	0.971	103.7	1.832
150	0.971	103.6	1.829
180	0.971	103.6	1.833
Exp.	0.957	104.5	1.85

Volume vs. Energy curves

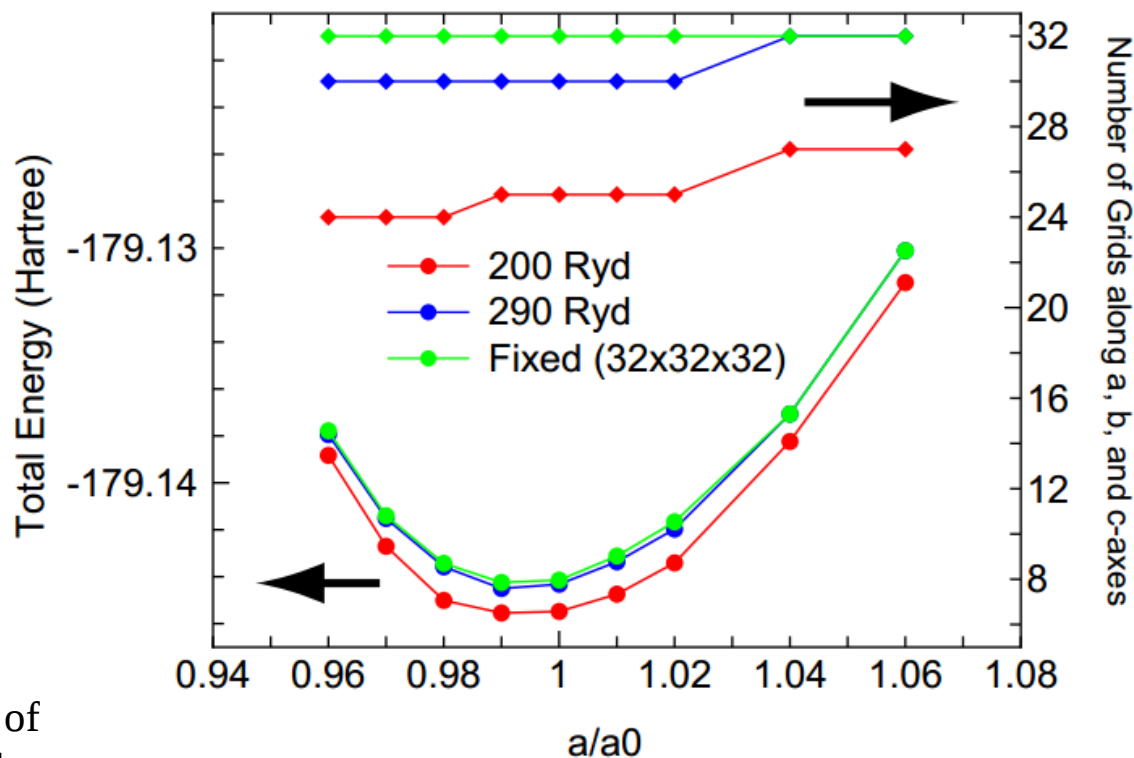
The following keywords are available to calculate energy curves.

MD.Type	EvsLC	#
MD.EvsLC.Step	0.4	# default=0.4%
MD.maxIter	32	# default=1

When the energy curve for bulk system is calculated as a function of the lattice parameter, a sudden change of the number of real space grids is a serious problem which produces an erratic discontinuity on the energy curve. To avoid this, the number of grids should be fixed by explicitly specifying the following keyword:

scf.Ngrid 32 32 32

The numbers correspond to the number of grid along a-, b-, and c-axes, respectively. scf.Ngrid is used if both the keywords scf.energycutoff and scf.Ngrid are specified.



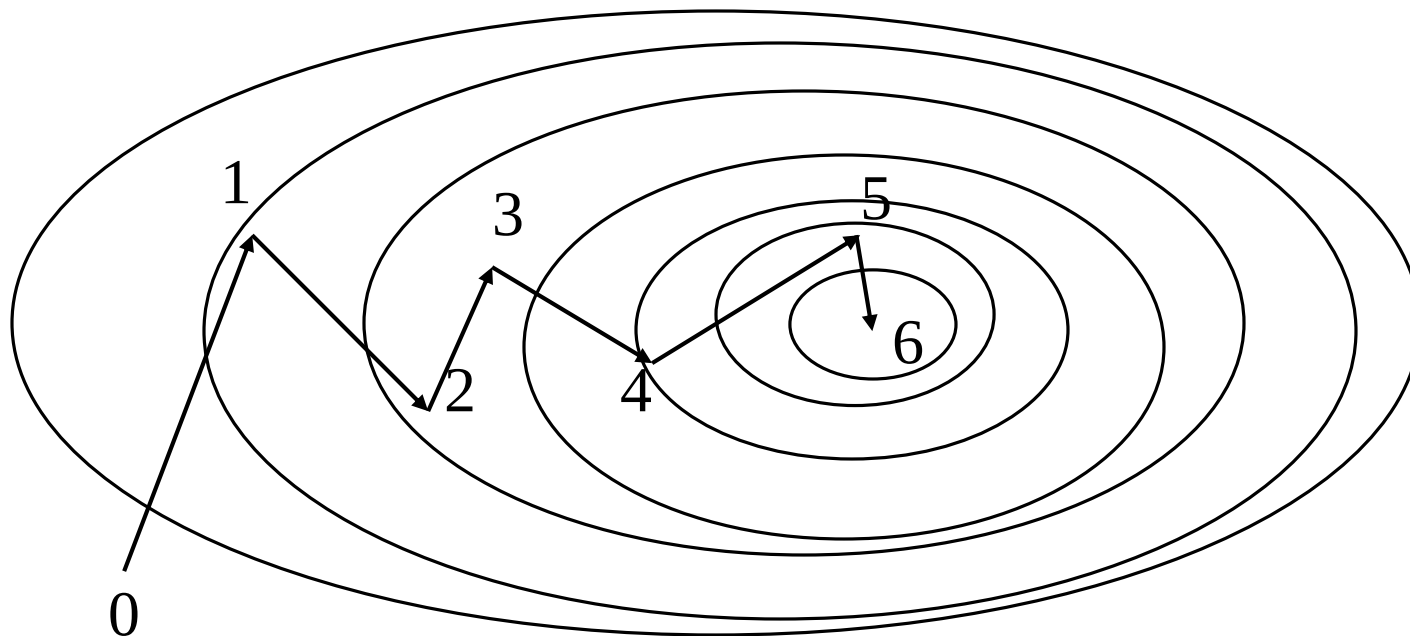
MD.EvsLC.flag 1 1 0 # along a, b, c-axes
1: varied, 0: fixed

Geometry optimization

To quantitatively investigate structural, physical, and chemical properties of molecules and solids, it would be important to obtain optimized structures.

Steepest decent (SD) method

The structure is changed along the steepest decent step by step.



It may not be so efficient.

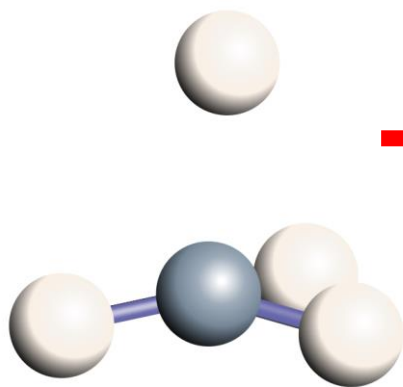
Geometry optimization by the SD method

Let us change the x-coordinate of carbon atom in a methane molecule to 0.3 Å as

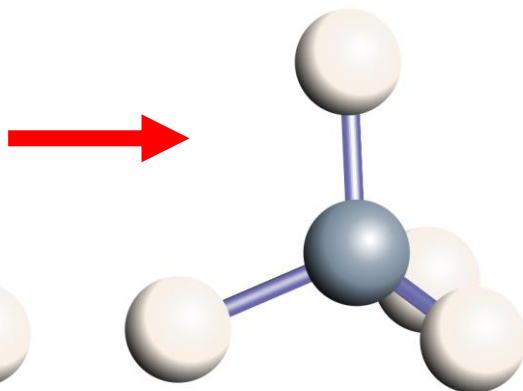
```
<Atoms.SpeciesAndCoordinates
 1  C   0.300000  0.000000  0.000000  2.0  2.0
 2  H   -0.889981 -0.629312  0.000000  0.5  0.5
 3  H    0.000000  0.629312 -0.889981  0.5  0.5
 4  H    0.000000  0.629312  0.889981  0.5  0.5
 5  H    0.889981 -0.629312  0.000000  0.5  0.5
Atoms.SpeciesAndCoordinates>
```

Using “Methane2.dat” in the directory
“work”, you can trace the calculation.

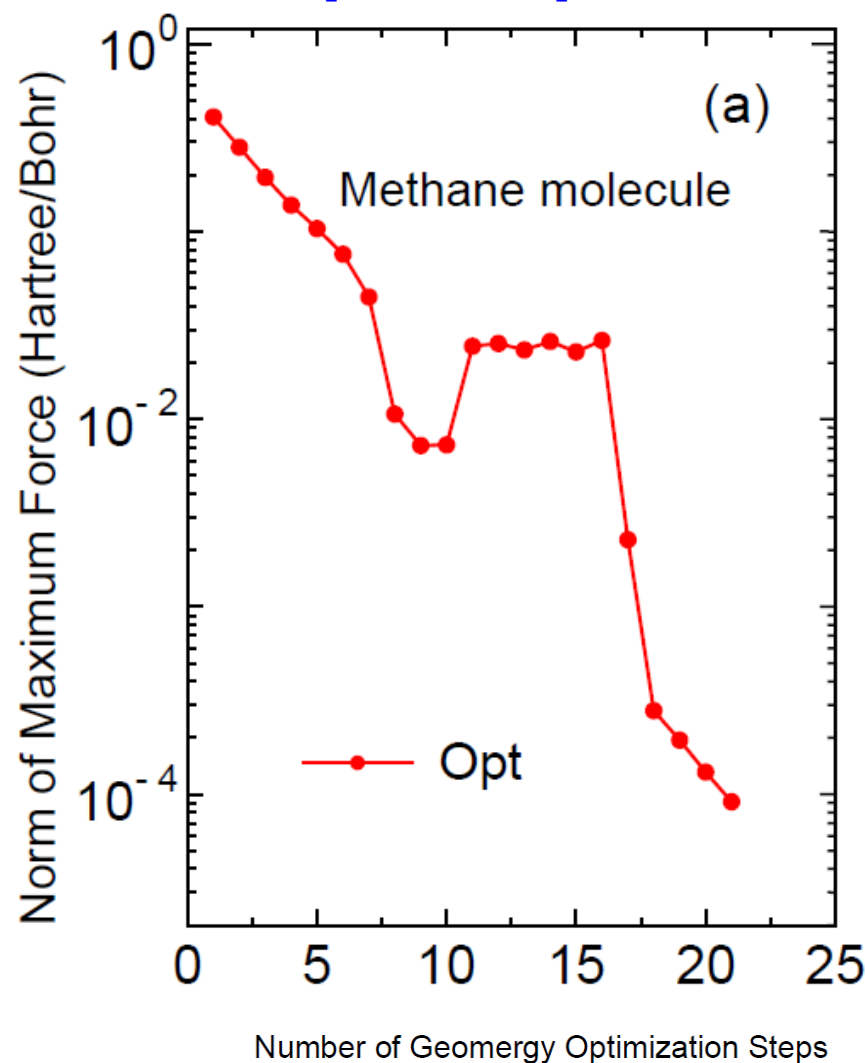
Initial structure



Final structure



Optimization process



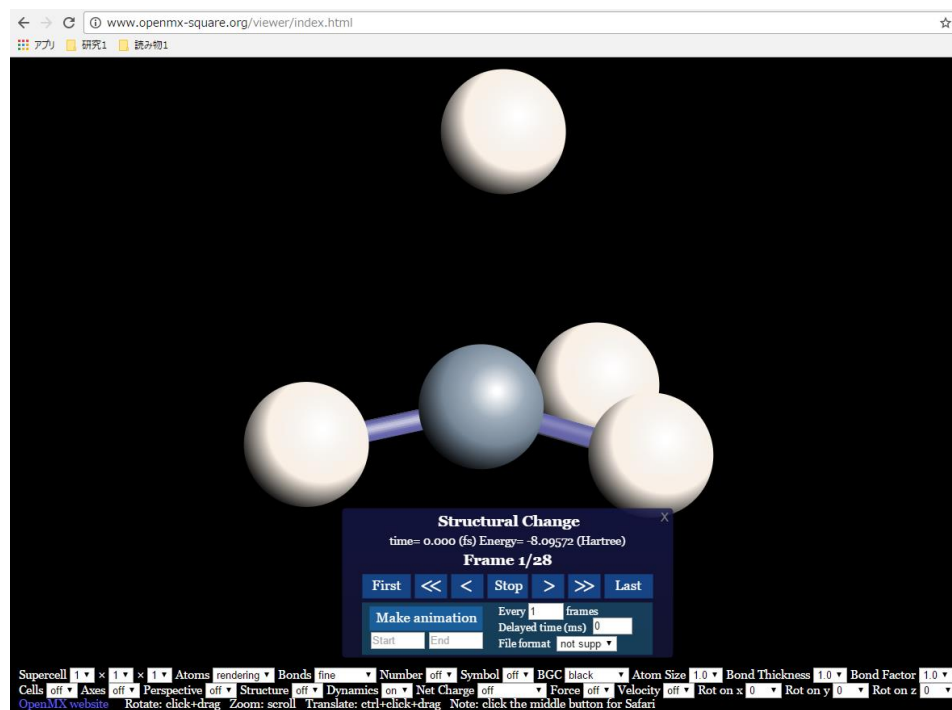
Relevant output files for the geometry optimization

In [met2.out](#), the history of optimization can be confirmed.

```
*****
*****
***** History of geometry optimization *****
*****
```

MD_iter	SD_scaling	Maximum force (Hartree/Bohr)	Maximum step (Ang)	Utot (Hartree)
1	1.25981733	0.40873710	0.10583545	-8.09571722
2	1.25981733	0.12000148	0.08000099	-8.16523842
3	1.25981733	0.06115237	0.04076824	-8.18400222
4	1.25981733	0.01811189	0.01207460	-8.18831175
5	3.14954331	0.01076505	0.01794175	-8.18898468
6	3.14954331	0.00576811	0.00961351	-8.18980390
7	3.14954331	0.00355034	0.00591724	-8.19010991
8	3.14954331	0.00255760	0.00426266	-8.19022887
9	7.87385828	0.00218904	0.00912099	-8.19027446
10	7.87385828	0.00863395	0.03597478	-8.19023586
11	1.57477166	0.04692594	0.03910495	-8.18816953
12	1.57477166	0.01357319	0.01131099	-8.19012703
13	3.93692914	0.00406864	0.00847634	-8.19029047
14	3.93692914	0.00915707	0.01907723	-8.19022710
15	0.78738583	0.02067017	0.00861257	-8.18990108
16	0.78738583	0.00721609	0.00300671	-8.19025699
17	1.96846457	0.00255359	0.00265999	-8.19030094
18	1.96846457	0.00163789	0.00170613	-8.19030460
19	1.96846457	0.00105501	0.00109897	-8.19030613
20	1.96846457	0.00068667	0.00071528	-8.19030671
21	4.92116143	0.00044434	0.00115713	-8.19030697
22	4.92116143	0.00138402	0.00360421	-8.19030537
23	0.98423229	0.00427346	0.00222576	-8.19028994
24	0.98423229	0.00078570	0.00040922	-8.19030656
25	2.46058071	0.00013628	0.00017745	-8.19030709
26	2.46058071	0.00014411	0.00018765	-8.19030714
27	0.49211614	0.00015118	0.00003937	-8.19030709
28	0.49211614	0.00008899	0.00002317	-8.19030713

By dragging and dropping [met2.md](#) to [OpenMX Viewer](#), the optimization process can be easily visualized.



Geometry optimization by Newton-type methods

By Taylor-expanding, we have

$$E = E_0 + \sum_i^{3N} \left(\frac{\partial E}{\partial x_i} \right)_0 (x_i - x_i^{(0)}) + \frac{1}{2} \sum_{i,j}^{3N} \left(\frac{\partial^2 E}{\partial x_i \partial x_j} \right)_0 (x_i - x_i^{(0)}) (x_j - x_j^{(0)}) + \dots,$$

The derivative of the total energy w.r.t. coordinates leads to

$$\frac{\partial E}{\partial x_k} = \left(\frac{\partial E}{\partial x_k} \right)_0 + \sum_i^{3N} \left(\frac{\partial^2 E}{\partial x_k \partial x_i} \right)_0 (x_i - x_i^{(0)}).$$

$$\frac{\partial E}{\partial x_k} = 0 \quad \longrightarrow \quad \begin{pmatrix} \left(\frac{\partial^2 E}{\partial x_1 \partial x_1} \right)_0 & \left(\frac{\partial^2 E}{\partial x_1 \partial x_2} \right)_0 & \cdots \\ \left(\frac{\partial^2 E}{\partial x_2 \partial x_1} \right)_0 & \left(\frac{\partial^2 E}{\partial x_2 \partial x_2} \right)_0 & \cdots \\ \cdots & \cdots & \cdots \end{pmatrix} \begin{pmatrix} (x_1 - x_1^{(0)}) \\ (x_2 - x_2^{(0)}) \\ \cdots \end{pmatrix} = - \begin{pmatrix} \left(\frac{\partial E}{\partial x_1} \right)_0 \\ \left(\frac{\partial E}{\partial x_2} \right)_0 \\ \cdots \end{pmatrix}.$$

$$H \Delta \mathbf{x} = -\mathbf{g},$$

By solving the Eq. for $\Delta \mathbf{x}$, we have

$$\mathbf{x}^{(n+1)} = \mathbf{x}^{(n)} - (H^{(n)})^{-1} \mathbf{g}^{(n)}.$$

If the Hessian matrix H can be computed, the method is efficient.
However, it may be difficult to evaluate H in general.

Geometry optimization by Newton-type methods

The geometry optimization in OpenMX is based on quasi Newton methods. In Ver. 3.9, the following four methods are available.

$$\mathbf{r}_{\text{new}} = \mathbf{r}_{\text{DIIS}} + \Delta\mathbf{r}$$

$$\Delta\mathbf{r} = -H^{-1}\mathbf{g}_{\text{DIIS}}$$

Methods of calculating approximate Hessian matrix H

DIIS

BFGS

RF(rational function)

EF(eigenvector following)

H=I

BFGS

BFGS+RF

BFGS plus monitoring
of eigenvalues of H

Broyden-Fletcher-Goldfarb-Shanno (BFGS) method

$$H_k = H_{k-1} + \frac{|\Delta\mathbf{g}_k\rangle\langle\Delta\mathbf{g}_k|}{\langle\Delta\mathbf{g}_k|\Delta\mathbf{r}_k\rangle} - \frac{|H_k\Delta\mathbf{r}_k\rangle\langle\Delta\mathbf{r}_kH_k|}{\langle\Delta\mathbf{r}_k|H_k|\Delta\mathbf{r}_k\rangle}$$

If the inner product in the red box is positive, the positive definiteness of H is guaranteed.

Approximate initial Hessian by Schlegel

Schlegel proposed a way of constructing an approximate Hessian. A force constant for every pair of elements is fitted to the following formula, where dataset were constructed by B3LYP calculations.

$$F = \frac{A}{(r - B)^3}$$

H.B. Schlegel, Theoret. Chim. Acta (Berl.) 66, 333 (1984); J.M. Wittbrodt and H.B. Schlegel, J. Mol. Struct. (Theochem) 398-399, 55 (1997).

Parameter B for Badger's rule computed at the B3LYP level of theory

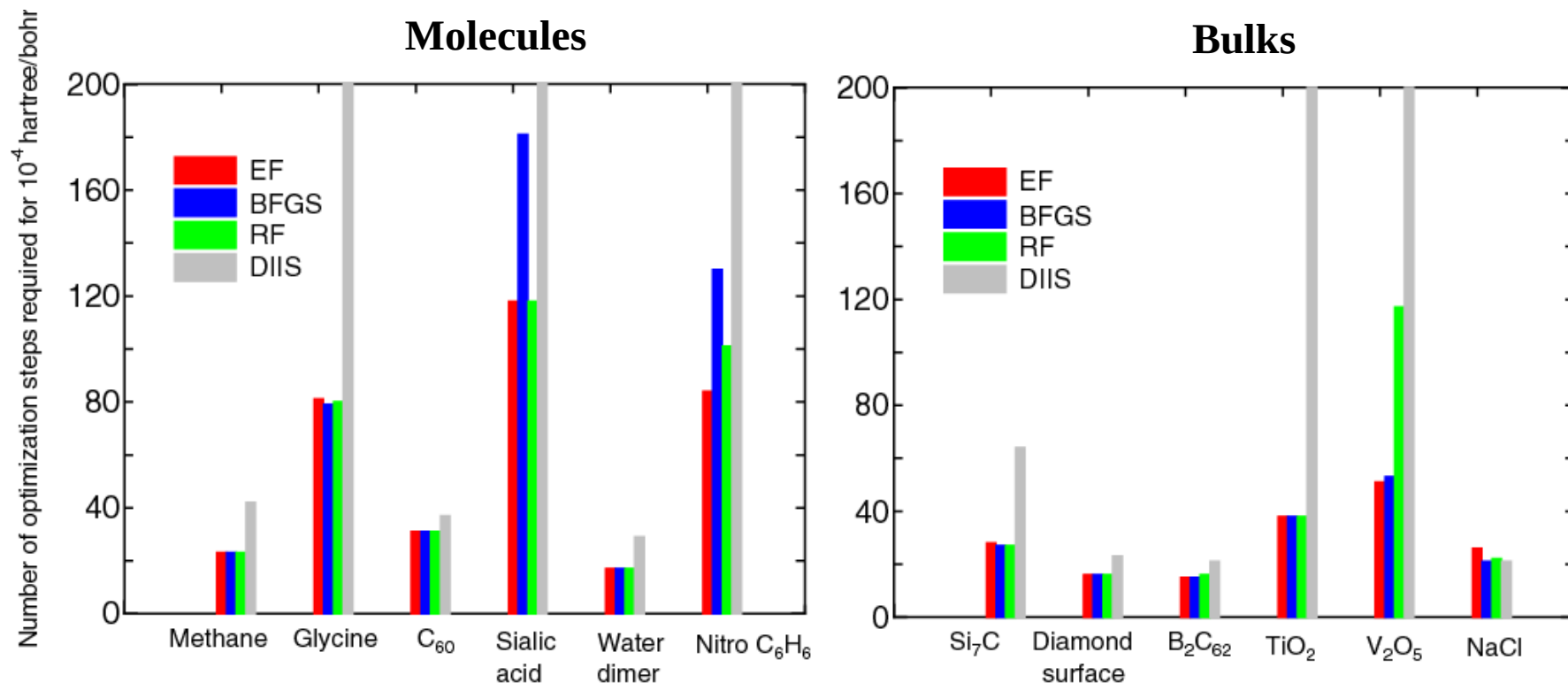
Period	1 H	2 Li-F	3 Na-Cl	4 K-Br	5 Rb-I	6 Cs-At
1	- 0.2573	0.3401	0.6937	0.7126	0.8335	0.9491
2		0.9652	1.2843	1.4725	1.6549	1.7190
3			1.6925	1.8238	2.1164	2.3185
4				2.0203	2.2137	2.5206
5					2.3718	2.5110

Suppose the total energy is given by the sum of pairwise potentials. Then, the derivatives lead to the following relation:

$$V_2 = \frac{1}{2} \sum_i \left(\sum_{Rn} \sum_j f(|r_i + R_n - r_j|) \right) \quad H = BF$$

where B is the B-matrix of Wilson, H is the approximate Hessian in Cartesian coordinate.

Comparison of four methods



- The benchmark calculations imply that the EF and RF work well.
- Large molecules with structural large freedom are hard to get convergence.

The input file and output files for the benchmark calculations are available in `openmx3.9/work/geoopt_example`.

Keywords relevant to geometry optimization

MD.Type	EF	# Opt DIIS BFGS RF EF
MD.Opt.DIIS.History	3	# default=3
MD.Opt.StartDIIS	5	# default=5
MD.Opt.EveryDIIS	200	# default=200
MD.maxIter	100	# default=1
MD.Opt.criterion	1.0e-4	# default=0.0003 (Hartree/Bohr)

The behavior of the quasi Newton methods can be controlled by the following two keywords:

MD.Opt.DIIS.History	3 # default=3
MD.Opt.StartDIIS	5 # default=5

The keyword 'MD.Opt.DIIS.History' gives the number of previous steps to estimate the optimized structure used in the geometry optimization by 'DIIS', 'EF', and 'RF'. The default value is 3.

The geometry optimization step at which 'DIIS', 'EF', or 'RF' starts is specified by the keyword 'MD.Opt.StartDIIS'. The geometry optimization steps before starting the DIIS-type method is performed by the steepest decent method. The default value is 5.

https://www.openmx-square.org/openmx_man3.9/node47.html

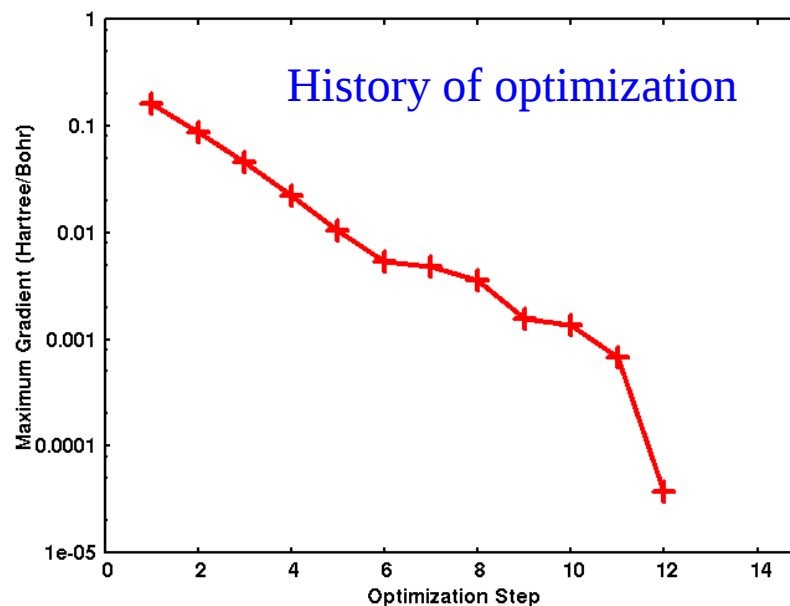
Variable cell optimization

Let us start optimization of diamond lattice with a displacement as

```
Atoms.Number      2
Atoms.SpeciesAndCoordinates.Unit  frac # Ang|AU
<Atoms.SpeciesAndCoordinates
  1  C  0.10000000000000  0.00000000000000  -0.05000000000000  2.0  2.0
  2  C  0.25000000000000  0.25000000000000  0.25000000000000  2.0  2.0
Atoms.SpeciesAndCoordinates>
Atoms.UnitVectors.Unit      Ang # Ang|AU
<Atoms.UnitVectors
  1.6400  1.6400  0.0000
  1.6400  0.0000  1.6400
  0.0000  1.6400  1.6400
Atoms.UnitVectors>
```

Relevant keywords:

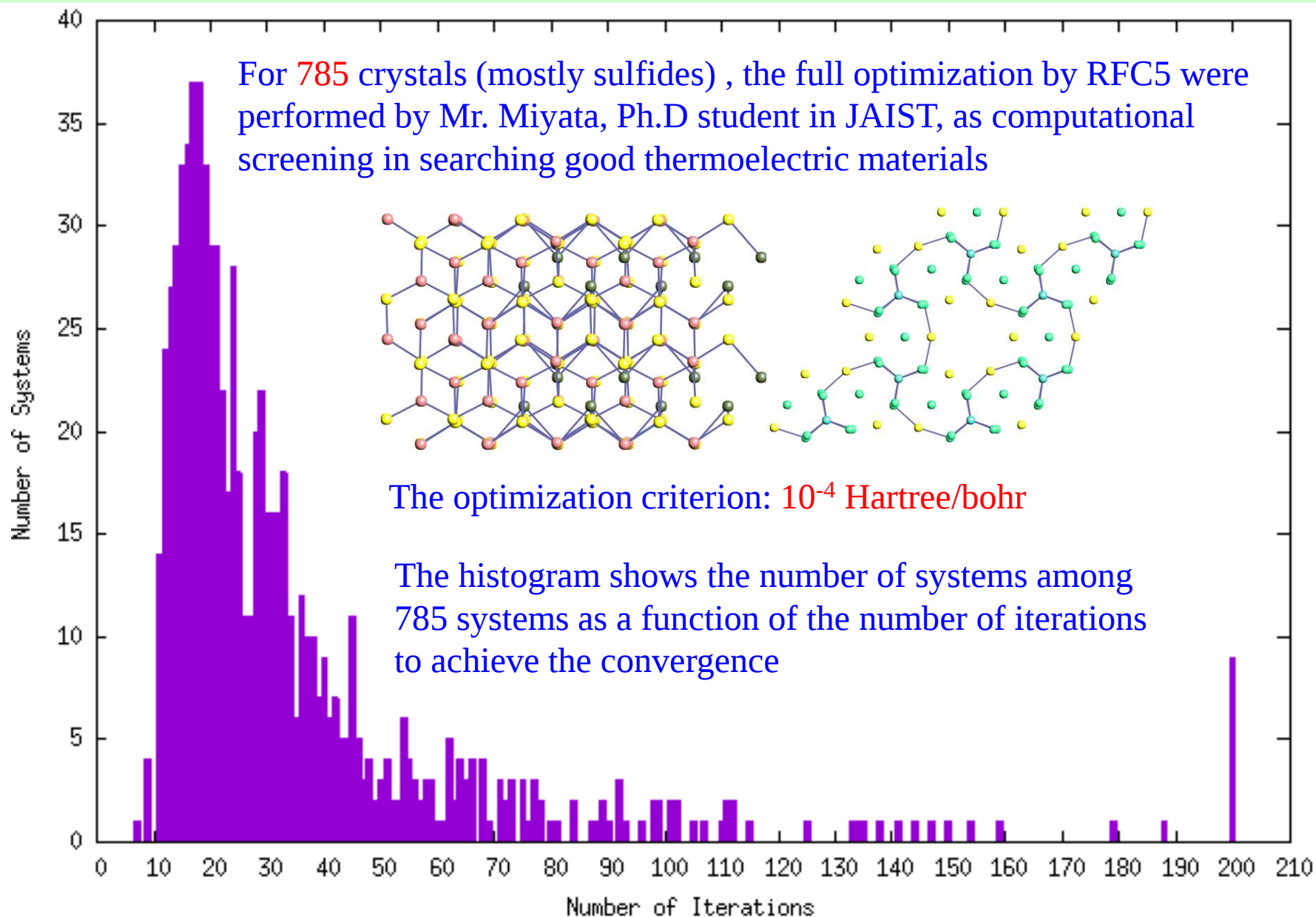
```
MD.Type      RFC5
MD.Opt.DIIS.History  3
MD.Opt.StartDIIS    7
MD.Opt.EveryDIIS    100000
MD.maxIter      100
MD.TimeStep     1.0
MD.Opt.criterion  0.0003
```



The calculation can be traced by “Cdia-RF5.dat” in work/cellopt_example.

Please see also the page 74 in the manual.

Benchmark calculations of RFC5



Restarting of calculations

- After finishing your first calculation or achieving the self consistency, you may want to continue the calculation or to calculate density of states, band dispersion, molecular orbitals, and etc. using the self consistent charge in order to save the computational time. To do this, a keyword 'scf.restart' is available.

`scf.restart` `on` `# on|off,default=off`

- If the first trial for geometry optimization does not reach a convergent result or molecular dynamics simulation is terminated due to a wall time, one can restart the geometry optimization using an input file 'System.Name.dat#' which is generated at every step for the restart calculation with the final structure.

See also the manual at

https://www.openmx-square.org/openmx_man3.9/node44.html

Output of large-sized files in binary mode

Large-scale calculations produce large-sized files in text mode such as cube files. The IO access to output such files can be very time consuming in machines of which IO access is not fast. In such a case, it is better to output those large-sized files in binary mode. The procedure is supported by the following keyword:

`OutData.bin.flag on # default=off, on|off`

Then, all large-sized files will be output in binary mode. The default is 'off'. The output binary files are converted using a small code 'bin2txt.c' stored in the directory 'source' which can be compiled as

`gcc bin2txt.c -lm -o bin2txt`

As a post processing, you will be able to convert as

`./bin2txt *.bin`

The functionality will be useful for machines of which IO access is not fast.

See also the manual: https://www.openmx-square.org/openmx_man3.9/node218.html