Localized basis methods Theory and implementations

- Introduction of OpenMX
- Implementation of OpenMX
 - ➢ Total energy
 - Pseudopontials
 - Basis functions
- Δ -gauge
- Practical guide to OpenMX calc.

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OpenMX Open source package for Material eXplorer

- Software package for density functional calculations of molecules and bulks
- Norm-conserving pseudopotentials (PPs)
- Variationally optimized numerical atomic basis functions

Basic functionalities

- SCF calc. by LDA, GGA, DFT+U
- Total energy and forces on atoms
- Band dispersion and density of states
- Geometry optimization by BFGS, RF, EF
- Charge analysis by Mullken, Voronoi, ESP
- Molecular dynamics with NEV and NVT ensembles
- Charge doping
- Fermi surface
- Analysis of charge, spin, potentials by cube files
- Database of optimized PPs and basis funcitons

Extensions

- O(N) and low-order scaling diagonalization
- Non-collinear DFT for non-collinear magnetism
- Spin-orbit coupling included self-consistently
- Electronic transport by non-equilibrium Green function
- Electronic polarization by the Berry phase formalism
- Maximally localized Wannier functions
- Effective screening medium method for biased system
- Reaction path search by the NEB method
- Band unfolding method
- STM image by the Tersoff-Hamann method
- etc.

History of OpenMX

- 2000 Start of development
- 2003 Public release (GNU-GPL)
- 2003 Collaboration: AIST, NIMS, SNU KAIST, JAIST, Kanazawa Univ. CAS, UAM NISSAN, Fujitsu Labs. etc.

2018 18 public releases Latest version: 3.8

About 500 papers published using OpenMX

Welcome to OpenMX

Contents

What's new

OpenMX Hands-On Workshop on 10th Oct. in Kobe Patch (Ver.3.7.8) to OpenMX Ver. 3.7 (17/Feb/2014) Release of OpenFFT Ver. 1.0 (04/Sep/2013)

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 Database of VI
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- <u>Acknowledgment</u>
- Links

http://www.openmx-square.org



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Materials studied by OpenMX

First characterization of silicene on ZrB₂ in collaboration with experimental groups

A. Fleurence et al., Phys. Rev. Lett. 108, 245501 (2012).

First identification of Jeff=1/2 Mott state of Ir oxides

B.J. Kim et al., Phys. Rev. Lett. 101, 076402 (2008).

Theoretical proposal of topological insulators

C.-H. Kim et al., Phys. Rev. Lett. 108, 106401 (2012). H. Weng et al., Phy. Rev. X 4, 011002 (2014).

First-principles molecular dynamics simulations for Li ion battery

T. Ohwaki et al., J. Chem. Phys. 136, 134101 (2012).T. Ohwaki et al., J. Chem. Phys. 140, 244105 (2014).

Magnetic anisotropy energy of magnets

Z. Torbatian et al., Appl. Phys. Lett. 104, 242403 (2014).I. Kitagawa et al., Phys. Rev. B 81, 214408 (2010).

Electronic transport of graphene nanoribbon on surface oxidized Si

H. Jippo et al., Appl. Phys. Express 7, 025101 (2014).M. Ohfuchi et al., Appl. Phys. Express 4, 095101 (2011).

Interface structures of carbide precipitate in bcc-Fe

H. Sawada et al., Modelling Simul. Mater. Sci. Eng. 21, 045012 (2013).

Universality of medium range ordered structure in amorphous metal oxides

K. Nishio et al., Phys. Rev. Lett. 340, 155502 (2013).

Materials treated so far

Silicene, graphene Carbon nanotubes Transition metal oxides Topological insulators Intermetallic compounds Molecular magnets Rare earth magnets Lithium ion related materials Structural materials etc.

About 500 published papers

Implementation of OpenMX

- Density functional theory
- Mathematical structure of KS eq.
- LCPAO method
- Total energy
- Pseudopotentials
- Basis functions

Density functional theory

The energy of non-degenerate ground state can be expressed by a functional of electron density. (Hohenberg and Kohn, 1964)

$$E[\rho] = T[\rho] + J[\rho] + \int \rho v_{\text{ext}} d\mathbf{r} + E_{\text{xc}}^{(0)}[\rho]$$

The many body problem of the ground state can be reduced to an one-particle problem with an effective potential. (Kohn-Sham, 1965)

 $\hat{H}_{\rm KS}\phi_i = \varepsilon_i\phi_i$



W.Kohn (1923-2016)

$$\hat{H}_{\rm KS} = -\frac{1}{2}\nabla^2 + v_{\rm eff}$$

$$v_{\rm eff}(\mathbf{r}) = v_{\rm ext}(\mathbf{r}) + v_{\rm Hartree}(\mathbf{r}) + \frac{\delta E_{xc}}{\delta\rho(\mathbf{r})}$$

Mathematical structure of KS eq.

3D coupled non-linear differential equations have to be solved self-consistently.



Input charge = Output charge \rightarrow Self-consistent condition

Flowchart of calculation



The DFT calculations basically consist of two loops. The inner loop is for SCF, and the outer loop is for geometry optimization.

The inner loop may have routines for construction of the KS matrix, eigenvalue problem, solution of Poisson eq., and charge mixing.

After getting a convergent structure, several physical quantities will be calculated.

Classification of the KS solvers

Treatment of core electrons

Basis functions

All electron (AE) method Pseudo-potential (PP) method Plane wave basis (PW) Mixed basis (MB) Local basis (LB)

	Accuracy	Efficiency
AE+MB: LAPW, LMTO	\bigcirc	×
AE+LB: Gaussian	0	0
PP+PW: Plane wave with PP	0	0
PP+LB: OpenMX, SIESTA	Δ	\bigcirc

LCPAO method

(Linear-Combination of Pseudo Atomic Orbital Method)

One-particle KS orbital

$$\psi_{\sigma\mu}^{(\mathbf{k})}(\mathbf{r}) = \frac{1}{\sqrt{N_{\rm c}}} \sum_{\rm n}^{N_{\rm c}} e^{i\mathbf{R}_{\rm n} \cdot \mathbf{k}} \sum_{i\alpha} c_{\sigma\mu,i\alpha}^{(\mathbf{k})} \phi_{i\alpha} (\mathbf{r} - \tau_i - \mathbf{R}_{\rm n})$$

is expressed by a linear combination of atomic like orbitals in the method.

$$\phi(\mathbf{r}) = Y_l^m(\hat{\mathbf{r}})R(r)$$

Features:

- It is easy to interpret physical and chemical meanings, since the KS orbitals are expressed by the atomic like basis functions.
- It gives rapid convergent results with respect to basis functions due to physical origin. (however, it is not a complete basis set, leading to difficulty in getting full convergence.)
- The memory and computational effort for calculation of matrix elements are O(N).
- It well matches the idea of linear scaling methods.

Total energy Pseudopotentials Basis functions

Implementation: Total energy (1)

The total energy is given by the sum of six terms, and a proper integration scheme for each term is applied to accurately evaluate the total energy.

 $E_{\rm tot} = E_{\rm kin} + E_{\rm ec} + E_{\rm ee} + E_{\rm xc} + E_{\rm cc} = E_{\rm kin} + E_{\rm na} + E_{\rm ec}^{(\rm NL)} + E_{\delta ee} + E_{\rm xc} + E_{\rm scc}.$

$$\begin{split} E_{\rm kin} &= \sum_{\sigma} \sum_{n}^{N} \sum_{i\alpha,j\beta} \rho_{\sigma,i\alpha j\beta}^{({\bf R}_n)} h_{i\alpha j\beta,\rm kin}^{({\bf R}_n)} \quad \text{Kinetic energy} \\ E_{\rm ec} &= \left[E_{\rm ec}^{({\bf L})} \right] + E_{\rm ec}^{({\bf NL})}, \quad \text{Coulomb energy with external potential} \\ &= \sum_{\sigma} \sum_{n}^{N} \sum_{i\alpha,j\beta} \rho_{\sigma,i\alpha j\beta}^{({\bf R}_n)} \phi_{i\alpha}({\bf r} - \tau_i) |\sum_{I} V_{\rm core,I}({\bf r} - \tau_I)| \phi_{j\beta}({\bf r} - \tau_j - {\bf R}_n) \rangle \\ &+ \sum_{\sigma} \sum_{n}^{N} \sum_{i\alpha,j\beta} \rho_{\sigma,i\alpha j\beta}^{({\bf R}_n)} \phi_{i\alpha}({\bf r} - \tau_i) |\sum_{I} V_{\rm NL,I}({\bf r} - \tau_I)| \phi_{j\beta}({\bf r} - \tau_j - {\bf R}_n) \rangle, \\ E_{\rm ee} &= \frac{1}{2} \int dr^3 n({\bf r}) V_{\rm H}({\bf r}), \quad \text{Hartree energy} \\ &= \frac{1}{2} \int dr^3 n({\bf r}) \{ V_{\rm H}^{(\alpha)}({\bf r}) + \delta V_{\rm H}({\bf r}) \} \\ E_{\rm xc} &= \int dr^3 \{ n_{\uparrow}({\bf r}) + n_{\downarrow}({\bf r}) + n_{\rm pcc}({\bf r}) \} \epsilon_{\rm xc} (n_{\uparrow} + \frac{1}{2} n_{\rm pcc}, n_{\downarrow} + \frac{1}{2} n_{\rm pcc}), \quad \underset{\rm energy}{\text{Exchange-correlation}} \\ &= \frac{1}{2} \sum_{I,J} \frac{Z_I Z_J}{|\tau_I - \tau_J|}. \quad \text{Core-core Coulomb energy} \\ \end{array}$$

Implementation: Total energy (2)

The reorganization of Coulomb energies gives three new energy terms.

$$E_{\rm ec}^{(L)} + E_{\rm ee} + E_{\rm cc} = E_{\rm na} + E_{\delta \rm ee} + E_{\rm scc},$$

The neutral atom energy $E_{\text{na}} = \int dr^3 n(\mathbf{r}) \sum_{I} V_{\text{na},I}(\mathbf{r} - \tau_I),$

Short range and separable to twocenter integrals

$$= \sum_{\sigma} \sum_{n}^{N} \sum_{i\alpha,j\beta} \rho_{\sigma,i\alpha j\beta}^{(\mathbf{R}_{n})} \sum_{I} \langle \phi_{i\alpha}(\mathbf{r}-\tau_{i}) | V_{\mathrm{na},I}(\mathbf{r}-\tau_{I}) | \phi_{j\beta}(\mathbf{r}-\tau_{j}-\mathbf{R}_{n}) \rangle,$$

Difference charge Hartree energy

$$E_{\delta ee} = \frac{1}{2} \int dr^3 \delta n(\mathbf{r}) \delta V_{\rm H}(\mathbf{r}),$$

Long range but minor contribution

Screened core-core repulsion energy

$$E_{\rm scc} = \frac{1}{2} \sum_{I,J} \left[\frac{Z_I Z_J}{|\tau_I - \tau_J|} - \int dr^3 n_I^{(\rm a)}(\mathbf{r}) V_{\rm H,J}^{(\rm a)}(\mathbf{r}) \right]$$

Short range and two-center integrals

Difference charge

$$\delta n(\mathbf{r}) = n(\mathbf{r}) - n^{(a)}(\mathbf{r}),$$

= $n(\mathbf{r}) - \sum_{i} n_{i}^{(a)}(\mathbf{r}),$

Neutral atom potential

$$V_{\mathrm{na},I}(\mathbf{r}-\tau_I) = V_{\mathrm{core},I}(\mathbf{r}-\tau_I) + V_{\mathrm{H},I}^{(\mathrm{a})}(\mathbf{r}-\tau_I).$$

Implementation: Total energy (3)

So, the total energy is given by

$$E_{\text{tot}} = E_{\text{kin}} + E_{\text{na}} + E_{\text{ec}}^{(\text{NL})} + E_{\delta \text{ee}} + E_{\text{xc}} + E_{\text{scc}}.$$

Each term is evaluated by using a different numerical grid with consideration on accuracy and efficiency.



Two center integrals

Fourier-transformation of basis functions

$$\begin{split} \tilde{\phi}_{i\alpha}(\mathbf{k}) &= \left(\frac{1}{\sqrt{2\pi}}\right)^3 \int dr^3 Y_{lm}(\hat{\mathbf{r}}) R_{pl}(r) \left\{ 4\pi \sum_{L=0}^{\infty} \sum_{M=-L}^{L} (-\mathbf{i})^L j_L(kr) Y_{LM}(\hat{\mathbf{k}}) Y_{LM}^*(\hat{\mathbf{r}}) \right\}, \\ &= \left(\frac{1}{\sqrt{2\pi}}\right)^3 4\pi \sum_{L=0}^{\infty} \sum_{M=-L}^{L} (-\mathbf{i})^L Y_{LM}(\hat{\mathbf{k}}) \int dr r^2 R_{pl}(r) j_L(kr) \int d\theta d\phi \sin(\theta) Y_{lm}(\hat{\mathbf{r}}) Y_{LM}^*(\hat{\mathbf{r}}), \\ &= \left[\left(\frac{1}{\sqrt{2\pi}}\right)^3 4\pi (-\mathbf{i})^l \int dr r^2 R_{pl}(r) j_l(kr) \right] Y_{lm}(\hat{\mathbf{k}}), \\ &= \tilde{R}_{pl}(k) Y_{lm}(\hat{\mathbf{k}}), \end{split}$$
 Integrals for angular parts are analytically

e.g., overlap integral

Integrals for angular parts are analytically performed. Thus, we only have to perform one-dimensional integrals along the radial direction.

$$\begin{split} \langle \phi_{i\alpha}(\mathbf{r}) | \phi_{j\beta}(\mathbf{r} - \tau) \rangle &= \int dr^3 \phi_{i\alpha}^*(\mathbf{r}) \phi_{j\beta}(\mathbf{r} - \tau), \\ &= \int dr^3 \left(\frac{1}{\sqrt{2\pi}}\right)^3 \int dk^3 \tilde{R}_{pl}^*(k) Y_{lm}^*(\hat{\mathbf{k}}) \mathrm{e}^{-\mathrm{i}\mathbf{k}\cdot\mathbf{r}} \left(\frac{1}{\sqrt{2\pi}}\right)^3 \int dk'^3 \tilde{R}_{p'l'}(k') Y_{l'm'}(\hat{\mathbf{k}}') \mathrm{e}^{\mathrm{i}\mathbf{k}'\cdot(\mathbf{r}-\tau)}, \\ &= \left(\frac{1}{2\pi}\right)^3 \int dk^3 \int dk'^3 \mathrm{e}^{-\mathrm{i}\mathbf{k}'\cdot\tau} \tilde{R}_{pl}^*(k) Y_{lm}^*(\hat{\mathbf{k}}) \tilde{R}_{p'l'}(k') Y_{l'm'}(\hat{\mathbf{k}}') \int dr^3 \mathrm{e}^{\mathrm{i}(\mathbf{k}'-\mathbf{k})\cdot\mathbf{r}}, \\ &= \int dk^3 \mathrm{e}^{-\mathrm{i}\mathbf{k}\cdot\tau} \tilde{R}_{pl}^*(k) Y_{lm}^*(\hat{\mathbf{k}}) \tilde{R}_{p'l'}(k) Y_{l'm'}(\hat{\mathbf{k}}), \end{split}$$

Cutoff energy for regular mesh

The two energy components $\mathbf{E}_{\delta ee} + \mathbf{E}_{xc}$ are calculated on real space regular mesh. The mesh fineness is determined by plane-wave cutoff energies.

scf.energycutoff



150.0 # default=150 (Ry)

The cutoff energy can be related to the mesh fineness by the following eqs.

$$E_{\rm cut}^{(1)} = \frac{1}{2} \mathbf{g} \mathbf{b}_1 \cdot \mathbf{g} \mathbf{b}_1, \quad E_{\rm cut}^{(2)} = \frac{1}{2} \mathbf{g} \mathbf{b}_2 \cdot \mathbf{g} \mathbf{b}_2, \quad E_{\rm cut}^{(3)} = \frac{1}{2} \mathbf{g} \mathbf{b}_3 \cdot \mathbf{g} \mathbf{b}_3,$$

$$\mathbf{g}\mathbf{a}_1 = \frac{\mathbf{a}_1}{N_1}, \quad \mathbf{g}\mathbf{a}_2 = \frac{\mathbf{a}_2}{N_2}, \quad \mathbf{g}\mathbf{a}_3 = \frac{\mathbf{a}_3}{N_3},$$

 $\mathbf{gb}_1 = 2\pi \frac{\mathbf{ga}_2 \times \mathbf{ga}_3}{\Delta V}, \quad \mathbf{gb}_2 = 2\pi \frac{\mathbf{ga}_3 \times \mathbf{ga}_1}{\Delta V}, \quad \mathbf{gb}_3 = 2\pi \frac{\mathbf{ga}_1 \times \mathbf{ga}_2}{\Delta V},$

 $\Delta V = \mathbf{g}\mathbf{a}_1 \cdot \left(\mathbf{g}\mathbf{a}_2 \times \mathbf{g}\mathbf{a}_3\right),$

Forces on atoms



Total energy Pseudopotentials Basis functions

Norm-conserving pseudopotential by MBK

I. Morrion, D.M. Bylander, and L. Kleinman, PRB 47, 6728 (1993).

If $Q_{ij} = 0$, the non-local terms can be transformed to a diagonal form.

$$V_{\rm NL} = \sum_{i,j} B_{ij} |\beta_i\rangle \langle\beta_j|,$$
$$= \sum_i \lambda_i |\alpha_i\rangle \langle\alpha_i|$$

The form is equivalent to that obtained from the Blochl expansion for TM norm-conserving pseudopotentials. Thus, common routines can be utilized for the MBK and TM pseudopotentials, resulting in easiness of the code development.

To satisfy $Q_{ij}=0$, pseudofunctions are now given by

$$\phi_i = \phi_{\mathrm{TM},i} + f_i \quad f_i = \sum_{i=0} c_i \left[r j_l \left(\frac{r}{r_c} u_{li} \right) \right]$$

The coefficients {c} are determined by agreement of derivatives and $Q_{ij}=0$. Once a set of {c} is determined, χ is given by

$$\chi_i = V_{\rm TM}^{(i)} \phi_{\rm TM,i} + \varepsilon_i f_i - V_{\rm loc} \phi_i - \frac{1}{2} \sum_i c_i \left(\frac{u_{li}}{r_c}\right)^2 \left[r j_l \left(\frac{r}{r_c} u_{li}\right)\right]$$

Optimization of pseudopotentials



Total energy Pseudopotentials Basis functions

Primitive basis functions

1. Solve an atomic Kohn-Sham eq. under a confinement potential:

$$V_{\text{core}}(r) = \begin{cases} -\frac{Z}{r} & \text{for } r \leq r_1 \\ \sum_{n=0}^{3} b_n r^n & \text{for } r_1 < r \leq r_c \\ h & \text{for } r_c < r, \end{cases}$$

- 2. Construct the norm-conserving pseudopotentials.
- 3. Solve ground and excited states for the the peudopotential for each L-channel.



In most cases, the accuracy and efficiency can be controlled by

Cutoff radius Number of orbitals

PRB 67, 155108 (2003) PRB 69, 195113 (2004)

Convergence with respect to basis functions

The two parameters can be regarded as variational parameters.



Benchmark of primitive basis functions

Dimer	Expt.	Calc.	Dimer	Expt.	Calc.
H_2 (H4.5-s2)	${}^{1}\Sigma_{g}^{+a}$	${}^{1}\Sigma_{g}^{+}(1s\sigma_{g}^{2})$	K ₂ (K10.0-s2p2)	${}^{1}\Sigma_{g}^{+ f}$	${}^{1}\Sigma_{g}^{+} (3p\pi_{g}^{4}3p\sigma_{u}^{2}4s\sigma_{g}^{2})$
He_2 (He7.0-s2)	${}^{1}\Sigma_{g}^{+}b$	${}^{1}\Sigma_{g}^{+}(1s\sigma_{g}^{2}1s\sigma_{u}^{2})$	CaO (Ca7.0- $s2p2d2$)	${}^{1}\Sigma^{\downarrow k}$	${}^{1}\Sigma^{+} (s\sigma^{2}s\sigma^{2}p\pi^{4})$
Li ₂ (Li8.0-s2)	${}^{1}\Sigma_{g}^{+}c$	${}^{1}\Sigma_{g}^{+}(2s\sigma_{g}^{2})$	ScO (Sc7.0- $s2p2d2$)	$2\Sigma + l$	$^{2}\Sigma^{+}$ $(d\pi^{4}s\sigma^{2}s\sigma^{1})$
BeO (Be6.0-s2p2)	${}^{1}\Sigma^{+} d$	${}^{1}\Sigma^{+} (s\sigma^{2}s\sigma^{2}p\pi^{4})$	Ti ₂ (Ti7.0-s2p2d2)	$^{3}\Delta_{g}m$	$^{3}\Delta_{g} (4s\sigma_{g}^{2}3d\sigma_{g}^{1}3d\pi_{u}^{4}3d\delta_{g}^{1})$
B ₂ (B5.5-s2p2)	${}^{3}\Sigma_{g}^{-e}$	${}^{3}\Sigma_{g}^{-}(2s\sigma_{g}^{2}2s\sigma_{u}^{2}2\pi_{u}^{2})$	V ₂ (V7.5-s2p2d2)	${}^{3}\Sigma_{q}^{-n}$	${}^{1}\Sigma_{q}^{+}$ $(4s\sigma_{q}^{2}3d\sigma_{q}^{2}3d\pi_{u}^{4}3d\delta_{q}^{2})$
C_2 (C5.0-s2p2)	${}^{1}\Sigma_{g}^{H}f$	${}^{1}\Sigma_{g}^{+}$ $(2s\sigma_{g}^{2}2s\sigma_{u}^{2}2p\pi_{u}^{4})$	V_2 (V7.5-s4p4d4f2)	${}^{3}\Sigma_{q}^{-n}$	${}^{3}\Sigma_{q}^{-}$ $(4s\sigma_{q}^{2}3d\sigma_{q}^{2}3d\pi_{u}^{4}3d\delta_{q}^{2})$
N ₂ (N5.0-s2p2)	${}^{1}\Sigma_{q}^{+}f$	${}^{1}\Sigma_{\sigma}^{+} (2s\sigma_{u}^{2}2p\pi_{u}^{4}2p\sigma_{a}^{2})$	Cr_2 (Cr7.0- <i>s</i> 2 <i>p</i> 2 <i>d</i> 2)	${}^{1}\Sigma_{a}^{+} {}^{o}$	${}^{1}\Sigma_{a}^{+}$ $(4s\sigma_{a}^{2}3d\sigma_{a}^{2}3d\pi_{u}^{4}3d\delta_{a}^{4})$
O_2 (O5.0-s2p2)	${}^{3}\Sigma_{q}^{n}f$	${}^{3}\Sigma_{a}^{-}(2p\sigma_{a}^{2}2p\pi_{u}^{4}2p\pi_{a}^{2})$	MnO (Mn7.0-s2p2d2)	$^{6}\Sigma^{+}p$	${}^{6}\Sigma^{+} (d\sigma^{1}d\pi^{4}d\delta^{2}d\pi^{*2})$
F ₂ (F5.0-s2p2)	${}^{1}\Sigma_{g}^{4}f$	${}^{1}\Sigma_{q}^{+}(2p\sigma_{q}^{2}2p\pi_{u}^{4}2p\pi_{q}^{4})$	Fe_2 (Fe7.0- $s2p2d2$)	$^{7}\Delta_{u}^{q}$	$^{7}\Delta_{u} \left(4s\sigma_{g}^{2}3d\sigma_{g}^{2}3d\sigma_{u}^{1}3d\pi_{u}^{4}3d\pi_{g}^{2}3d\delta_{g}^{3}3d\delta_{u}^{2}\right)$
Ne ₂ (Ne7.0-s2p2)	${}^{1}\Sigma_{g}^{+}g$	${}^{1}\Sigma_{\sigma}^{+}(2p\pi_{u}^{4}2p\pi_{o}^{4}2p\sigma_{u}^{2})$	Co_2 (Co7.0-s2p2d2)		${}^{5}\Delta_{g} \left(4s\sigma_{a}^{2}3d\sigma_{a}^{2}3d\sigma_{u}^{1}3d\pi_{u}^{4}3d\pi_{a}^{2}3d\delta_{a}^{4}3d\delta_{u}^{3}\right)$
Na ₂ (Na9.0-s2p2)	${}^{1}\Sigma_{g}^{4}f$	${}^{1}\Sigma_{q}^{+}(2p\pi_{q}^{4}2p\sigma_{u}^{2}3s\sigma_{q}^{2})$	Ni ₂ (Ni7.0-s2p2d2)	Ω^r	${}^{3}\Sigma_{q}^{-}$ $(4s\sigma_{q}^{2}3d\sigma_{q}^{2}3d\sigma_{u}^{2}3d\pi_{u}^{4}3d\pi_{q}^{2}3d\delta_{q}^{4}3d\delta_{u}^{4})$
MgO (Mg7.0-s2p2)	${}^{1}\Sigma^{+}h$	${}^{1}\Sigma^{+}(s\sigma^{2}s\sigma^{2}p\pi^{4})$	Cu_2 (Cu7.0-s2p2d2)	${}^{1}\Sigma_{q}^{+s}$	${}^{1}\Sigma_{q}^{+} (4s\sigma_{q}^{2}3d\sigma_{q}^{2}3d\sigma_{u}^{2}3d\pi_{u}^{4}3d\pi_{q}^{4}3d\delta_{q}^{4}3d\delta_{u}^{4})$
Al ₂ (Al6.5-s2p2)	${}^{3}\Pi_{u} i$	${}^{3}\Sigma_{\sigma}^{-}(3s\sigma_{a}^{2}3s\sigma_{u}^{2}3p\pi_{u}^{2})$	ZnH (Zn7.0-s2p2d2)	${}^{2}\Sigma_{a}^{+}t$	$^{2}\Sigma_{a}^{+}(s\sigma^{2}s\sigma^{*1}d\sigma^{2}d\pi^{4}d\delta^{4})$
Al ₂ (Al6.5-s4p4d2)	${}^{3}\Pi_{u} i$	${}^{3}\Sigma_{q}^{-}(3s\sigma_{q}^{2}3s\sigma_{u}^{2}3p\pi_{u}^{2})$	GaH (Ga7.0-s2p2)	${}^{1}\Sigma^{+}u$	$^{1}\Sigma^{+}(s\sigma^{2}s\sigma^{*2})$
Si ₂ (Si6.5-s2p2)	${}^{3}\Sigma_{g}^{-}f$	${}^{3}\Pi_{u} (3s\sigma_{u}^{2}3s\sigma_{q}^{1}3p\pi_{u}^{3})$	GeO (Ge7.0-s2p2)	${}^{1}\Sigma^{+}f$	$^{1}\Sigma^{+}$ (ss $\sigma^{2}sp\sigma^{2}pp\pi^{4}pp\sigma^{2}$)
Si ₂ (Si6.5-s2p2d1)	${}^{3}\Sigma_{\sigma}^{-}f$	${}^{3}\Sigma_{\sigma}^{-}(3s\sigma_{u}^{2}3p\pi_{u}^{2}3s\sigma_{a}^{2})$	As ₂ (As7.0-s2p2d1)	${}^{1}\Sigma_{a}^{+}f$	${}^{1}\Sigma_{a}^{+}$ $(4s\sigma_{a}^{2}4s\sigma_{u}^{2}4p\sigma_{a}^{2}4p\pi_{u}^{4})$
P_2 (P6.0-s2p2d1)	${}^{1}\Sigma_{g}^{4}f$	${}^{1}\Sigma_{q}^{+}(3s\sigma_{u}^{2}3p\sigma_{g}^{2}3p\pi_{u}^{4})$	Se_2 (Se7.0-s2p2d1)	${}^{3}\Sigma_{q}^{s}f$	${}^{3}\Sigma_{a}^{-}(4s\sigma_{a}^{2}4s\sigma_{u}^{2}4p\sigma_{a}^{2}4p\pi_{u}^{4}4p\pi_{a}^{2})$
S ₂ (S6.0-s2p2)	${}^{3}\Sigma_{g}^{n}f$	${}^{3}\Sigma_{g}^{-}(3p\sigma_{g}^{2}3p\pi_{u}^{4}3p\pi_{g}^{2})$	Br ₂ (Br7.0-s2p2d1)	${}^{1}\Sigma_{q}^{+}f$	${}^{1}\Sigma_{q}^{+} (4s\sigma_{q}^{2}4s\sigma_{u}^{2}4p\sigma_{q}^{2}4p\pi_{u}^{4}4p\pi_{q}^{4})$
Cl_2 (Cl6.0-s2p2d2)	${}^{1}\Sigma_{g}^{H}f$	${}^{1}\Sigma_{g}^{+} (3p\sigma_{g}^{2}3p\pi_{u}^{4}3p\pi_{g}^{4})$	Kr ₂ (Kr7.0-s2p2)	${}^{1}\Sigma_{g}^{+} v$	${}^{1}\Sigma_{q}^{+}$ $(4s\sigma_{q}^{2}4s\sigma_{u}^{2}4p\sigma_{q}^{2}4p\sigma_{u}^{2}4p\pi_{u}^{4}4p\pi_{q}^{4})$
Ar_2 (Ar7.0-s2p2)	${}^{1}\Sigma_{a}^{+}j$	${}^{1}\Sigma_{\pi}^{+}(3p\pi_{*}^{4}3p\pi_{*}^{4}3p\sigma_{*}^{2})$		2	u u - u

Ground state calculations of dimer using primitive basis functions

All the successes and failures by the LDA are reproduced by the modest size of basis functions (DNP in most cases)

Variational optimization of basis functions

One-particle wave functions

 $\psi_{\mu}(\mathbf{r}) = \sum_{i\alpha} c_{\mu,i\alpha} \phi_{i\alpha}(\mathbf{r} - \mathbf{r}_i)$

Contracted orbitals

$$\phi_{i\alpha}(\mathbf{r}) = \sum_{q} a_{i\alpha q} \chi_{i\eta}(\mathbf{r})$$

The variation of E with respect to c with fixed a gives

$$\partial E_{\text{tot}} / \partial c_{\mu,i\alpha} = 0 \longrightarrow \sum_{j\beta} \langle \phi_{i\alpha} | \hat{H} | \phi_{j\beta} \rangle c_{\mu,j\beta} = \varepsilon_{\mu} \sum_{j\beta} \langle \phi_{i\alpha} | \phi_{j\beta} \rangle c_{\mu,j\beta}$$

Regarding c as dependent variables on a and assuming KS eq. is solved self-consistently with respect to c, we have

$$\frac{\partial E_{\text{tot}}}{\partial a_{i\alpha q}} = \frac{\delta E_{\text{tot}}}{\delta \rho(\mathbf{r})} \frac{\delta \rho(\mathbf{r})}{\delta a_{i\alpha q}}$$
$$= 2 \sum_{j\beta} \left(\Theta_{i\alpha,j\beta} \langle \chi_{i\eta} | \hat{H} | \phi_{j\beta} \rangle - E_{i\alpha,j\beta} \langle \chi_{i\eta} | \phi_{j\beta} \rangle \right)$$

Ozaki, PRB 67, 155108 (2003)

Comparison between primitive and optimized basis functions



Ozaki, PRB 67, 155108 (2003).

Optimization of basis functions

1. Choose typical chemical environments



2. Optimize variationally the radial functions



3. Rotate a set of optimized orbitals within the subspace, and discard the redundant functions





Database of optimized VPS and PAO

Database (2013) of optimized VPS and PAO

The database (2013) of fully relativistic pseudopotentials (VPS) and pseudo-atomic orbitals (PAO), generated by ADPACK, which could be an input data of program package, OpenMX. The data of elements with the underline are currently available. When you use these data, VPS and PAO, in the program package, OpenMX, then copy them to the directory, openmx*.*/DFT_DATA13/VPS/ and openmx*.*/DFT_DATA13/PAO/, respectively. The delta factor of OpenMX with the database (2013) is found at <u>here</u>.

E		Public release of optimized and well tested VPS and PAO so															
H		that	hat users can easily start their calculations. <u>He</u>														
<u>Li</u>	<u>Be</u>											<u>B</u>	<u>C</u>	<u>N</u>	<u>0</u>	E	<u>Ne</u>
<u>Na</u>	<u>Mg</u>											<u>AI</u>	<u>Si</u>	Р	<u>s</u>	<u>CI</u>	<u>Ar</u>
<u>K</u>	<u>Ca</u>	<u>Sc</u>	<u>Ti</u>	V	<u>Cr</u>	<u>Mn</u>	<u>Fe</u>	<u>Co</u>	<u>Ni</u>	<u>Cu</u>	<u>Zn</u>	<u>Ga</u>	<u>Ge</u>	<u>As</u>	<u>Se</u>	<u>Br</u>	<u>Kr</u>
<u>Rb</u>	<u>Sr</u>	Y	<u>Zr</u>	<u>Nb</u>	<u>Mo</u>	<u>Tc</u>	<u>Ru</u>	<u>Rh</u>	<u>Pd</u>	<u>Ag</u>	<u>Cd</u>	<u>In</u>	<u>Sn</u>	<u>Sb</u>	<u>Te</u>	I	<u>Xe</u>
<u>Cs</u>	<u>Ba</u>	L	<u>Hf</u>	<u>Ta</u>	W	<u>Re</u>	<u>Os</u>	<u>Ir</u>	<u>Pt</u>	<u>Au</u>	<u>Hg</u>	ΤI	<u>Pb</u>	<u>Bi</u>	<u>Po</u>	At	<u>Rn</u>
Fr	Ra	А															
	L	La	Ce	Pr	<u>Nd</u>	Pm	<u>Sm</u>	Eu	Gd	Tb	<u>Dy</u>	<u>Ho</u>	Er	Tm	Yb	<u>Lu</u>	
	А	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	

Reproducibility in DFT calcs

RESEARCH ARTICLE

Science 351, aad3000 (2016)

DFT METHODS

Reproducibility in density functional theory calculations of solids

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15 codes
69 researchers
71 elemental bulks
GGA-PBE
Scalar relativistic

PBE lattice constant of Si



Δ-gauge A way of comparing accuracy of codes



Volume (Å³ /atom)

Evaluation of GGA-PBE By Δ -gauge

In comparison of GGA-PBE with Expts. of 58 elements, the mean Δ -gauge is 23.5meV/atom.

н												He					
Li	Be	$\Delta(\exp) = 23.5 \text{ meV/atom}$													Ne		
0.4		17.6 17.1														17.1	
Na	Mg											Al	Si	р	s	CI	Ar
0.2	1.5						3.9	13.7	19.0	44.0	46.1	38.6					
К	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
1.4	3.0	1.7	3.0	13.1	1.6	47.7	10.6	3.3	6.2	13.5	7.6		28.2	17.1	11.1	22.3	54.6
Rb	Sr	Y	Zr	Nb	Mo	Тс	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
1.3	2.4	0.7	8.2	14.8	26.5		30.4	44.6	49.1	39.3	34.0	23.6	40.7	25.1	9.9	68.1	59.8
Cs	Ba	Lu	Hf	Та	W	Re	Os	lr	Pt	Au	Hg	TI	Pb	Bi	Ро	At	Rn
4.0	2.5		2.8	16.8	35.6	29.9	65.2	50.7	54.9	64.0		42.4	32.9	32.0			

Comparison of codes by Δ -gauge

		AE							PAW						USPP					P					
	Elk	exciting	FHI-aims/tier2	FLEUR	FPLO/T+F+s	RSPt	WIEN2k/acc	GBRV12/ABINIT	GPAW09/ABINIT	GPAW09/GPAW	JTH02/ABINIT	PSlib100/QE	VASPGW2015/VASP	GBRV14/CASTEP	GBRV14/QE	OTFG9/CASTEP	SSSP/QE	Vdb2/DACAPO	FHI98pp/ABINIT	HGH/ABINIT	HGH-NLCC/BigDFT	MBK2013/OpenMX	PSP(PD0.1)/ABINIT	ONCVPSP (SG15) 1/QE	PSP (SG15) 2/CASTEP
average < Δ >	0.6	0.5	0.5	0.6	0.9	0.8	0.5	0.9	1.4	1.6	0.6	0.9	0.6	1.1	1.1	0.7	0.5	6.3	13.3	2.2	1.1	2.0	0.7	1.4	1.4
Elk	2	0.3	0.3	0.6	1.0	0.9	0.3	0.9	1.3	1.5	0.6	0.9	0.4	1.1	1.0	0.4	0.4	6.3	13.5	2.2	1.1	2.1	0.7	1.4	1.4
exciting	0.3		0.1	0.5	0.9	0.8	0.2	0.8	1.3	15	0.6	0.8	0.4	1.1	1.0	0.5	0.3	6.3	13.4	2.2	Ľ	2.1	0.7	1.3	1.4
FHI-aims/tier2	0.3	0.1		0.5	0.9	0.8	0.2	0.8	1.3	1.5	0.6	0.8	0.4	1.0	0.9	0.5	0.3	6.3	13.4	2.2	11	2.1	0.7	1.3	1.4
	0.6	0.5	0.5		0.8	0.6	0.4	0.9	1.3	1.5	0.6	0.8	0.6	1.0	1.0	0.7	0.5	6.3	13.2	2.0	1.0	1.9	0.6	1.3	1.3
FPLO/T+F+s	1.0	0.9	0.9	0.8		0.9	0.9	1.3	1.7	1.8	0.9	1.3	1.0	1.4	1.4	1.0	0.9	6.4	13.0	2.3	1.2	1.8	1.0	1.6	1.6
RSPt	0.9	0.8	0.8	0.6	0.9		0.8	1.1	1.5	1.7	0.7	1.1	0.8	1.3	1.3	1.0	0.8	6.5	13.2	2.2	1.1	1.8	0.8	1.5	1.5
WIEN2k/acc	0.3	0.2	0.2	0.4	0.9	0.8		0.8	1.3	15	0.5	0.8	0.3	1.0	1.0	0.5	0.3	6.2	13.4	2.1	1.0	2.0	0.6	1.3	1.4

The mean Δ -gauge of OpenMX is 2.0meV/atom.

Practical guide to OpenMX calculations

- Choice of cutoff energy
- Calculations of energy curves
- SCF calculations
- How to choose basis functions
- Work functions and floating states
- Overcompleteness
- Restarting
- Outputting in a binary mode

Choice of cutoff energy

scf.energycutoff 200 # default=150 Ryd

The FFT grid is used to discretize real space and calculate $E_{\delta ee}$, E_{xc} , and can be specified by scf.energycutoff.

In most cases, 200 Ryd is enough to get convergence.

However, large cutoff energy (300-400 Ryd) has be used cases such as use of pseudopotentials with deep semi-core states.

Memory requiment O(E^{3/2})



Choice of cutoff energy

Geometry optimization of H_2O

Dependency of optimized structure of H2O on scf.energycutoff. It turns out that 180Ryd. is enough to reach the convergence.



Table 1: Convergence of structural parameters, dipole moment of a water molecule with respect to the cutoff energy. The input file is 'H2O.dat' in the directory 'work'.

Ecut(Ryd)	r(H-O) (Å)	\angle (H-O-H) (deg)	Dipole moment (Debye)
60	0.970	103.4	1.838
90	0.971	103.7	1.829
120	0.971	103.7	1.832
150	0.971	103.6	1.829
180	0.971	103.6	1.833
Exp.	0.957	104.5	1.85

Volume vs. Energy curves

The following keywords are available to calculate energy curves.

32

MD.Type EvsLC MD.EvsLC.Step 0.4 **MD**.maxIter

default=0.4% # default=1

When the energy curve for bulk system is calculated as a function of the lattice parameter, a sudden change of the number of real space grids is a serious problem which produces an erratic discontinuity on the energy curve. To avoid this, the number of grids should be fixed by explicitly specifying the following keyword:

scf.Ngrid 32 32 32

The numbers correspond to the number of grid along a-, b-, and c-axes, respectively. scf.Ngrid is used if both the keywords scf.energycutoff and scf.Ngrid are specified.



Self-consistency: Simple charge mixing

The KS effective is constructed from ρ . However, ρ is evaluated from eigenfaunctions of KS eq.

$$\hat{H}_{\text{KS}}\phi_i = \varepsilon_i\phi_i \qquad \hat{H}_{\text{KS}} = -\frac{1}{2}\nabla^2 + v_{\text{eff}}$$
$$v_{\text{eff}} = v_{\text{ext}}(\mathbf{r}) + v_{\text{Hartree}}(\mathbf{r}) + \frac{\delta E_{\text{xc}}}{\delta\rho(\mathbf{r})}$$
$$\rho(\mathbf{r}) = \sum_i \phi_i^*(\mathbf{r})\phi_i(\mathbf{r})$$

Simple charge mixing method

The next input density is constructed by a simple mixing of input and output densities.

$$\rho_{n+1}^{(\mathrm{in})} = \alpha \rho_n^{(\mathrm{in})} + (1 - \alpha) \rho_n^{(\mathrm{out})},$$

It works well for large gap systems and small sized systems.

Self-consistency: RMM-DIIS

Idea:

Minimize the norm of a linear combination of previous residual vectors.

Minimization of F leads to

Long wave length components corresponding to small |q| are taken into account.

$$\frac{\partial F}{\partial \alpha_k} = 0 \qquad \begin{pmatrix} \langle R_{n-(p-1)} | R_{n-(p-1)} \rangle & \cdots & \cdots & 1 \\ \cdots & \cdots & \cdots & 1 \\ \vdots & \vdots & \ddots & \ddots & \ddots \\ 1 & 1 & \cdots & 0 \end{pmatrix} \begin{pmatrix} \alpha_{n-(p-1)} \\ \alpha_{n-(p-1)+1} \\ \vdots \\ \frac{1}{2}\lambda \end{pmatrix} = \begin{pmatrix} 0 \\ 0 \\ \vdots \\ 1 \end{pmatrix}$$

Optimum input density might be given by

$$\rho_{n+1}^{(\text{in})} = \sum_{m=n-(p-1)}^{n} \alpha_m \rho_m^{(\text{in})} + \beta \sum_{m=n-(p-1)}^{n} \alpha_m R_m$$

G.Kresse and J. Furthmeuller, PRB 54, 11169 (1996).

Mixing methods



Available mixing methods:

Simple mixing (Simple)

Residual minimization method in the direct inversion iterative subspace (RMM-DIIS)

Guaranteed reduction Pulay method (GR-Pulay)

Kerker mixing (Kerker)

RMM-DIIS with Kerker metric (RMM-DIISK)

RMM-DIIS for Hamiltonian (RMM-DIISH)

Recommendation: RMM-DIISK or RMM-DIISH

See also the page 56 in the manual.

Norm of residual density matrix or charge density

Specification of PAO and VPS

PAO and VPS are specified by the following keyword:

<Definition.of.Atomic.Species
 0 07.0-s2p2d1 0_PBE13
 H H7.0-s2p1 H_PBE13
Definition.of.Atomic.Species>

- O7.0 means O7.0.pao.
- -s2p2d1 means 2, 2, and 1 radial functions are allocated to s-, p-, and d-orbitals.
- In this case, for oxygen atom, 2×1+2×3+1×5=13 basis functions are allocated.
- O_PBE13 meand O_PBE13.vps.

The path for O7.0.pao and O_PBE13.vps is specified by

DATA.PATH /home/soft/openmx3.8/DFT_DATA13 Default value is '../DFT_DATA13'.

How to choose basis functions: H₂O case

By clicking H7.0.pao and O7.0.pao in the database(2013), you may find the following

http://www.jaist.ac.jp/~t-ozaki/vps_pao2013/H/index.html

http://www.jaist.ac.jp/~t-ozaki/vps_pao2013/O/index.html

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1 5 4.156 mu 1 8 9.172 mu 1 9 11.271 mu 1 10 13.583 mu 1 11 16.109 mu 1 12 18.847 mu 1 13 21.799 mu 1 14 24.965 mu 2 0 0.276 mu 2 1 0.769 mu 2 1 0.769 mu 2 3 2.398	************************************	************************************	Eigen values (Hartree) of pse max= 3 Mul=15 mu 0 0 -0.23595211036 mu 0 1 0.14109389991 mu 0 2 0.61751730037 mu 0 3 1.31890671596 mu 0 4 2.24052765606 mu 0 5 3.37954791544 mu 0 6 4.73486369825 mu 0 7 6.30608874477 mu 0 8 8.09282718517 mu 0 9 10.09464035733 mu 0 10 12.31085267019 mu 0 11 14.74057314485 mu 0 12 17.38277845742 mu 0 12 17.3827784574457 mu 0 13 20.23645090755 mu 0 14 23.3007392659 mu 0 14 23.3007392659 mu 1 0 0.10914684880 mu 1 2 1.06988680483 mu 1 3 1.88261331124 mu 1 4 2.91175885863 mu 1 5 4.15601184783 mu 1 5 4.15601184783 mu 1 5 4.15601184783 mu 1 6 5.61454131066 mu 1 7 7.28681796296 mu 1 8 9.17254361476 mu 1 9 11.27156766390 mu 1 10 13.58381334565 mu 1 12 18.84767560575 mu 1 13 21.79902110024 mu 1 14 24.96300480786 mu 1 12 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	Eigen	values	(Hartree) of pseudo atomic orbitals
*	*****	******	**************
			070 nao
E	igenva	lues	
L	max= 3	Mul=15	1
	Imu	ÖÖ	-0.87913976280231
	lmu	ŌĨ	0.06809061901229
	l mu	ňż	0.52709275865941
	l mu	ňā	1.24995140722317
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	l mu	ñ s	3.41945468859267
	l mu	ăă	4.84509607059749
	l mu	Ň Ž	6.48825090142865
	l mu	ň ś	8.34207134316001
	l mu	ñй	10.39973244132192
	l mu	ñ 1ñ	12.65513764955926
	l mu	ñ iĭ	15.10428688136984
	l mu	ň iż	17.74677530362947
	l mu	ñ 13	20.58633940582683
	l mu	ñ 14	23.62957122674031
	l mu	ĩċ	-0.33075182895384
	lmu	i ī	0.16376499567753
	Imu	12	0.64129274864838 🔨 💶
	Imu	13	1.35995471521821
	lmu	1 4	2.31377697411480
	lmu	1 5	3.50052618379026
	lmu	16	4.91841590421346
	l mu	1 7	6.56499793348396
	l mu	1 8	8.43634105410091
	l mu	19	10.52733540479795
	l mu	1 10	12.83276989780839
	l mu	1 11	15.34880171573570
	l mu	1 12	18.07414632417004
	l mu	1 13	21.01013110934429 Q
	l mu	1 14	24.15933607166566 📈 💙
	l mu	20	0.26162948257116 🦰
	l mu	21	0.70705247436937
	l mu	22	1.34714706672243
	l mu	23	2.19799459356269
	l mu	2 A	3.26511989658328

Choosing states with lower eigenvalues leads to H7.0-s2p1 and O7.0-s2p2d1.

How to choose basis functions: Si case(1)

Si7.0.pao Eigenvalues Lmax= 3 Mul=15

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Si7.0-s2p2d1 is enough to discuss structural properties.

By comparing Si7.0s3p2d2f1 with Si8.0s3p2d2f1, it turns out that convergence is achieved at the cutoff of 7.0(a.u.).

How to choose basis functions: Si case(2)

With respect to band structure, one can confirm that Si7.0-s2p2d1 provides a nearly convent result.



While the convergent result is achieved by use of Si7.0-s3p2d2f1(Si7.0-s3p3d2f1), Si7.0-s2p2d1 is a balanced basis functions compromising accuracy and efficiency to perform a vast range of materials exploration.

Floating states in 3C-SiC



Inclusion of polarization orbitals is important to reproduce band structures.
The band structure up to 5 eV is reproduced by s2p2d1.

Basis set superposition Error (BSSE)



	Equilibrium O-O distance (Ang.)	Dipole moment (Debye)	Bind (I	ding energy kcal/mol)	Binding ene (couterpoi corrected (kcal/mol	ergy se I))
O7.0-s2p2d1, H7.0-s2p1	2.899	2.54		5.57	5.21	
O7.0-s3p3d2, H7.0-s3p2	2.897	2.45		5.48	5.48	
Other calc.	2.893 ^a			5.15 ^a		
Expt.	2.98 ^b	2.60 ^b		5.44 ^b		

http://www.jaist.ac.jp/~t-ozaki/vps_pao2013/O/index.html

A series of benchmark calculations implies that BSSE is ~ 0.5 kcal/mol for molecular systems.

Work functions fcc Al (111) surface

By allocating empty atoms in vacuum near the surface, one can calculate work functions accurately.



See: http://www.openmx-square.org/openmx_man3.8/node32.html http://www.openmx-square.org/forum/patio.cgi?mode=view&no=2305

Overcompleteness of basis functions



A numerical instability, called "overcompleteness", tends to appear if a lot of basis functions are used for dense structures such as fcc, hcp, and bcc.

Cause of overcompleteness



Restarting of calculations

• After finishing your first calculation or achieving the self consistency, you may want to continue the calculation or to calculate density of states, band dispersion, molecular orbitals, and etc. using the self consistent charge in order to save the computational time. To do this, a keyword 'scf.restart' is available.

scf.restart on # on|off,default=off

• If the first trial for geometry optimization does not reach a convergent result or molecular a dynamics simulation is terminated due to a wall time, one can restart the geometry optimization using an input file 'System.Name.dat#' which is generated at every step for the restart calculation with the final structure.

See also

http://www.openmx-square.org/openmx_man3.8/node46.html http://www.openmx-square.org/openmx_man3.8/node54.html

Output of large-sized files in binary mode

Large-scale calculations produce large-sized files in text mode such as cube files. The IO access to output such files can be very time consuming in machines of which IO access is not fast. In such a case, it is better to output those large-sized files in binary mode. The procedure is supported by the following keyword:

OutData.bin.flag on # default=off, on|off

Then, all large-sized files will be output in binary mode. The default is 'off'. The output binary files are converted using a small code 'bin2txt.c' stored in the directory 'source' which can be compiled as

gcc bin2txt.c -lm -o bin2txt

As a post processing, you will be able to convert as

./bin2txt *.bin

The functionality will be useful for machines of which IO access is not fast.

See also http://www.openmx-square.org/openmx_man3.8/node172.html

Large-scale calculations

The following is a result of 'runtestL2' performed using 264 MPI processes and 2 OpenMP threads on CRAY-XC30.

\$ mpirun –np 264 openmx -runtestL2 -nt 2

1	large2_example/C1000.dat	Elapsed time(s)= 1731.83	diff Utot= 0.00000002838	diff Force= 0.00000007504
2	large2_example/Fe1000.dat	Elapsed time(s)=21731.24	diff Utot= 0.000000010856	diff Force= 0.00000000580
3	large2_example/GRA1024.dat	Elapsed time(s)= 2245.67	diff Utot= 0.00000002291	diff Force= 0.000000015333
4	large2_example/Ih-Ice1200.dat	Elapsed time(s)= 952.84	diff Utot= 0.00000000031	diff Force= 0.00000000213
5	large2_example/Pt500.dat	Elapsed time(s)= 6831.16	diff Utot= 0.00000002285	diff Force= 0.00000004010
6	large2_example/R-TiO2-1050.dat	Elapsed time(s)= 2259.97	diff Utot= 0.00000000106	diff Force= 0.00000001249
7	large2_example/Si1000.dat	Elapsed time(s)= 1655.25	diff Utot= 0.00000001615	diff Force= 0.00000005764
	Total elapsed time (s) 37407.95			

The elapsed time implies that geometry optimization for systems consisting of 1000 atoms is possible if several hundreds processor cores are available.

See also http://www.openmx-square.org/openmx_man3.8/node87.html http://www.openmx-square.org/openmx_man3.8/node88.html

On the manual

Please download the manual at

http://www.openmx-square.org/openmx_man3.8/openmx3.8.pdf

The manual is self-contained, the most of calculations explained in the manual are traceable by using the input file stored in the directory 'work'.

Please try to perform those calculations one by one depending on your interests.

Recommended trials

1. Geometry optimization

Perform a geometry optimization using 'Methane2.dat'. See the page 65 in the manual.

2. Density of states

Calculate DOS using 'Cdia.dat' See the page 79 in the manual.

All the input files can be found in the directory 'work'.

3. Wannier functions

Calculate Wannier functions for Si bulk using 'work/wf_example/Si.dat', and perform the band interpolation. See the page 159 in the manual.

4. Reaction barrier by the nudged elastic band (NEB) method

Calculate a reaction barrier using 'C2H4_NEB.dat'. See the page 182 in the manual.

5. Transmission of a carbon chain

Calculate an electric transmission of a carbon chain using 'Lead-Chain.dat', 'NEGF-Chain.dat'. See the page 136 in the manual.

6. Spin-orbit coupling

Calculate a band structure by taking account of SOC using 'GaAs.dat'. See the page 117 in the manual.

Outlook

- A localized basis method, implemented in OpenMX, was discussed with the following focuses:
 - Total energy
 - Pseudopontials
 - Basis functions
 - The careful evaluation of the total energy and optimization of PPs and PAOs guarantee accurate and fast DFT calculations in a balanced way.
 - The practical guideline may be useful for getting started with OpenMX calculations.